## THE DEVELOPMENT OF GLYCOSAMINOGLYCAN-BASED MATERIALS TO PROMOTE CHONDROGENIC DIFFERENTIATION OF MESENCHYMAL STEM CELLS

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by

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# THE DEVELOPMENT OF GLYCOSAMINOGLYCAN-BASED MATERIALS TO PROMOTE CHONDROGENIC DIFFERENTIATION OF MESENCHYMAL STEM CELLS

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Dedicated to my loving family: To Mom and Dad for the support, devotion, and confidence, and to Court and T.J. for the laughter, guidance, and companionship.



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## LIST OF ABBREVIATIONS

| AcCl          | acryloyl chloride                                       |
|---------------|---|
| ACI           | autologous chondrocyte implantation                     |
| ACL           | anterior cruciate ligament                              |
| ANOVA         | analysis of variance                                    |
| APMAm         | N-(3-aminopropyl)methacrylamide                         |
| APS           | ammonium persulfate                                     |
| bFGF          | basic fibroblast growth factor                          |
| BMP           | bone morphogenetic protein                              |
| BMSC          | bone marrow stromal cell                                |
| BSA           | bovine serum albumin                                    |
| C4st1         | chondroitin-4-sulfotransferase 1                        |
| C6st1         | chondroitin-6-sulfotransferase 1                        |
| Ch-MA         | chondroitin methacrylate                                |
| ChSy1         | chondroitin sulfate synthase 1                          |
| COMP          | cartilage oligomeric matrix protein                     |
| CS            | chondroitin sulfate                                     |
| CSGalNAcT1    | chondroitin sulfate N-acetylgalactosaminyltransferase 1 |
| CSMA or CS-MA | chondroitin sulfate methacrylate                        |
| CSMAm         | chondroitin sulfate methacrylamide                      |
| D2959         | Irgacure 2959   |
| $D_2O$        | deuterated water  |
| DAB           | diaminobenzidine  |
| DAPI          | 4',6-diamidino-2-phenylindole                           |



| ddH <sub>2</sub> O | distilled, deionized water                                 |
|--------------------|--|
| dH <sub>2</sub> O  | distilled water  |
| DLS                | dynamic light scattering                                   |
| DMMB               | dimethylmethylene blue                                     |
| DMSO               | dimethyl sulfoxide   |
| EB                 | embryoid body  |
| ECM                | extracellular matrix                                       |
| EDC                | N-(3-dimethylaminopropyl)-N'-ethylcarbodiimide             |
| EDTA               | ethylenediaminetetraacetic acid                            |
| ELISA              | enzyme-linked immunosorbent assay                          |
| ESC                | embryonic stem cell  |
| ESPA               | equine systemic proteoglycan accumulation                  |
| FBS                | fetal bovine serum   |
| FGF                | fibroblast growth factor                                   |
| FTIR               | Fourier transform infrared spectroscopy                    |
| FuCl               | fumaryl chloride   |
| GAG                | glycosaminoglycan  |
| GAPDH              | glyceraldehyde-3-phosphate dehydrogenase                   |
| GDNF               | glial cell line-derived neurotrophic factor                |
| GMA                | glycidyl methacrylate                                      |
| H&E                | hematoxylin and eosin                                      |
| HB-EGF             | heparin-binding epidermal growth factor-like growth factor |
| <sup>1</sup> H NMR | proton nuclear magnetic resonance                          |
| HRP                | horseradish peroxidase                                     |
| IGF                | insulin-like growth factor                                 |



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| IHC            | immunohistochemistry                  |
|----------------|---------------------------------------|
| Ihh            | Indian hedgehog                       |
| ITS            | insulin-transferrin-selenium          |
| LIF            | leukemia inhibitory factor            |
| MeCl           | methylene chloride                    |
| МК             | midkine                               |
| MMP            | matrix metalloproteinase              |
| M <sub>n</sub> | number average molecular mass         |
| MSC            | mesenchymal stem cell                 |
| $M_{\rm w}$    | weight average molecular mass         |
| MWCO           | molecular weight cutoff               |
| N-cadherin     | neural-cadherin                       |
| N-CAM          | neural cell adhesion molecule         |
| NEAA           | nonessential amino acids              |
| OPF            | oligo(poly(ethylene glycol) fumarate) |
| PBS            | phosphate-buffered saline             |
| PCL            | poly(ɛ-caprolactone)                  |
| PCM            | pericellular matrix                   |
| PCR            | polymerase chain reaction             |
| PDGF           | platelet-derived growth factor        |
| PEG            | poly(ethylene glycol)                 |
| PEG-DA         | poly(ethylene glycol)-diacrylate      |
| PEG-DMA        | poly(ethylene glycol)-dimethacrylate  |
| PGA            | poly(glycolic acid)                   |
| pI             | isoelectric point                     |



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| PI        | polydispersity index   |
|-----------|--|
| PLA       | poly(lactic acid)  |
| PLGA      | poly(lactic-co-glycolic acid)                                |
| PPAR-γ2   | peroxisome proliferator-activated receptor $\gamma 2$        |
| Ptc       | Patched  |
| PTHrP     | parathyroid hormone-related protein                          |
| PTN       | pleiotrophin   |
| RGD       | arginine-glycine-aspartic acid                               |
| RHAMM     | receptor for hyaluronan-mediated motility                    |
| ROCK      | Rho kinase   |
| RT-PCR    | reverse transcription polymerase chain reaction              |
| SAX-HPLC  | strong anion exchange high performance liquid chromatography |
| SEC-HPLC  | size exclusion high performance liquid chromatography        |
| SLRP      | small leucine-rich proteoglycan                              |
| SPR       | surface plasmon resonance                                    |
| Sulfo-NHS | N-hydroxysulfosuccinimide                                    |
| TCPS      | tissue culture-treated polystyrene                           |
| TEA       | triethylamine  |
| TEMED     | tetramethylethylenediamine                                   |
| TGF-β     | transforming growth factor-β                                 |
| TNF-α     | tumor necrosis factor-α                                      |
| Tris      | tris(hydroxymethyl)aminomethane                              |
| UV        | ultraviolet  |
| VEGF      | vascular endothelial growth factor                           |



#### SUMMARY

Tissue engineering strategies represent exciting potential therapies to repair cartilage injuries; however, difficulty regenerating the complex extracellular matrix (ECM) organization of native cartilage remains a significant challenge. Cartilaginous ECM molecules, specifically chondroitin sulfate (CS) glycosaminoglycan, may possess the ability to promote and direct MSC differentiation down a chondrogenic lineage. CS may interact with the stem cell microenvironment through its highly negative charge, generation of osmotic pressure, and sequestration of growth factors; however, the role of CS in directing differentiation down a chondrogenic lineage remains unclear. The overall goal of this dissertation was to develop versatile biomaterial platforms to control CS presentation to mesenchymal stem cells (MSCs) in order to improve understanding of the interactions with CS that promote chondrogenic differentiation.

To investigate chondrogenic response to a diverse set of CS materials, progenitor cells were cultured in the presence of CS proteoglycans and CS chains in a variety of 2D and 3D material systems. Surfaces were coated with aggrecan proteoglycan to alter cell morphology, CS-based nano- and microspheres were developed as small particle carriers for growth factor delivery, and desulfated chondroitin hydrogels were synthesized to examine electrostatic interactions with growth factors and the role of sulfation in the chondrogenic differentiation of MSCs. Together these studies provided valuable insight into the unique ability of CS-based materials to control cellular microenvironments via morphological and material cues to promote chondrogenic differentiation in the development of tissue engineering strategies for cartilage regeneration and repair.



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## CHAPTER 1

#### INTRODUCTION

#### 1.1 Motivation

Cartilaginous tissues play important structural and mechanical roles throughout the body. Particularly in the articular joints, hyaline cartilage distributes loads across the ends of long bones and facilitates motion with low-friction gliding along its surfaces. Unfortunately, cartilage tissue possesses a low capacity for healing, largely because it is a poorly vascularized tissue [1]. As a load bearing tissue, damage to articular cartilage by arthritis and physical trauma can be both debilitating and extremely painful. Of nearly 995,000 arthroscopic procedures performed on the knee in the United States in 2006, 466,000 (47%) of those performed in an ambulatory (outpatient) setting were diagnosed with a tear of the medial or lateral cartilage or menisci. An additional 56,500 (6%) were diagnosed with osteoarthritis of the knee [2]. In 2005, an estimated 27 million adults in the US had clinical osteoarthritis with indirect costs totaling approximately \$89 billion per year [3-4]. To address this growing need, novel tissue engineering therapies seek to promote repair of cartilaginous tissue and restore long-term joint function.

Tissue engineering seeks to promote regeneration of tissue replacements through the combination of cells, scaffold materials, and various soluble, physical, or mechanical differentiation stimuli. While cartilage appears to be a relatively simple tissue that lacks vascularization and nerves and primarily contains only one cell type, challenges to cartilage regeneration include maintaining the phenotype of chondrocytes *in vitro* and recapitulating the complex extracellular matrix (ECM) organization of cartilaginous



tissues [5]. Mesenchymal stem cells (MSCs) are multipotent progenitor cells, found in adult bone marrow, that are capable of differentiating into cartilaginous tissues [6], and recent research has investigated the ability of cartilaginous ECM molecules to direct differentiation of MSCs down a chondrogenic lineage [7-11]. Chondroitin sulfate (CS) is a glycosaminoglycan (GAG) that is prevalent in cartilage, primarily linked with aggrecan proteoglycan. While is it well established that the high negative charge density of CS contributes to cartilage's compressive strength by osmotically retaining water within the tissue matrix [12-13], its role in promoting the development, maintenance, and repair of cartilage tissue is not as well understood.

CS-containing proteoglycans versican and perlecan regulate mesenchymal condensation during cartilage development [14-15], and several enzymes involved in CS initiation, elongation, and sulfation are required for proper skeletal development and patterning [16-18], indicating that CS glycosaminoglycans (GAGs) play an important role in chondrogenesis *in vivo*. Recent studies have investigated the ability of CS-modified biomaterials to promote the production of cartilaginous ECM *in vitro* as well. Aggrecan-coated surfaces promoted the aggregation and production of GAG and collagen II in dermal fibroblasts [10], and culture in CS-containing hydrogels upregulated expression and production of cartilaginous ECM by encapsulated goat and mouse MSCs [8-9]. Sulfated GAGs may alter the extracellular microenvironment via altered osmotic swelling pressure, streaming potential under dynamic loading, and sequestration of growth factors [19-22]. Additionally, downstream effects of cell clustering in CS-modified materials may play a role in differentiation, including increased cell-cell contact [8, 10, 23]. While some of the effects of CS materials on ECM production have been



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broadly characterized, an improved understanding of the role of GAGs in stem cell differentiation is a vital step in developing tissue engineering therapies to regenerate cartilage.

#### **1.2 Research Objectives**

The objective of the research presented is this dissertation was to develop CSbased materials as platforms to control CS presentation to MSCs in order to study the role of CS in chondrogenic differentiation. To investigate the chondrogenic response to CS, MSCs were cultured in the presence of CS proteoglycan and CS GAG chains on 2D surfaces, with nano- and microparticles, and in 3D hydrogels. MSC response on 2D aggrecan-coated surfaces was explored for cell morphology, including aggregation and cell contact, and expression and production of ECM. CS-based materials were also developed, over numerous size scales, to study their role in electrostatic complexation with positively charged growth factors. CS particles were developed for controlled delivery of growth factors within micromass culture, and particles were characterized for size, charge, cytocompatibility, and ability to bind and release growth factor. Given the role of sulfation in the high negative charge density characteristic of sulfated GAGs, a biomaterial platform was developed to explore the role of sulfate moieties in growth factor sequestration through the desulfation of CS. These materials were then used to examine the role of sulfation in electrostatic sequestration of the chondrogenic growth factor transforming growth factor- $\beta$ 1 (TGF- $\beta$ 1) and in subsequent chondrogenic differentiation of MSCs. The goal of these studies was to provide additional insight regarding the interactions with CS that alter stem cell microenvironments to drive



chondrogenic differentiation of MSCs. The **central hypothesis** of this research was that engineering of GAG-based materials to control presentation of CS matrix within a variety of 2D and 3D systems would enhance production of chondrocytic ECM in MSCs. The presence of CS matrix would drive chondrogenic differentiation of MSCs by regulating cellular response through morphological cues and through sequestration of chondrogenic TGF- $\beta$ 1 from the culture medium. The role of CS in chondrogenic differentiation of MSCs was explored in the following three specific aims:

**<u>Hypothesis I:</u>** 2D surfaces modified with cartilaginous GAGs will promote aggregation of bovine MSCs and differentiation toward a cartilaginous phenotype.

**Specific Aim I:** Determine the effect of sulfated GAGs on cellular aggregation and chondrogenic differentiation of bovine MSCs when cultured on 2D surfaces, in the absence of chondrogenic growth factors.

The cartilaginous chondroitin sulfate proteoglycan aggrecan was previously shown to promote cell aggregation and upregulation of GAG production by rabbit dermal fibroblasts [10]; therefore, aggrecan-coated surfaces were investigated as a potential preculture technique to promote the production of chondrogenic ECM prior to cell implantation. The morphology of bovine MSC aggregates was observed on 2D aggrecancoated surfaces, in the absence of chondrogenic growth factors, and anterior cruciate ligament (ACL) fibroblasts were also cultured on aggrecan to examine their potential for the formation of a more fibrochondrocytic phenotype for regeneration of a fibrocartilaginous ligament-bone insertion. Aggregation promotes high density culture and cell-cell contact, similar to that seen in pre-cartilaginous condensations during



cartilage development, suggesting that aggregation may facilitate increased production of cartilaginous ECM by MSCs and ACL fibroblasts [24]. Gene expression and production of ECM were examined over the course of 2 weeks in MSCs and ACL fibroblasts isolated from 3 different bovine donors, compared to nonadhesive surfaces that promoted cell aggregation in the absence of aggrecan and MSC monolayers on unmodified tissue culture-treated polystyrene surfaces.

**<u>Hypothesis</u>** II: Chondroitin sulfate-based nanoparticles and microparticles will electrostatically sequester positively charged growth factors.

**Specific Aim II:** Develop CS-based nanoparticles and microparticles for growth factor delivery and characterize their ability to sequester positively charged growth factors, as a means to direct stem cell differentiation.

Micromass culture is a functional technique in the culture of embryonic stem cell embryoid bodies and chondrogenic MSC pellets; however, dense, multicellular spheroids possess numerous boundaries to diffusion, and growth factor supplementation from the culture medium may result in insufficient or heterogeneous stem cell differentiation. Small particle carriers are valuable tools for controlled delivery to a variety of tissues, and controlled size scale allows for tailored release kinetics, including diffusion rate and degradation properties; therefore, nanospheres and microspheres were fabricated over a range of different size scales from CS materials. CS particles were synthesized and characterized for size, morphology, surface charge, and cytocompatibility. To explore the ability of negatively charged GAGs to electrostatically sequester positively charged proteins, CS microparticles were loaded with the positively charged growth factor TGF-



 $\beta$ 1, and release was observed over 5 days, compared to the negatively charged cytokine tumor necrosis factor- $\alpha$  (TNF- $\alpha$ ).

<u>**Hypothesis III:**</u> Desulfation of CS will diminish its affinity to electrostatically sequester TGF- $\beta$ 1, and desulfated chondroitin hydrogels will promote a weaker chondrogenic differentiation response in MSCs, compared to CS.

<u>Hypothesis IIIA</u>: Desulfated chondroitin materials will have a weaker affinity for sequestration of TGF- $\beta$ 1, compared to CS, due to decreased negative charge density. <u>Specific Aim IIIA</u>: Chemically desulfate CS to produce desulfated chondroitin materials, and determine the effect of sulfation on sequestration and release of TGF- $\beta$ 1.

While previous studies have explored the role of increased GAG sulfation on sequestration of positively charged growth factors [25-26], development of a nonsulfated variant would permit investigation on the role of decreased sulfation, and therefore decreased negative charge density, without modification of the remaining GAG structure. CS was chemically desulfated, and desulfated chondroitin materials were characterized to determine that sulfates were removed from CS without modification of the remaining CS chemical structure. Chondroitin and CS chains were then methacrylated to permit the formation of GAG-containing hydrogels, and modified materials were also examined to determine if chemical modification prevented enzymatic degradation by chondroitinase ABC enzyme. Finally, release of TGF- $\beta$ 1 from CS and chondroitin hydrogels over 7 days and sequestration of soluble TGF- $\beta$ 1 out of solution were measured by enzyme-linked



immunosorbent assay (ELISA) to determine the role of sulfation on interactions with TGF- $\beta$ 1.

**<u>Hypothesis IIIB</u>**: Sulfation of CS plays an essential role in CS-mediated chondrogenic differentiation, and desulfated chondroitin hydrogels will promote a weaker chondrogenic response in encapsulated human MSCs cultured in the presence of TGF- $\beta$ 1, compared to CS.

**Specific Aim IIIB:** Determine the effect of sulfation of chondroitin on the chondrogenic differentiation of human MSCs when encapsulated in CS and desulfated chondroitin hydrogels.

While CS-containing hydrogels have been shown to promote the production of chondrogenic ECM in the presence of chondrogenic media [8-9], the role of sulfation and negative charge density in directing chondrogenic differentiation is not currently well understood. The high degree of sulfation of CS carries a highly negative fixed charge density that facilitates a variety of interactions with cartilaginous ECM, signaling molecules, and interstitial fluid; therefore, sulfation is expected to play an essential role in CS-mediated chondrogenic differentiation. To independently examine the roles of TGF- $\beta$ 1 and sulfation of CS, human MSCs were encapsulated in PEG-based hydrogels containing 50% CS or 50% chondroitin by mass or in PEG-only controls and were cultured for 6 weeks in the presence of medium with or without 10 ng/mL TGF- $\beta$ 1. Encapsulated MSCs were analyzed over 42 days for viability, total DNA, gene expression by quantitative reverse transcription polymerase chain reaction (RT-PCR), and ECM production by immunostaining, compared to PEG-only control hydrogels.



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#### **1.3 Significance and Scientific Contributions**

The studies in this dissertation provide significant insights into the role of sulfated GAGs in altering stem cell microenvironments to control differentiation. A variety of CSbased materials, including 2D surfaces, nano- and microparticles, and bulk hydrogels with varying degrees of sulfation, were developed, and these novel tools provided information on the diverse interactions of sulfated GAGs that may influence differentiation. Culture on 2D aggrecan surfaces provided insight on the (non)adhesive properties of cartilage proteoglycans and the role of cell-cell contact in the production of cartilaginous ECM. GAG-based particles were fabricated over a range of size scales that improve understanding of the electrostatic interactions between sulfated GAGs and positively charged growth factors. Desulfation of CS materials facilitated well-controlled study of the role of sulfation in growth factor sequestration by CS and provided new information on the role of sulfation and growth factor interactions in directing differentiation of MSCs. Controlled presentation of CS through diverse GAG-based biomaterial platforms improved understanding of the various interactions with GAGs that influence stem cell microenvironments, and these principles may be applied to direct differentiation of various multipotent and pluripotent progenitor cells down other noncartilaginous lineages. The knowledge garnered from these studies, in turn, may advance understanding of the role of GAG matrix during development, maintenance, and repair of cartilaginous tissues.

In addition to advancing the current understanding of stem cell differentiation, development of GAG-based materials to control CS presentation has important



implications for tissue engineering and regenerative medicine. Materials that facilitate careful control of cell-cell and cell-matrix interactions could be developed as pre-culture techniques to "prime" stem cell fate prior to implantation. Novel CS-based particles have potential as delivery vehicles to promote improved growth factor transport and more homogeneous differentiation within a variety of tissue and stem cell environments, and careful control over GAG sulfation could contribute to the development of chemically and spatially controlled constructs for tissue regeneration. This research provides additional insight into the contributions of local chemical and biomolecular environments on many broader applications within the fields of cellular differentiation, tissue engineering, and treatment of orthopaedic injuries. Together these findings provide a framework for future investigations into the use of GAG-based biomaterials for cartilage repair and regeneration, and this research will aid in the development of design principles and new strategies to treat of a wide range of health-based problems.



#### **CHAPTER 2**

#### BACKGROUND

#### 2.1 Cartilage

#### 2.1.1 Articular Cartilage Composition and Function

Cartilage is connective tissue that is found throughout the body to perform a variety of mechanical and structural functions [27-28]. Cartilage can be classified as elastic cartilage, hyaline cartilage, or fibrocartilage [27-29]. Elastic cartilage contains high elastin content and is found in the ears and nose, while fibrocartilage has collagen type I and appears in the menisci, annulus fibrosis of intervertebral discs, temporomandibular joint, as well as the insertions of tendon/ligament into bone [27-29]. Hyaline cartilage includes articular cartilage which covers the ends of long bones and facilitates joint loading and motion [27-28]. The structure of articular cartilage facilitates function by providing compressive strength and allowing distribution of load across the joint [13, 30]. Cartilage also facilitates joint motion by providing a low friction surface for gliding [13, 30]. 65-80% of the total weight of hyaline cartilage is composed of water [13]. This fluid interacts with the extracellular matrix (ECM) to absorb loads, minimize peak pressures on the subchondral bone, and lubricate the joint [13]. In articular cartilage, nutrition and elimination of waste is primarily dependent on diffusion to and from the synovial fluid, so the high water content also facilitates the delivery of nutrients and removal of waste products in the largely avascular, aneural, and alymphatic tissue [28].

ECM plays critical roles in cartilage function and maintenance, including protection of chondrocytes from loading forces, storage of cytokines and growth factors,



regulation of mass transport throughout the tissue, and transduction of extracellular signals to the cells [31-32]. Cartilaginous ECM is primarily composed of collagen (60-70% dry weight), proteoglycans (20-35%), and glycoprotein (5-15%), while chondrocytes (1-5% total volume) are responsible for maintaining and remodeling the ECM network [12-13, 28, 31]. Collagen type II is the predominant collagen (90-95% of total collagen) in hyaline cartilage, and collagen II fibers form the primary fibrillar network that provides the tissue with tensile strength [31]. In lesser amounts, collagen IX forms inter-fibrillar connections, and collagen type VI is also present in the pericellular matrix to support chondrocyte attachment and link chondrocytes to the matrix [31]. Collagen type X is only expressed by hypertrophic chondrocytes in the mineralized zone of cartilage to support mineralization and provide structural support [13, 30-31].

Aggrecan is the most prevalent proteoglycan (90% of proteoglycan) found in adult cartilage tissue [31]. Aggrecan is a large 1-3 million Da molecule composed of a core protein with hundreds of sulfated glycosaminoglycan (GAG) side chains [12, 28, 33]. These aggrecan proteoglycans become trapped within the ECM network by aggregating along hyaluronan chains via link proteins to form massive 100-200 million Da structures [12, 28]. The sulfated GAGs in aggrecan, chondroitin sulfate and keratan sulfate, possess negatively charged sulfates and carboxylates, resulting in a highly negative fixed charge density [12-13]. This negative charge density attracts high concentrations of positively charged molecules, while repelling negatively charged molecules, thus increasing the osmolarity of cartilage and creating a Donnan effect [13]. The resulting osmotic pressure causes the tissue to swell and retain water, which is then



constrained from expansion by tension of the collagen network. This fluid-ECM interaction allows the tissue to support high compressive loads while also lubricating joint motion [12, 28]. Compressive loading of the joint causes the internal hydrostatic pressure of the cartilage to increase, and once it exceeds the osmotic pressure of the cartilage, water is pushed out of the ECM, resulting in "weeping" lubrication of the joint [28]. The natural fluid flow that occurs with loading and unloading of the joint also plays an important role to stimulate transport of nutrients and waste within the tissue [28]. In addition to aggrecan proteoglycan, biglycan, decorin, and fibromodulin are small leucine-rich proteoglycans (SLRPs) with shorter protein cores and fewer GAG chains that are found in cartilage tissue. These SLRPs interact with the collagen network to influence fibrillogenesis [31, 34], and have also been shown to bind TGF- $\beta$  to influence cell signaling and function [34-35].

Glycoproteins, on the other hand, consist primarily of protein with only a few attached monosaccharides or oligosaccharides. Structural glycoproteins in cartilage interact with cellular receptors and regulate adhesion, migration, proliferation, and differentiation of chondrocytes [28]. Annexin V and cartilage oligomeric matrix protein (COMP) anchor chondrocytes to the surrounding matrix [31]. Fibronectin and tenascin have important roles in matrix organization and cell-matrix interactions. Fibronectin possesses binding affinity for fibrin, collagen, and heparin, and cell binding is mediated by integrins [28], while tenascin may bind and inhibit cell attachment to fibronectin [36]. Laminin is also present in cartilage and binds integrins to link chondrocytes to the surrounding ECM [37].



Chondrocytes are the cells that are responsible for synthesizing, degrading, and maintaining the complex ECM network of cartilage through normal ECM turnover. Their intracellular components are directly linked to the ECM through receptors on the cell surface [28]. While individual chondrocytes have high individual metabolic activity, due to their low total volume in cartilage the total metabolic activity of chondrocytes is relative low, allowing chondrocytes to function in low oxygen conditions [13, 27]. Chondrocytes, however, are quite sensitive to toxic influences, and even slight fluctuations in pH from physiological 7.4 can disrupt the highly specialized matrix infrastructure [13, 28]. Chondrocytes have also been shown to alter their expression to respond to differences in loading, including the different forces and mechanics experienced throughout the various zones in cartilage [31].

#### 2.1.2 Cartilage Structure and Organization

In addition to the many diverse ECM components, cartilage contains four depthdependent zones with different ECM composition and organization in each: the superficial zone, transitional zone, middle (radial) zone, and calcified cartilage zone. These zones act to facilitate load-dependent deformation in each zone [31]. The superficial zone, found at the surface of cartilage, is the thinnest zone and begins at the surface with a thin sheet of fibrils and film of synovial fluid called lamina splendens [13, 31]. Beneath this sheet, flattened ellipsoid cells lie parallel to the joint surface, within a matrix characterized by high collagen production, low proteoglycan content, and the highest water content of the zones [13, 31]. The collagen in the superficial zone is aligned parallel to the surface to provide tensile and shear strength [13, 31]. The transitional zone provides a spatial and structural intermediate between the superficial and middle zones.



In the transitional zone, spheroid shaped cells are present at a lower cell density than the superficial zone; however, larger diameter collagen fibers are randomly oriented and proteoglycan content is higher than the superficial zone [13, 31]. In the middle (radial) zone, spheroidal cells are arranged in a columnar fashion perpendicular to surface, and this zone has the largest diameter collagen fibrils, aligned perpendicular to the joint surface, as well as the highest proteoglycan concentration and lowest water content and cell density [13, 31]. A tidemark divides the middle and calcified zones, which provides a transition from cartilage to the subchondral bone. In the calcified matrix. These hypertrophic cells have very low metabolic activity and synthesize collagen X, which helps to provide structural integrity and shock absorbance with subchondral bone [13, 31].

The matrix can also be classified into three compartments, based on their cellular interactions, including the pericellular, territorial, and interterritorial matrix regions. The pericellular matrix (PCM) encompasses a thin rim of matrix in close contact with the cell membrane (~2 µm wide) [13]. The cartilaginous PCM is rich in proteoglycan and non-collagenous proteins, as well as non-fibrillar collagen VI, and links the cell surface to the matrix [13, 31]. The territorial matrix surrounds the pericellular region of individual or clusters of chondrocytes. In this region, collagen fibrils are arranged in a crisscrossing manner, forming a fibrillar basket around the chondrocytes and protecting them from mechanical impact [13, 31]. The interterritorial matrix forms most of the volume of the cartilage matrix, and contains large diameter collagen fibrils, oriented differently by zone, as well as proteoglycan aggregates [13, 31].



#### 2.1.3 Cartilage Development

To form the complex ECM structure of cartilage, the development of cartilage and long bones *in vivo* are uniquely linked. The developmental process begins with migration of undifferentiated mesenchymal cells to areas of bone formation, followed by cellular condensation [24]. As the cells condense, they increase their cell packing and cell density, without an increase in proliferation, resulting in an increase in cell-cell contacts, cell-cell adhesion molecules, and gap junctions [24]. Undifferentiated mesenchymal cells begin the process by producing collagen I, hyaluronan, tenascin, and fibronectin; however, condensation and differentiation prompts a shift in ECM production to produce large quantities of collagen II, IX, XI, Gla protein, aggrecan, and link protein [24]. ECM deposition eventually pushes the cells apart to form chondrocytes that are encased in ECM with a characteristic rounded morphology [13, 27].

Cartilaginous condensation and differentiation appear to be directed by a variety of cell-cell and cell-matrix interactions. Prior to condensation, mesenchymal cells increase their production of hyaluronidase, remodeling the hyaluronan-rich matrix to permit increased cell-cell interaction [24]. Fibronectin is also increased in condensations and may facilitate matrix-driven translocation, while tenascin may act to inhibit cell attachment to fibronectin [24]. Within the condensing mesenchyme, neural-cadherin (Ncadherin) and neural cell adhesion molecule (N-CAM) are highly expressed, forming cell-cell contacts that play a critical role in directing differentiation of the cells toward a chondrocytic phenotype [24]. Additionally, hyaluronan receptor CD44 appears to anchor cells to the aggrecan-rich PCM and direct assembly of the chondrocyte PCM during differentiation [24]. Proteoglycans and GAGs may also act as molecular tethers of


soluble factors that play a role in condensation and differentiation, possibly through modulation of the Wnt signaling pathway [24].

During terminal differentiation, chondrocytes undergo hypertrophy, express collagen X and alkaline phosphatase, decrease expression of collagen II, and begin to mineralize the ECM. During endochondral ossification, hypertrophic cartilage is vascularized by the invasion of blood vessels from the perichondrium. Osteoblasts are transported into the tissue through the blood vessels, and begin replacing cartilage with mineralized bone [5, 24]. Indian hedgehog (Ihh) and parathyroid hormone-related protein (PTHrP) signaling may interact to regulate chondrocyte maturation and hypertrophy. It has been proposed that hypertrophic chondrocytes produce Ihh, which acts on Patched (Ptc) and Gli expressing cells in the perichondrium, adjacent to the pre-hypertrophic zone, inducing the expression of PTHrP. PTHrP then signals back to chondrocytes expressing PTHrP receptor and prevents them from proceeding down the hypertrophic pathway in a negative feedback loop that regulates maturation and hypertrophy [38]. Additional Ihh signaling pathways that are independent of PTHrP have also been identified to examine the role of Ihh signaling in promoting chondrocyte hypertrophy [38-40].

#### 2.2 Cartilage Injury and Repair

## 2.2.1 Cartilage Pathology and Healing

Cartilage injury may occur from direct blunt trauma, indirect impact loading, or torsional loading of a joint [41]; however, due to the low proliferative ability and metabolic activity of chondrocytes, cartilage has a low potential for natural healing.



Additionally, cartilage does not have direct access to progenitor cells, the dense ECM may impede cell migration, and proteoglycans tend to resist cell adhesion, further undermining the healing process [1, 42-43]. If severe damage is allowed to persist, the tissue may degenerate further and progress into development of osteoarthritis, which is a chronic disease characterized by wear and tear of the cartilage surface [44]. Cartilage damage can generally be classified into three types of injury, depending on the depth of the defect: matrix disruption, partial thickness defects, and full thickness defects. During matrix disruption, the damage to the ECM is relatively mild, and viable chondrocytes are capable of repairing the tissue by increasing their natural synthetic ability [41, 45]. Partial thickness defects describe more severe injuries that disrupt the cartilage surface but do not extend to the subchondral bone. In partial thickness injuries, the surrounding chondrocytes respond by increasing their proliferation, but cellular attempts to naturally repair the tissue cease before the defect is fully healed [41, 45]. Full thickness defects, on the other hand, transverse the entire cartilage thickness and penetrate the subchondral bone. In this case, subchondral blood vessels are disrupted, defects are filled with a fibrin clot, and the classic wound healing response ensues [41, 45]. Unlike partial thickness defects, access to the subchondral bone permits access to a population of progenitor cells from the bone marrow, which can migrate to fill the defect. At the end of the healing process, the fibrin clot is replaced with an intermediate tissue with properties between those of hyaline cartilage and fibrocartilage that is typically less stiff and more permeable than healthy cartilage [27, 41-42].



## 2.2.2 Clinical Interventions for Cartilage Repair

While a total knee arthroplasty, in which the joint is replaced with artificial surfaces, is the most common treatment for end stage osteoarthritis, the surgery is extremely invasive, requires replacement every 10-15 years, and is not ideal for young patients with active lifestyles [13, 46]. Osteotomy can also relieve severe joint pain by removing part of the bone in order to redistribute load within the joint surface and correct misalignment of the joint [47-48]; however, these techniques are invasive and are typically only employed after less invasion interventions have failed [13, 27, 49].

To encourage the natural healing response, one strategy to treat cartilage defects involves intentionally penetrating the subchondral bone in order to stimulate the bone marrow and promote the formation of a fibrin clot. The impetus for this technique is to disrupt subchondral blood vessels, similar to full thickness defects, to form a fibrin clot and induce bone marrow-derived chondroprogenitors to migrate into the lesion, proliferate, and differentiate. [50-51]. Bone marrow stimulation may also enhance expression of cytokines to promote repair [52]. Techniques include subchondral drilling, abrasion, and microfracture, and are commonly used in conjugation with debridement to remove necrotic tissue from the joint surfaces [53-56]. Marrow stimulation techniques are fairly easy to perform and relatively low cost; however, similar to full thickness healing, the tissue is often replaced with fibrous or fibrocartilaginous tissue, that is more prone to future degeneration [50-51, 56].

In another approach, autografts are taken from non-load bearing regions to replace the injured cartilage tissue. In a mosaicplasty, cylindrical osteochondral plugs are harvested from non-weight bearing regions of the knee and transplanted into the defects.



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While this approach replaces the damaged tissue with hyaline cartilage, the gaps between the plugs are often replaced with fibrocartilage, resulting in a lack of integration with the native tissue, as well as increased permeability [57-58]. As with all autografts, donor site morbidity and limited availability of autologous tissue limits this approach, especially for repair of large lesions [51, 59]. Perichondrium and periosteum have also been used as sources for autologous grafts, due to their chondrogenic and osteogenic potential; however, despite some promising results, they are still unable to consistently replace healthy cartilage tissue [13, 60-62].

More recently, autologous chondrocytes have been cultured in vitro and reimplanted to promote healing. Autologous chondrocyte implantation (ACI) involves taking a tissue biopsy from a non-weight bearing region of cartilage, enzymatically isolating chondrocytes from the tissue, and expanding the excised chondrocytes for 2-3 weeks in vitro [13]. After expansion, the defect is covered with an autologous periosteal flap, and a suspension of expanded chondrocytes is injected underneath the patch into the defect [13]. This approach has yielded some promising results; however, limitations include occasional leakage of transplanted cells, an invasive surgical method, hypertrophy of the periosteum, and the loss of chondrocytic phenotype associated with expansion of chondrocytes in culture [51, 63-65]. In addition, the regenerated cartilage is often more fibrous in nature, possibly due to the low proliferation potential of chondrocytes [66-67]. Overall, there has been large variation and contrasting degrees of success when comparing the current intervention techniques for cartilage repair. Jakobsen et al. evaluated sixty-one clinical cartilage repair studies including a total of 3,987 surgical procedures using microfracture, autologous osteochondral transplantation



(mosaicplasty), autologous periosteal transplantation, and ACI [68]. Large variation was observed between each treatment modality, and no significant differences were found across techniques, suggesting that an improved clinical approach is necessary to properly repair cartilage after injury [51].

## 2.3 Tissue Engineering and Regenerative Medicine

## 2.3.1 Tissue Engineering of Cartilage

The limited ability of cartilage to heal and the limitations of surgical repair have introduced the need for tissue engineering strategies for cartilage regeneration. Tissue engineering typically applies a combination of biological, chemical, and engineering principles to regenerate functional tissue. The typical tissue engineering approach uses a 3D scaffold to promote tissue formation, cells that can be expanded and maintained *in vitro*, and the use of differentiation factors to drive tissue development [69]. Bioreactor systems have also been developed to apply mechanical or physical stimulation to the tissue and to facilitate precise control of the extracellular culture environment [70]. Cartilaginous tissues appear to be a strong fit for tissue engineering strategies, because of their relative simplicity as a tissue. Cartilage is largely avascular, lacks nerves, and contains primarily only one cell type; however, barriers to cartilage regeneration include maintaining the phenotype of chondrocytes *in vitro* and recapitulating the complex ECM organization of cartilaginous tissues [5].



## 2.3.2 Current Tissue Engineering Approaches

#### 2.3.2.1 Cell Sources

An obvious choice of cell type for cartilage tissue engineering is chondrocytes, since chondrocytes are the predominant cell type in cartilaginous tissue and are responsible for maintaining the ECM. Unfortunately, chondrocytes have a limited proliferative capacity *in vitro* and tend to de-differentiate in extended culture, especially in monolayer, becoming increasingly fibrous in nature [71-72]. 3D encapsulation in a variety of materials improves maintenance of chondrocyte phenotype; however, culture in these materials also slows cell growth. Therefore a number of bioreactors, including spinner flasks, microcarrier suspensions, perfusion systems, and rotating-wall bioreactors have been developed for mass culture of chondrocytes to maintain mature phenotype [73-76].

Mesenchymal stem cells (MSCs) are adult stem cells present in the bone marrow that can selectively differentiate into any mesenchymal cell type, including osteoblasts, chondrocytes, adipocytes, myoblasts, and ligament fibroblasts [77-79]. Marrow cells can be noninvasively collected from adult bone marrow by needle biopsy, and MSCs can be expanded *in vitro*, making them a promising cell source for cartilage tissue engineering. MSCs are mostly commonly differentiated in micromass/pellet culture to promote chondrogenesis [6]. High cell density and close cell-cell contact mimics mesenchymal condensation observed during cartilage development [24]. Common markers for chondrogenic differentiation include ECM markers collagen II, aggrecan, and cartilage oligomeric matrix protein (COMP), as well as SOX-9 transcription factor [80-82]. MSClike progenitor cells, with similar colony forming ability and multilineage potential, have



also been isolated from adipose tissue, periosteum, synovium, skeletal muscle, tendon, trabecular bone, and umbilical cord blood [83-89]; however, the differentiation potential of each population varies. MSC-like cells have also been derived from embryonic stem cells, which are pluripotent cells isolated from the inner cell mass of the blastocyst [90]. Embryonic-derived MSCs were capable of producing cartilage-like tissue *in vitro* and *in vivo* [91].

## 2.3.2.2 Differentiation Factors

A variety of soluble biochemical factors have been shown to influence MSC differentiation toward a cartilaginous phenotype. Most commonly medium containing transforming growth factor- $\beta$  (TGF- $\beta$ ) and dexamethasone is established to drive MSCs toward a chondrogenic phenotype [6]. A number of TGF- $\beta$  superfamily proteins are known play a crucial role in cartilage formation. In particular, TGF- $\beta$ 1, - $\beta$ 2, and - $\beta$ 3 are highly expressed in pre-cartilaginous condensations of the developing mesenchyme [24, 92-93], and all three isoforms possess chondrogenic potential [94-97]. Bone morphogenetic proteins (BMPs) including BMP-2, -4, and -6 have also been shown to enhance chondrogenic differentiation of MSCs [98-99]. BMP-2, especially, has been demonstrated to exhibit an additive differentiation response in MSC pellet cultures when supplemented in combination with TGF- $\beta$  [99]. Similarly, insulin-like growth factor-1 (IGF-1) has been shown to upregulate expression of chondrocytic markers in MSCs [100-102]. IGF-1 has independent signaling pathways to TGF- $\beta$  and is capable of promoting a comparable chondrogenic response; however, IGF-1 has also been shown to promote an additive response when supplemented either in combination or sequentially with TGF- $\beta$ [101-102].



Unfortunately, chondrogenic MSCs cultured under these conditions tend to undergo hypertrophy, as the cells produce collagen X and alkaline phosphatase and undergo terminal differentiation [97, 103-104]. Chondrogenic MSCs also do not entirely suppress expression of collagen I, often resulting in a more fibrous tissue [105]. Basic fibroblast growth factor (bFGF) and PTHrP have been investigated for their ability to delay hypertrophy and maintain chondrocytic phenotype *in vitro* [106-109]. Recently, Gawlitta et al. published that oxygen tension may also play a role in chondrogenic differentiation and hypertrophy. In this study, hypoxic conditions (5% O<sub>2</sub>) were capable of delaying hypertrophy of chondrogenic MSC pellets [110].

## 2.3.2.3 Scaffold Materials

Various natural and synthetic materials have been investigated as scaffolds to support cartilage regeneration. Various synthetic hydrophobic polymers can be extruded or electrospun into micro- or nanofibers, and these fibers can be layered to form porous sponges. The resulting fibrous meshes are mechanically stiff, and MSCs can be seeded on and between the fibers [111]. Hydrolytically degradable materials, including poly(lactic acid) (PLA), poly(glycolic acid) (PGA), their copolymer poly(lactic-co-glycolic acid) (PLGA), and poly(ɛ-caprolactone) (PCL) have been shown to support chondrogenic differentiation and production of cartilaginous ECM by MSCs [112-116]. As these hydrolytically labile materials degrade, the scaffold can be replaced with ECM deposited by the embedded cells. These materials and their copolymers can also be chemically modified to control degradation rate [117] and modified with bioactive proteins, peptides, and molecules to support bioactivity [118-121].



Due to the high water content of native cartilage, hydrogel materials are popular scaffolds for 3D culture and chondrogenic differentiation of MSCs. Through the crosslinking of polymer chains, hydrogel scaffolds can be swelled in aqueous solution to create highly hydrated network structures [122-123]. Cells can be homogenously encapsulated within polymer networks and cultured for extended periods of time [124]. Encapsulation of MSCs reconstructs a 3D tissue environment in which the cells are immobilized within matrix with a rounded morphology, characteristic of chondrocytes, without the need for cell adhesion [71]. Prior to crosslinking, macromer solution can also be injected to a site of injury and crosslinked *in situ* for minimally invasive surgical techniques and cell delivery [124].

Naturally-derived hydrogels for cartilaginous tissue engineering include collagen types I and II, fibrin, gelatin, and hyaluronan [11, 81, 125-129]. Along with the inherent biocompatibility of natural materials, encapsulated cells and bioactive molecules can actively interact with these materials and remodel the matrix over time. While alginate and agarose are naturally-occurring polysaccharides with application in chondrogenic culture, they are derived from marine algae and are cannot be produced or degraded by mammalian cells [81, 129-133]; therefore, they can be utilized to limit cellular remodeling when desirable. Unfortunately, the inherent complexity of natural scaffolds, along with batch-to-batch variability and limitations on chemical modification, are significant limitations for cartilage regeneration [134].

While synthetic hydrogels do not possess inherent bioactivity, these materials can be precisely engineered to exhibit specific chemical and mechanical properties, and bioactive functionality must be expressly designed into the system [135]. Poly(ethylene



glycol) (PEG) is a common synthetic polymer used to support chondrogenic differentiation of MSCs [136-137]. By engineering bioactive moieties into the polymer network in a functionally, spatially, and temporally tailored fashion, one can study specific cell stimuli and responses in a discrete and controlled way. Design of PEG-based hydrogels for chondrogenic differentiation of MSC will be discussed in further detail in Section 2.3.3.

#### 2.3.2.4 Physical Stimuli

Due to the established importance of mechanical forces in the development, maintenance, and remodeling of orthopaedic tissues, mechanical loading has been investigated as a technique to stimulate cartilage tissue formation. Cyclic, compressive loading has been shown to increase production of cartilage specific ECM, including collagen II and aggrecan, in MSCs encapsulated in a variety of hydrogel materials [125, 138-141]. Similarly, application of cyclic, hydrostatic pressure increased collagen and proteoglycan production in encapsulated chondrogenic MSCs [142], as well as in scaffold-free MSC aggregates [143]. Huang et al. found that cyclic compression improved the matrix distribution and increased mechanical stiffness of chondrogenic MSC-laden agarose scaffolds after long-term culture for 42 days [144]. In studies by Kisiday et al., proteoglycan content was found to increase in agarose gels in response to loading after 15 days, even in the absence of chondrogenic cytokines [145].

Other physical stimuli, including cell shape and the stiffness of the surrounding matrix, interact with chemical, molecular, and genetic factors to regulate stem cell fate [146]. McBeath et al. demonstrated that cell shape and area regulated osteogenic and adipogenic differentiation in MSCs [147]. Small cell areas and rounded shape dictated a



more adipogenic fate, while large areas and spreading promoted an osteogenic phenotype in identical mixed media, via RhoA/Rho kinase (ROCK) regulation of cytoskeletal tension. Kilian et al. followed up this finding by showing that cell shapes that enabled higher acto-myosin contractility, even under identical cell areas, promoted osteogenic differentiation over adipogenic differentiation [148]. These studies implicated a relationship between cytoskeletal organization and differentiation lineage.

In chondrocyte culture or chondrogenic differentiation of MSCs, it has been well established that micromass culture and encapsulation, which maintain a rounded cell shape, are beneficial to maintaining or promoting a chondrocytic phenotype, compared to monolayer culture, suggesting that cell shape, cytoskeletal tension, and matrix interactions may also play important roles in chondrogenesis [149]. Gao et al. demonstrated that in the presence of TGF- $\beta$ 3 a rounded cell shape promoted chondrogenic differentiation of MSCs on 2D surfaces, while spreading promoted a myogenic smooth muscle cell phenotype [150]. In addition, while RhoA exhibited little change in activity, another GTPase Rac1 inhibited chondrogenesis, upregulated expression of N-cadherin, and induced smooth muscle differentiation. In an examination of nucleus shape, McBride and Knothe Tate also demonstrated that rounder nuclei were associated with greater expression of chondrogenic markers by C3H/10T1/2 stem cells on 2D surfaces [151-152]. A series of studies on 2D surfaces demonstrated that chemical disruption of cytoskeletal tension and inhibition of RhoA/ROCK signaling promoted expression of chondrogenic markers by stem cells [153-157]; however, RhoA/ROCK and Rac1 signaling has been found to be vastly different in 3D micromass culture. Woods and Beier observed that in 3D micromass cultures of primary mouse mesenchymal limb bud



cells, RhoA/ROCK inhibition instead decreased expression of chondrogenic markers [158]. Furthermore, Woods et al. showed that Rac1 promoted chondrogenic differentiation in micromass culture through enhanced N-cadherin expression [159]. These studies suggest that while cytoskeletal tension and RhoA-ROCK and Rac1 signaling may play important roles in chondrogenic differentiation, significant differences occur between 2D and 3D culture, possibly due to inherent differences in cell-cell interactions, cell-matrix interactions, and cell shape.

Along with geometry, matrix stiffness also influences stem cell lineage. In a seminal paper by Engler et al., stiff poly(acrylamide) substrates (Young's modulus 25-40 kPa) directed MSCs toward a more osteogenic phenotype, compared to moderate stiffness (8-17 kPa) for myogenic, and soft (0.1-1 kPa) for neurogenic differentiation. As formation of focal adhesions increased on stiffer substrates, nonmuscle myosin II was required for matrix stiffness-based direction of MSC lineage [160]. For culture of chondrocytes, it appears that relatively soft substrates which limit cytoskeletal tension encourage chondrogenic phenotype. Schuh et al. observed that softer poly(acrylamide) substrates (Young's modulus 4 kPa) promoted greater production of collagen II and aggrecan by chondrocytes, along with decreased proliferation and actin organization, compared to stiffer substrates in 2D culture [161]. In addition, Park et al. determined that stiffness regulated MSC response to TGF- $\beta$ 1, promoting expression of chondrogenic markers on soft collagen I and poly(acrylamide) substrates and myogenic phenotype on stiffer substrates [162]. Overall, these studies have established that cell shape and matrix stiffness appear to regulate cell phenotype through cell-matrix interactions and cytoskeletal organization and tension in 2D. In 3D culture, however, these effects are



more difficult to discern, because cell shape and matrix stiffness are also highly dependent on the material properties of the scaffold, including crosslinking density, porosity, and resulting transport properties. In 3D, local ECM organization also regulates cellular interactions with the matrix. While Schuh et al. found that softer agarose constructs supported greater GAG production by encapsulated chondrocytes, softer constructs also supported formation of larger cell clusters and a different local ECM environment than stiffer gels [163].

## 2.3.3 PEG-Based Hydrogels for Cartilage Tissue Engineering

Poly(ethylene glycol) (PEG) is a nonadhesive synthetic material that is highly resistant to protein adsorption, making it an attractive material for cartilage tissue engineering [164-166]. These non-fouling properties enable PEG-based materials to act as a "blank slate" upon which additional bioactive functionality can be specifically tailored into the hydrogel formulation. PEG hydrogels have been extensively investigated for bone, cartilage, vascular, and neural engineering [136, 167-173]. To crosslink and form PEG-based hydrogels, PEG is most often chemically modified to include acrylate groups. The resulting PEG-diacrylate (PEG-DA) or -dimethacrylate (PEG-DMA) can be crosslinked via free radical initiation, where polymerization occurs through a chaingrowth mechanism that involves chain transfer of the radical to a free double bond on another acrylate group [174]. Radical initiators include the thermally responsive tandem ammonium persulfate (APS) and tetramethylethylenediamine (TEMED), as well as the photosensitive Irgacure 2959 [175-178]. Alternative step-growth crosslinking mechanisms, including Michael-type addition and "click" chemistry, have been utilized for crosslinking of PEG-based materials, and these techniques can also be used together



with chain-growth initiators for sequential or mixed-mode crosslinking reactions [174, 179-182].

PEG's mechanical and biochemical properties can be easily modified for a variety of tissue engineering applications [167, 183-184]. Hydrolytically labile components have been added into PEG networks to control degradation [169, 185], and enzymatically degradable peptides have also been incorporated within PEG hydrogels for cell-mediated degradation [186-187]. More recently, novel photodegradable groups have been investigated as a means to degrade PEG networks on demand in the presence of ultraviolet (UV) light [188]. Functional peptides like the adhesive peptide arginineglycine-aspartic acid (RGD) and growth factors including TGF- $\beta$ , bFGF, and vascular endothelial growth factor (VEGF) have been tethered into PEG networks to modulate cell response [171, 189-191]. Additionally, PEG hydrogels are capable of micropatterning to create well-defined physical and biochemical features via photolithography [192-197].

Oligo(poly(ethylene glycol) fumarate) (OPF) is a PEG-based, hydrolytically degradable hydrogel material that supports various *in vivo* and *in vitro* tissue engineering applications, including chondrogenic and osteogenic differentiation of MSCs [175, 198-202]. OPF's swelling, degradation, and mechanical properties can be easily controlled by altering the molecular weight of its initial PEG chains [45, 202]. Using PEG-DA as a crosslinker, OPF/PEG-DA scaffolds have been modified with functional peptides, including RGD [201], and OPF hydrogels have also been utilized as a delivery vehicle for a variety of growth factors [203-205]. OPF is a versatile and biocompatible hydrogel biomaterial with potential for use in the regeneration of a variety of tissues.



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For these reasons, PEG and OPF-based materials have been used in controlling chondrogenic differentiation of MSCs [136, 199, 206-208]. PEG-DA hydrogels encourage increased production of cartilaginous matrix, over monolayer culture, in chondrogenic media [136], and the network structure and mesh size can be precisely controlled to influence differentiation [207, 209]. Many forms of biofunctionality have also been designed into the hydrogel system to promote enhanced bioactivity, including various ECM molecules and biological mimics. Most commonly the adhesive peptide RGD has been incorporated into PEG-based hydrogels to promote adhesion and MSC viability [210-211]. The role of RGD, an integrin binding peptide located in the  $III_{10}$ repeat of fibronectin, in chondrogenic differentiation of MSCs has yet to be fully elucidated [212]. While fibronectin is expressed in pre-chondrogenic condensations during cartilage development, RGD incorporation into PEG hydrogels has been shown to increased differentiation of chondrogenic MSCs [211], while RGD inhibited chondrogenesis in agarose gels [213]. Hypothesizing that RGD may promote early chondrogenic differentiation of MSCs, Salinas et al. used a matrix metalloproteinase (MMP)-sensitive peptide to achieve temporal presentation of RGD to cells [137]. As the cells underwent early differentiation in the presence of RGD, the RGD was enzymatically cleaved and released from the network by cell-secreted MMP-13. After 21 days, chondrogenic MSCs encapsulated with cleavable RGD moieties exhibited significantly greater production of cartilaginous ECM than those with uncleaveable RGD, suggesting that both spatial and temporal presentation of biological cues may be necessary for regeneration. Additional work has incorporated collagen types I and II, collagen mimetic peptides GFOGER and (POG)<sub>7</sub>, decorin sequence KLER, and GAGs hyaluronan and



chondroitin sulfate to examine their roles in chondrogenic differentiation [7-9, 214-216]. Together, the research presented here strongly suggests that ECM interactions play a critical role in directing MSCs down a chondrogenic lineage for cartilage repair and regeneration. Specifically, chondroitin sulfate (CS) is a cartilaginous GAG that plays important structural and biological roles in cartilaginous tissues; however, the role of CS in chondrogenic differentiation and the interactions between CS and MSCs are not well understood.

#### 2.4 Chondroitin Sulfate Interactions to Promote Chondrogenic Differentiation

## 2.4.1 Role of CS Proteoglycans in Chondrogenesis

Perlecan and versican are proteoglycans that play important roles in tissue development. CS proteoglycan versican influences embryonic cell adhesion, migration, proliferation, and ECM assembly in a variety of tissues [217], and versican exhibits increased expression during mesenchymal condensation in early cartilage development [14, 218-219]. Versican may act as an anti-adhesion molecule during the initiation of matrix assembly, and versican expression ceases after deposition of aggrecan in the ECM [220]. Perlecan is also essential for proper cartilage development [221-223]. Perlecan is expressed within cartilaginous condensations after expression of collagen II and aggrecan, and remains in the adult pericellular matrix where it interacts with other ECM molecules including laminin and fibronectin [224]. Perlecan containing both heparan sulfate and CS has been implicated in bFGF signaling to regulate chondrocyte proliferation during endochondral ossification in the growth plate [15, 223, 225-226]. Heparan sulfate chains bind bFGF in perlecan, sequestering it from FGF receptors, while



CS chains on perlecan appear to block FGF receptor binding [15]. Additionally, aggrecan also plays a critical role in skeletal formation and patterning, ECM production, and collagen fibrillogenesis [33, 227]. The major influence of CS proteoglycans in chondrogenesis suggests that CS GAGs may play an important role in directing chondrogenic differentiation.

Furthermore, chondroitin sulfate production and patterning play critical roles in the development of cartilage and the skeleton, as mutations affecting production and patterning of CS are largely not viable. A recent study by Wilson et al. demonstrated that chondroitin sulfate synthase 1 (Chsy1), which is one member of a family of enzymes responsible for extension of chondroitin sulfate chains, is required for proper bone development and joint/digit patterning in mice [16]. Chsy1 mutations resulted in chondrodysplasia, decreased bone density, and abnormal digit patterning, possibly resulting from abnormal sulfation of CS. Similarly, Watanabe et al. showed that chondroitin sulfate N-acetylgalactosaminyltransferase 1 (CSGalNAcT1), another enzyme responsible for the initiation and elongation of CS chains, is required for normal cartilage development, as knockout in mice resulted in 50% decrease in chondroitin sulfate production and significantly thinner growth plate cartilage with disorganized arrangement of collagen fibers [17]. Kluppel et al. also showed that chondroitin-4-sulfotransferase 1 (C4st1) expression, which is responsible for transfer of sulfate groups onto the 4-O position of chondroitin, was required for development and growth factor signaling during cartilage morphogenesis in mice [18]. C4st1 mutations demonstrated underexpression and mislocalization of CS in the growth plate, resulting in disorganized ECM and abnormal chondrocyte orientation. Defective CS balance in C4st1 mutants also resulted



in upregulation of TGF- $\beta$ 3 signaling, and downregulation of BMP, suggesting a role in growth factor signaling. Interestingly, while deletion of chondroitin-6-sulfotransferase 1 (C6st1), which transfers sulfates onto the 6-O position, did not affect skeletal development in mice, mutations in C6st1 in humans are associated with chondrodysplasia [228].

CS structure is regulated during cartilage differentiation and development, as sulfation patterns may regulate the affinity of CS to bind growth factors [229-230], and CS sulfation and structure continues to change through maturation and into adulthood. In human articular cartilage, CS is primarily monosulfated on either the 4- or 6-carbon of Nacetylgalactoseamine; however, CS structure and patterning in cartilage continues to change through adulthood. As the tissue matures and growth cartilage is replaced by adult cartilage, the ratio of 6-sulfated CS to 4-sulfated CS increases from ~0.77 to ~23, indicating a shift in the sulfation balance by adulthood [20, 231]. Additionally, sulfation pattern varies by zone, as deeper zones of cartilage contain more 4-sulfated CS than superficial zones [20]. In addition to sulfation pattern, the average CS chain length also decreases over time. The average chain size decreases from ~32 kDa in fetal cartilage to ~20 kDa in postnatal cartilage, then down to ~8 kDa by skeletal maturity [231]. The change in structure and sulfation pattern from fetal cartilage to adulthood suggests that CS may actively regulate its structure over time to facilitate specific interactions and signaling mechanisms within the maturing tissue.

## 2.4.2 Chondroitin Sulfate Hydrogels for Tissue Engineering

Recent work has investigated the ability of ECM molecules to direct stem cell differentiation down specific lineages, including GAGs for chondrogenic differentiation.



Culture on aggrecan-coated surfaces was shown to promote cell aggregation and production of GAG and collagen II in rabbit dermal fibroblasts, following pretreatment with IGF-1 [10]. Studies have also shown that encapsulation of chondrocytes and mesenchymal stem cells within hyaluronan-based hydrogels increased production of cartilaginous ECM [11, 232-236], suggesting that cells interact with GAGs and that CS hydrogels may be a useful tool in promoting chondrogenic differentiation of MSCs.

Various research groups have developed CS-containing hydrogels as tool to study 3D interactions with cells, most often in combination with PEG. The Elisseeff research group has developed several forms of crosslinkable CS-based tissue adhesives to promote repair and regeneration. By chemically modifying CS with methacrylates, aldehydes, or N-hydroxysuccinimde, CS adhesives were crosslinked to themselves, PEG-DA, or with the amine groups present in proteins. These adhesives have demonstrated the ability to bond articular cartilage and cornea tissue with good biocompatibility [237-241].

CS proteoglycans also play important roles in directing growth in neural systems, so Conovaloff and Panitch used CS/PEG hydrogels to study neurite growth for neural regeneration applications [242]. Chondroitin-6-sulfate was shown to bind NGF, BDNF, and NT-3 cytokines with high affinity, compared to heparin, by capillary electrophoresis, while hyaluronan bound NGF with higher affinity than CS. Culture of E8 chick dorsal root ganglia in the presence of NGF demonstrated more robust neurite outgrowth in 2% CS/98% PEG-DA hydrogels, compared to 2% hyaluronan/98% PEG-DA after 3 days, suggesting CS acts as a better scaffold than hyaluronan despite weaker affinity for NGF, possibly due to inhibition of growth by HA. While this study suggested that CS/PEG hydrogels support neurite growth, this result contrasts other studies that have shown that



CS and various CS proteoglycans inhibited neurite growth in glial scar and that inhibition is dependent on the degree of sulfation [243-247]; however, differences in sulfation pattern, CS presentation without the proteoglycan core protein, and growth factor supplementation may account for these differences.

As a prominent GAG in cartilaginous tissues, CS/PEG hydrogels have also been developed for 3D culture of chondrocytes. Bryant et al. crosslinked varying amounts of methacrylated chondroitin-4-sulfate into PEG hydrogels to alter the macroscopic properties of the hydrogels. 40% CS/60% PEG hydrogels were shown to promote greater production of cartilaginous ECM by chondrocytes, than CS-only gels, suggesting that the addition of PEG-DA was necessary to support ECM deposition in these CS-containing hydrogels [248]. To compare CS hydrogels to other ECM components, Hwang et al. showed that chondrocytes encapsulated in CS/PEG hydrogels supported greater accumulation of cartilaginous matrix by superficial and deep zone chondrocytes than gels with collagen type I or hyaluronan, suggesting that CS possesses tissue specific benefits for maintaining chondrocytic phenotype [249]. Furthermore, to investigate the role of mechanical loading in CS/PEG hydrogels, Villanueva et al. demonstrated that 20% CS/80% PEG hydrogels simulated greater production of GAG and collagen by bovine chondrocytes under dynamic compression, compared to loaded PEG gels [21]. These studies suggest that CS/PEG hydrogels interact with chondrocytes in 3D hydrogels and may support the production of cartilaginous ECM.

Sulfated GAGs generate osmotic swelling pressure in cartilaginous tissues as a result of the associated high negative charge density [20]. Mobile positive ions in solution rush into the tissue to shield the negatively charged GAGs, and the resulting osmotic



pressure provides the tissue with enhanced ability to withstand compressive loads [20]. Osmotic swelling pressure also facilitates streaming potential during loading, in which the mobile positive ions are forced in and out of the matrix resulting in an electric potential [21, 250]. Sulfation, osmotic swelling, and streaming potential appear to influence ECM production in chondrocytes [21]; however, its role in differentiation of MSCs remains unclear.

To examine the role of CS interactions in differentiation of MSCs, Varghese et al. examined the chondrogenic differentiation of MSCs in CS/PEG hydrogels. Goat bone marrow MSCs aggregated in 50% CS/50% PEG-DA hydrogels and upregulated production of chondrocytic markers in chondrogenic media, compared to PEG-only controls [8]. Cadherin11 expression was also temporally upregulated at early stages of aggregation, suggesting that increased cell-cell contact in CS materials may encourage a more chondrogenic phenotype. Additional analysis by Nguyen et al. suggested that controlled presentation of the CS matrix could direct differentiation toward the different cartilaginous zones [9]. 50% CS/50% PEG-DA hydrogels promoted production of high levels of collagen II and proteoglycan similar to that found in transitional zone of cartilage by D1 mouse bone marrow stem cells. Meanwhile, incorporation of MMPcleavable peptide sequences into the CS/PEG hydrogel promoted a more superficial zone-like ECM with low proteoglycan, and hyaluronan/PEG produced proteoglycan production with lower collagen II similar to the middle (radial) zone of cartilage. CS-only hydrogels, on the other hand, produced higher levels of collagen X, reminiscent of the calcified zone of cartilage.



Of these studies investigating CS hydrogels, few studies have investigated the effect of CS on human MSCs. Chen et al. observed that 2D surfaces with crosslinked chondroitin-6-sulfate supported greater differentiation of human MSCs than MSCs cultured with CS and CS oligosaccharide present in the culture medium [251]. Steinmetz and Bryant also examined the effect of cyclic compressive loading on the chondrogenic differentiation of human MSC in CS/PEG hydrogels [252]. While the effects of CS in 20% CS/80% PEG-DA hydrogels were not as apparent after 14 days as previously reported in animal-derived MSCs with little ECM accumulation, the results suggested that CS may slow the terminal differentiation process in response to loading, as observed by downregulation of collagen X production and Runx2 expression.

## 2.4.3 Growth Factor Sequestration by Chondroitin Sulfate

Charged GAGs, primarily associated with proteoglycans *in vivo*, play a major role in sequestration of growth factors, stabilizing growth factors and preventing denaturation [22, 253-254]. Negatively charged GAGs, including heparin, heparan sulfate, and CS, bind positively charged growth factors, such as bFGF, IGF, VEGF, platelet-derived growth factor (PDGF), and TGF- $\beta$ , by electrostatic interactions [255-264]. GAGs electrostatically complex with growth factors due to their highly negative charge density, and increased degree of sulfation appears to correlate with stronger electrostatic interaction. Heparin binds with bFGF, IGF, VEGF, and glial cell line-derived neurotrophic factor (GDNF) in a sulfation-dependent manner and with special importance of 2-O-sulfation for binding [265-272]. While a majority of studies have focused on heparin and heparan sulfate, due to their relatively higher charge densities, CS has also been shown to bind to various growth factors *in vitro* [22, 261-264], indicating that



sequestration of chondrogenic growth factors by CS hydrogels may be a promising strategy to enhance chondrogenic differentiation of MSCs.

Deepa et al. found that CS-E, which is 4,6-disulfated, bound directly to a variety of heparin-binding growth factors including midkine (MK), pleiotrophin (PTN), heparinbinding epidermal growth factor-like growth factor (HB-EGF), FGF-2, FGF-10, FGF-16, and FGF-18. [261]. Binding affinity to CS-E was greater than or comparable to that observed in heparin, and growth factors bound CS-E in a specific and concentration dependent manner. Additionally, Hintze et al. demonstrated that chemical sulfation of CS and hyaluronan enhanced affinity for TGF- $\beta$ 1 and BMP-4 [25, 273]. Electrostatic binding was enhanced as degree of sulfation was increased from 0 to 3 sulfates per disaccharide, and interaction was inhibited by increasing the salt concentration of the buffer and through competitive inhibition by soluble GAGs, indicating that CS possesses sulfatedependent affinity for chondrogenic growth factors of the TGF- $\beta$  superfamily.

In an examination to determine the binding mechanism of sulfated GAGS to TGF- $\beta$ , Lyon et al. demonstrated that heparin and highly sulfated liver heparan sulfate electrostatically bound TGF- $\beta$ 1 and TGF- $\beta$ 2 by affinity chromatography, while TGF- $\beta$ 3 isoform did not [274]. The bound fraction was eluted with  $\geq$ 0.5 M NaCl. These highly sulfated GAGs also potentiated TGF- $\beta$ 1 activity in mink lung epithelial cells, while a low sulfated mucosal heparan sulfate did not. TGF- $\beta$ 1 and TGF- $\beta$ 2 are reported to have isoelectric points (pI) of approximately 9.5 and 8.5, respectively [275-276], while TGF- $\beta$ 3 has a lower pI of 6.8, suggesting that it is actually negatively charged at physiological pH and unable to electrostatically complex with sulfated GAGs [275].



Based on analysis of the TGF- $\beta$ 1 crystal structure (Figure 2.1a) and amino acid differences with TGF- $\beta$ 3, Lyon et al. predicted that TGF- $\beta$ 1 interacts with heparin via basic arginine and lysine residues at positions 25, 26, 31, and 37 and a histidine at 34 at the tip of first  $\beta$ -strand loop in TGF- $\beta$ 1, along with Arg/Lys at position 94 on the adjacent tip of another  $\beta$ -strand loop [272, 274]. In TGF- $\beta$ 3, the basic amino acid at position 26 is replaced with a neutral amino acid, suggesting that position 26 is critical for binding. While the proposed binding sites form two distinct sites at opposite poles but on the same face in the TGF- $\beta$  dimer, they could be potentially be engaged by a single heparin chain at two points approximately 60 Å (or ~7 disaccharides) apart, as well as by two separate heparin chains (Figure 2.1b). The proposed binding site is also in a similar location, at the tips of the  $\beta$ -strand loops, to where TGF- $\beta$  binds its receptors; however, little competition has been reported and in many cases heparin appears to potentiate the effect [272].



Figure 2.1. GAG binding site in TGF-β1. Crystal structure of TGF-β1 as determined by Hinck et al. [277] (b) Diagram of heparin/heparan sulfate binding sites as proposed by Lyon et al. [274]

Sulfation-dependent GAG/TGF- $\beta$ 1 interactions may play important roles in TGF- $\beta$ 1 signaling, feedback to subsequently regulate GAG sulfation, as well as disease. Merceron et al. reported that sulfated polysaccharides may potentiate TGF- $\beta$ 1 signaling for chondrogenic differentiation [278]. The oversulfated marine polysaccharide GY785 DRS possessed greater binding affinity to TGF- $\beta$ 1 than its less sulfated counterpart



GY785 DR, as determined by surface plasmon resonance (SPR). Additionally, human adipose-derived MSCs produced greater GAG and collagen when cultured with GY785 DRS in the medium, and ERK1/2 phosphorylation and MAPK signaling suggested that GY785 DRS potentiated TGF- $\beta$ 1 signaling to promote chondrogenesis and that this interaction was dependent on the sulfation of the polysaccharide. Additionally, Zanni et al. reported that TGF- $\beta$ 1 signaling in cartilage explants resulted in reduction in chondroitin-4-sulfation and an increase in nonsulfated disaccharides, suggesting that a potential feedback loop may exist between sulfate-dependent GAG/TGF- $\beta$  signaling and subsequent CS sulfation patterning via altered activity or synthesis of sulfortansferases [279]. Kim et al. also demonstrated that abnormal balance of 4- or 6-sulfation pattern and subsequent growth factor affinity may play a role in disease [280]. Decorin from the tendons of horses with equine systemic proteoglycan accumulation (ESPA) possessed enhanced 6-sulfation, as well as a decrease in 4- to 6-sulfation ratio, and that ESPA decorin exhibited diminished TGF- $\beta$ 1 binding *in vitro*. These finding may have relation to Ehlers-Danlos syndrome, a genetic disorder characterized by abnormal collagen synthesis in connect tissue, which has been linked to a mutation in galactosyl transferase I gene.

Electrostatic interactions between CS and growth factors have been exploited for delivery and controlled release. Park et al. developed a porous chitosan-CS sponge for controlled sequestration and release of PDGF-BB [262]. Release kinetics were dependent on CS content in the sponge composition, up to 40% CS, and PDGF-BB-loaded sponges were shown to promote migration and proliferation of rat osteoblasts *in vitro*, compared to chitosan-only scaffolds. Mullen et al. also designed collagen-CS materials for gradual,



sustained release of the chondrogenic growth factor IGF-1 over 14 days [264]. IGF-1 adsorption was characterized over time and as a function of loading concentration, and binding also varied with buffer ionic strength, indicating electrostatic interaction. Released IGF-1 retained bioactivity and promoted proteoglycan production of seeded human osteoarthritic chondrocytes *in vitro*, compared to unloaded constructs.

Collectively, these studies suggest that CS sequesters and regulates important interactions with growth factors, including the chondrogenic growth factor TGF- $\beta$ 1, and that the degree of sulfation may play a role in binding affinity. Further examination of the role of CS in TGF- $\beta$ 1 signaling and the cellular interactions that promote chondrogenic differentiation of MSCs may aid in the development of novel tissue engineering strategies to promote cartilage repair and regeneration.



## **CHAPTER 3**

# AGGREGATION OF BOVINE MARROW STROMAL CELLS AND ANTERIOR CRUCIATE LIGAMENT FIBROBLASTS PROMOTES AGGRECAN PRODUCTION<sup>1</sup>

## 3.1 Introduction

Several recent studies have demonstrated that adult dermal fibroblasts may possess multilineage potential and specifically the ability to produce cartilaginous extracellular matrix (ECM) in controlled culture conditions [10, 281-284]. Previous work has also investigated the ability of ECM molecules to direct cell differentiation and promote ECM production. Aggrecan is a cartilaginous ECM proteoglycan, and culture on aggrecan-coated surfaces has been shown to promote cell aggregation and production of glycosaminoglycan (GAG) and collagen II in RAB-9 adult rabbit dermal fibroblasts pretreated with insulin-like growth factor-1 (IGF-1) [10]. These results suggest that adult dermal fibroblasts maintain the ability to differentiate/transdifferentiate and produce cartilaginous ECM on aggrecan-coated surfaces, and that aggrecan-modified materials could potentially be used to promote production of chondrocytic ECM.

Production of cartilaginous ECM by a fibroblastic cell type may possess particular application for regeneration of fibrocartilaginous tissues. Fibrocartilage appears in the menisci, annulus fibrosis of intervertebral discs, and temporomandibular joint, and

<sup>&</sup>lt;sup>1</sup> Portions of this chapter were adapted from Lim JJ, Scott L, Jr., Temenoff JS. Aggregation of bovine anterior cruciate ligament fibroblasts or marrow stromal cells promotes aggrecan production. Biotechnol Bioeng. 2011;108(1):151-62.



contains a fibrous collagen I matrix along with cartilaginous ECM, including collagen II and aggrecan [27-29]. Fibrocartilaginous tissues also mediate direct insertions of tendon/ligament into bone [285-286], as fibrocartilaginous regions permit a gradual increase in stiffness from the fibrous tendon/ligament to the highly mineralized bone, thus preventing the formation of stress concentrations in the tissue and decreasing the risk of failure [287-288]. Production of chondrocytic ECM in fibroblastic cell types may possess unique application to tendon/ligament insertions, because fibroblasts are the predominant cell type in tendons and ligaments [69, 78, 289]; therefore the production of cartilaginous ECM in ligament fibroblasts may present a novel approach to regenerate fibrocartilaginous interfaces for ligament repair.

The anterior cruciate ligament (ACL) of knee is one of the most commonly injured ligaments with 250,000 ACL ruptures each year in the United States, resulting in over 100,000 ACL reconstructions annually [290]. In ACL reconstruction surgery, graft tissue is drawn across the knee to replace the ligament, and the graft is commonly held in place within bone tunnels using interference screws [287, 291]. During healing, collagenous fibers and mineralized tissue formation in the bone tunnels help anchor the graft to the bone; however, these fixation methods are unable to physiologically replicate the native fibrocartilaginous insertion points that are present at many interfaces of ligament and bone [287]. In addition, secondary surgeries for harvest of autograft tissue often results in significant donor site morbidity [78, 291], and deficiencies in fixation strength and graft positioning may lead to secondary pathologies, such as osteoarthritis [287, 292].



The limitations of surgical repair have introduced a need for tissue engineering strategies for ligament regeneration. A potential approach would be to develop a tissue-engineered bone-ligament-bone graft that possesses a fibrous ligament midsubstance with the mechanical strength to restore function to the injured joint, as well as fibrocartilaginous insertions and osseous tissue for improved fixation of the graft to bone. In order to function similarly to native ligaments, the fibrous ligament body would possess structurally organized collagen type I fibers aligned in parallel to maximize tensile strength, along with sparsely distributed fibroblasts to maintain the ECM [293-294]. The fibrocartilaginous insertions, on the other hand, would contain chondrocyte-like cells within a matrix of collagen I, collagen II, aggrecan, and some collagen X [287-288, 295-296].

To engineer such tissue interfaces, methods to produce zonally-varied ECM similar to that found in the ligament-bone insertion point are needed. In prior work, fibrous poly(lactic-co-glycolic acid) (PLGA) has been combined with bioactive glass in a composite scaffold to support both ligament and bone development [297-298], and these scaffolds have been used as a multiphasic co-culture system for ligament fibroblasts, chondrocytes, and osteoblasts [298-299]. Additionally, gradients of retroviral transfection with RUNX2/CBFA1 osteogenic transcription factor in fibroblasts have been utilized to produce graded distributions of mineral deposition for transitional tissue interfaces [300]. While the experiments detailed above have explored the culture of multiple cell types or biomaterial-based induction of phenotypic changes to produce the ECM gradient required for these applications, few studies have examined the possibility of altering pre-culture techniques to encourage one cell type to produce the range of phenotypes needed for this



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complex tissue structure. Therefore, the focus of this set of experiments was to examine the potential for formation of cartilaginous ECM through altering cell pre-culture parameters for ACL fibroblasts. This was then compared with the response under the same culture conditions for bone marrow stromal cells (BMSCs), which are known to differentiate into various mesenchymal cell types, including osteoblasts, chondrocytes, adipocytes, myoblasts, and tendon/ligament fibroblasts [77-79].

Specifically, the purpose of this study was to investigate the role of aggrecan in chondrogenic differentiation of bovine ACL fibroblasts and BMSCs. We hypothesized that cells would aggregate on aggrecan-coated surfaces, and that aggrecan-coated surfaces would promote the expression and production of cartilaginous ECM by both ACL fibroblasts and BMSCs. Because cells were found to aggregate on aggrecan-coated surfaces, another aggrecan-free surface treatment that induced aggregation was used as a control to determine whether the resulting phenotypic differences were aggrecandependent. In particular, cell morphology, gene expression, and ECM production were examined over 14 days for bovine ACL fibroblasts and BMSCs cultured on uncoated tissue culture-treated polystyrene (TCPS), aggrecan-coated TCPS, and nonadhesive culture plates in order to determine how the culture surface affected phenotypic responses for both cell types.

#### 3.2 Materials and Methods

#### **3.2.1 Bovine ACL Fibroblast and BMSC Harvest**

Bovine ACL fibroblasts were harvested according to a previously described protocol [174]. The femur and tibia of an immature calf (Research 87, Marlborough, MA)



were isolated, and the ACL was sterilely removed and cut into ~1 mm<sup>3</sup> pieces. Ligament tissue was then digested in 0.4% (w/v) collagenase type II (Invitrogen, Carlsbad, CA) (Mediatech. solution in high-glucose DMEM Manassas. VA). 1% penicillin/streptomycin/neomycin (Invitrogen), 1% kanamycin (Mediatech), 0.1% gentamicin (Mediatech), and 0.1% Fungizone (Invitrogen) for 24 hr. The digested cell solution was filtered through a cell strainer with 80 µm mesh (Small Parts, Miramar, FL), collagenase was removed by centrifugation, and cells were resuspended in "fibroblast medium" containing high-glucose DMEM, 10% fetal bovine serum (FBS; Thermo Scientific HyClone, Waltham, MA), 1% antibiotic/antimycotic (Mediatech), 1% nonessential amino acids (Mediatech), 1% HEPES buffer (Mediatech), and 50 µg/ml ascorbate (Sigma-Aldrich, St. Louis, MO) and cryopreserved until use.

Bovine bone marrow stromal cells (BMSCs) were harvested according to a previously described protocol [213]. The femur and tibia of an immature calf were isolated, and marrow was placed into phosphate-buffered saline (PBS) with 1% antibiotic/antimycotic. The resulting mixture was physically disrupted by repeatedly pipetting through 50 and 10 mL pipettes, followed by 16, 18, and 20 gauge syringe needles. After centrifugation, the fatty layer was aspirated off, and the cell solution was plated for 30 minutes in "BMSC medium" containing low-glucose DMEM (Mediatech), 10% FBS, 1% antibiotic/antimycotic, and 1 ng/mL basic fibroblast growth factor (bFGF, Peprotech, Rocky Hill, NJ) to allow rapidly adhering cells to attach to the tissue culture plastic. The remaining cells in solution were then plated in tissue culture flasks. After 24 hours, nonadherent cells were aspirated off, culture medium was replaced, and BMSCs



were allowed to expand until confluency. Confluent cells were trypsinized in 0.05% trypsin/EDTA (Mediatech) and cryopreserved until use.

For long-term storage, ACL fibroblasts and BMSCs were placed in medium supplemented with 20% FBS and 10% dimethyl sulfoxide (DMSO; Sigma-Aldrich) and stored in liquid nitrogen until further use. Before use, cells were replated at 1.0x10<sup>6</sup> cells/flask for at least 3 days in T-150 flasks to eliminate transitory effects from the thawing process. To examine the effects of interdonor variability, ACL fibroblasts and BMSCs were each isolated from 3 different animals (donors) for this study, including 1 donor from which both ACL fibroblasts and BMSCs were isolate from a single leg (donor #3).

## 3.2.2 Cell Culture on Aggrecan, Nonadhesive, and Control Surfaces

Similar to [10], wells of a tissue culture-treated 24-well plate were coated with 5  $\mu$ g of bovine aggrecan (Sigma-Aldrich) by suspending aggrecan in PBS at a concentration of 50  $\mu$ g/mL and allowing the PBS to evaporate overnight in a sterile environment at room temperature. After washing wells with PBS to remove residual unadsorbed aggrecan, wells were stained for 30 minutes with 16  $\mu$ g/mL dimethylmethylene blue (DMMB; pH 3.0; Sigma-Aldrich) for sulfated GAGs to determine the distribution of aggrecan on the surfaces. For other samples, bovine ACL fibroblasts (P1) and BMSCs (P2) were plated at 8.8x10<sup>4</sup> cells/well on aggrecan-coated surfaces and TCPS control surfaces in FBS-supplemented media. ACL fibroblasts were cultured in fibroblast medium as described above and including 1% nonessential amino acids, 1% HEPES buffer, and 50  $\mu$ g/mL ascorbate. BMSCs were cultured in BMSC medium as described above and including 1 g/mL bFGF. For all experiments in FBS-



supplemented media, FBS from a single lot was used to prevent variability between serum lots. These two cell types were also cultured on nonadhesive surfaces that resist cell attachment (Ultra-Low Attachment surfaces, Corning, Lowell, MA). To examine the nature of the cellular interaction with aggrecan, ACL fibroblasts and BMSCs were cultured in FBS-free media on aggrecan-coated and control surfaces.

Additionally, for specific experiments, aggrecan was added to the media of ACL fibroblasts and BMSCs cultured on control surfaces to examine the effect of aggrecan when suspended in the culture medium. 5  $\mu$ g of aggrecan was added to the medium either at the time of seeding or 24 hours after seeding, instead of pre-treating the wells prior to seeding. For all experiments, culture medium was replaced every 2-3 days, and cell morphology was observed with a phase contrast microscope using a 10X objective (Nikon Eclipse TE2000-U, Tokyo, Japan).

## **3.2.3 Reverse Transcription Polymerase Chain Reaction**

Gene expression of ACL fibroblasts and BMSCs on both aggrecan-coated surfaces and nonadhesive surfaces was analyzed after 3, 7, and 14 days by reverse transcription polymerase chain reaction (RT-PCR) for collagen I (fibroblastic marker), collagen II (chondrocytic marker), aggrecan (chondrocytic marker), and peroxisome proliferator-activated receptor  $\gamma 2$  (PPAR- $\gamma 2$ ; adipocytic marker). ACL fibroblasts and BMSCs isolated from 3 different bovine donors, including 1 donor from which both ACL fibroblasts and BMSCs were isolated from a single leg (donor #3), were analyzed for interdonor variability. On aggrecan-coated surfaces, aggregates were separated from nonaggregating cells for analysis using a 1000 µL pipette tip to isolate the samples.



Nonadherent aggregates on nonadhesive surfaces were simply resuspended in culture medium, and TCPS monolayers were trypsinized and resuspended.

RNA was extracted from cell samples using a QIAshredder tissue homogenizer and RNeasy kit with DNase I digestion (Qiagen, Hilden, Germany). Reverse transcription was performed using SuperScript III Reverse Transcriptase (Invitrogen) with Oligo(dT)<sub>15</sub> primers (Promega, Madison, WI) and nucleotides (Promega). Custom primers (Invitrogen) specific to bovine mRNA for the target sequences are shown in Table 3.1, and quantitative PCR amplification was performed on a StepOnePlus System (Applied Biosystems, Carlsbad, CA) with SYBR Green master mix (Applied Biosystems). Fold regulation over control TCPS surfaces was calculated using the  $\Delta\Delta C_{T}$  method with glyceraldehyde-3-phosphate dehydrogenase (GAPDH) as an endogenous control [301].

Primer Sequences (5'-3') GenBank Target AAGAACCCAGCTCGCACATG Collagen I (a2) AB008683 GGTTAGGGTCAATCCAGTAGTAACCA GCATTGCCTACCTGGACGAA Collagen II (a1) X02420 CGTTGGAGCCCTGGATGA CCTCAGGGTTTCCTGACATTA Aggrecan NM\_173981 TAAGCTCAGTCACGCCAGATA CGCACTGGAATTAGATGACAGC PPAR-v2 NM\_181024 CACAATCTGTCTGAGGTCTGTC CCTTCATTGACCTTCACTACATGGTCTA GAPDH NM\_001034034 TGGAAGATGGTGATGGCCTTTCCATTG

Table 3.1. Bovine primer sequences for quantitative polymerase chain reaction

#### **3.2.4 Dimethylmethylene Blue (DMMB) and PicoGreen Assay**

ACL fibroblast and BMSC aggregates were also analyzed by DMMB assay for sulfated GAG production and by PicoGreen assay for DNA content. ACL fibroblast (donor #1) and BMSC (donor #3) aggregates cultured on aggrecan-coated and nonadhesive surfaces were isolated as described above and digested in 1 mg/mL tris(hydroxymethyl)aminomethane proteinase Κ (Sigma-Aldrich) in



(Tris)/ethylenediaminetetraacetic acid (EDTA) digestion buffer (pH 6.5) with 185 µg/mL iodacetamide (Sigma-Aldrich) and 10 µg/ml pepstatin A (Sigma-Aldrich) for 16 hours at 56°C, and analyzed by DMMB assay, measuring absorbance at 520 nm (SpectraMax M2e; Molecular Devices, Sunnyvale, CA), compared to acellular blanks and a chondroitin sulfate standard curve [302]. Cells were then lysed through a series of freeze/thaw cycles and sonication, and PicoGreen assay (Invitrogen) was used to evaluate the total DNA content in each sample, according to established protocols [303]. Fluorescence was read at excitation 485 nm, emission 525 nm (SpectraMax M2e), and DNA content was determined using a standard curve of DNA and translated to cell number, compared to known numbers of bovine ACL fibroblasts and BMSCs. GAG content and cell number was normalized to the approximate number of aggregates per sample, and GAG content was also normalized to cell number.

## **3.2.5 Histological Staining**

For this study, staining was only undertaken for proteins that were upregulated according to gene expression results (in this case, aggrecan only). Nonadherent ACL fibroblast and BMSC aggregates, from a single bovine donor (donor #3), cultured on nonadhesive surfaces were rinsed in PBS and fixed for 60 minutes in 10% buffered formalin after 3, 7, and 14 days. Fixed aggregates were encapsulated in Histogel (Richard-Allan, Waltham, MA), embedded in paraffin wax, and sectioned at 5  $\mu$ m thickness. Sections were either stained with hematoxylin and eosin (H&E; Leica Autostainer XL, Wetzlar, Germany) or stained by immunohistochemistry (IHC) for aggrecan. For immunostaining, antigen retrieval was achieved with 20 µg/mL proteinase K for 10 minutes. Samples were deglycosylated with 10 µL of 0.75 U/mL chondroitinase



ABC (Sigma-Aldrich) for 1.5 hours to uncover the aggrecan core protein epitope. After blocking nonspecific binding with a solution of 2% goat serum, 1% bovine serum albumin (BSA), 0.1% gelatin, and 0.1% Triton-X in PBS and blocking endogenous peroxidases with 0.3%  $H_2O_2$  in methanol, monoclonal mouse antibodies to bovine aggrecan (Abcam, Cambridge, UK) were allowed to bind overnight, and horseradish peroxidase-conjugated (HRP) secondary goat anti-mouse antibodies (Abcam) were allowed to incubate for 30 minutes, before diaminobenzidine (DAB) chromogen (Abcam) was used to develop the brown color that indicated positive immunostaining for aggrecan. Histological sections were imaged using a brightfield microscope at 10X and 40X magnifications (Nikon Eclipse E600).

#### **3.2.6 Statistical Analysis**

All values were reported as mean  $\pm$  standard deviation. Statistical significance of quantitative results was determined using a two-way (quantitative PCR, DMMB) or three-way (PicoGreen) analysis of variance (ANOVA) with Tukey's multiple comparison test ( $p \le 0.05$ ). For quantitative PCR analysis, the two factors were surface and time. For DMMB analysis, the two factors were cell type and time. For PicoGreen analysis, the three factors were surface, cell type, and time. Statistical analysis was carried out using Minitab (v15.1, State College, PA).


#### 3.3 Results

# **3.3.1** Aggregation of ACL fibroblasts and BMSCs on aggrecan-coated and nonadhesive surfaces

Aggrecan-coated surfaces induced formation of dense cellular aggregates along the periphery of the wells within 24 hours of cell seeding (Figure 3.1a,e), compared to TCPS surfaces (Figure 3.1d,h). DMMB staining of aggrecan-coated surfaces verified that the majority of aggrecan adsorbed in a ring along the periphery of each well, aligning closely with where cell aggregation occurred. These aggregates were tethered to the surface, but some aggregates were still able to move when the medium surrounding them was agitated. Aggregation occurred in both ACL fibroblasts (Figure 3.1a) and BMSCs (Figure 3.1e). Aggregates were consistently less than 100  $\mu$ m in diameter with most aggregates between 30 and 80  $\mu$ m. Aggregate formation on aggrecan-coated surfaces was not dependent on the presence of serum proteins, with ACL fibroblast and BMSC aggregates forming in serum-free media (Figure 3.1b,f). When aggrecan was suspended in the culture medium, rather than pre-adhered on culture surfaces, the cells appeared morphologically similar to cells on control surfaces, indicating that aggrecan must be adsorbed to the surface prior to cell seeding to induce aggregation. Large cellular aggregates also formed on nonadhesive surfaces in the absence of aggrecan (Figure 3.1c,g), and unlike aggrecan-coated surfaces, these aggregates were not adhered to the culture surface. Aggregate size for both ACL fibroblasts and BMSCs on nonadhesive surfaces was highly variable with aggregates ranging from less than 100 up to 400 µm in diameter.



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Figure 3.1. ACL fibroblasts (a-d) and BMSCs (e-h) on aggrecan-coated, nonadhesive, and control tissue culture-treated polystyrene surfaces in serumsupplemented and serum-free media after 3 days. Within 24 hours, aggregation (arrows) was observed on aggrecan-coated (a,e) and nonadhesive (c,g) surfaces, and aggregate morphology was maintained over 14 days. Aggregate formation on aggrecan-coated surfaces was not dependent on the presence of serum proteins (b,f). Scale bars in all images are 100 μm.

PicoGreen DNA assay was used to estimate the number of cells in ACL fibroblast (donor #1) and BMSC (donor #3) aggregates cultured on aggrecan-coated and nonadhesive surfaces. On aggrecan-coated surfaces, ACL fibroblast aggregates were composed of approximately 150±34 cells, and BMSC aggregates contained 51±17 cells on day 3. Aggregates cultured on nonadhesive surfaces, however, demonstrated higher cell number than those on aggrecan-coated surfaces, regardless of cell type; ACL fibroblast aggregates possessed approximately 3,920±480 cells, and BMSC aggregates contained 600±95 cells on nonadhesive surfaces on day 3. From statistical analysis, ACL fibroblast aggregates had significantly higher cellularity than BMSC aggregates when cultured on nonadhesive surfaces, although aggregate cell number was not significantly



different between cell types on aggrecan-coated surfaces. PicoGreen results further indicated that aggregate cell number did not significantly change over time on either surface.

# **3.3.2 Upregulation of aggrecan gene expression in ACL fibroblast and BMSC aggregates on aggrecan-coated and nonadhesive surfaces**

Overall, ACL fibroblast and BMSC aggregates on aggrecan-coated and nonadhesive surfaces showed significantly upregulated aggrecan gene expression. Though some interdonor variability was evident, in all donors (donors #1-3), ACL fibroblast aggregates on aggrecan-coated surfaces demonstrated significantly upregulated aggrecan expression at all time points, compared to cells on TCPS controls (Figure 3.2ac). On nonadhesive surfaces, ACL fibroblasts experienced significant upregulation of aggrecan expression on day 14 in all donors (Figure 3.2d-f), and donor #1 also showed upregulated aggrecan expression on days 3 and 7, compared to TCPS controls (Figure 3.2d). In addition, collagen I expression was significantly downregulated in ACL fibroblasts on nonadhesive surfaces over all time points in all 3 donors. Collagen II and PPAR- $\gamma$ 2 were generally downregulated over most time points on nonadhesive surfaces in cells from donors #1 and #2. Minimal change in gene expression of collagen I, collagen II, and PPAR- $\gamma$ 2 was observed on aggrecan-coated surfaces with ACL fibroblasts.



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Figure 3.2. Gene expression of ACL fibroblasts from 3 different bovine donors on aggrecan-coated (a-c) and nonadhesive (d-f) surfaces over 14 days, relative to cells on untreated control surfaces. (Aggrecan expression is depicted on separate axes using a different vertical scale.) Some interdonor variability was evident; however, ACL aggregates significantly upregulated aggrecan gene expression on aggrecan-coated and nonadhesive surfaces (n=3), particularly by day 14. \* indicates significantly different from control samples (*p*≤0.05).

Similar to ACL fibroblasts, BMSC aggregates on aggrecan-coated and nonadhesive surfaces also demonstrated an increase in aggrecan gene expression. Though some interdonor variability was evident, BMSC aggregates on aggrecan-coated surfaces showed significantly upregulated aggrecan expression on day 3, compared to TCPS controls, in all bovine donors (donors #3-5; Figure 3.3a-c); however, expression diminished to control levels on days 7 and 14. On nonadhesive surfaces, BMSCs from all 3 donors experienced significant upregulation of aggrecan expression over TCPS controls



on all time points (Figure 3.3d-f). Collagen II was significantly downregulated on both surfaces over all times in cells from donor #5 and on nonadhesive surfaces over most time points with donor #3. Collagen I and PPAR- $\gamma$ 2 were significantly downregulated over most time points on nonadhesive surfaces in cells from donors #3 and #4. Minimal change in gene expression of collagen I and PPAR- $\gamma$ 2 was observed on aggrecan-coated surfaces.

#### 3.3.3 Production of sulfated GAG in ACL fibroblast and BMSC aggregates

ACL fibroblast (donor #1) and BMSC (donor #3) aggregates on nonadhesive surfaces were analyzed by DMMB assay for production of sulfated GAG. ACL fibroblast aggregates produced 69±3 ng GAG per aggregate, while BMSC aggregates produced 12±2 ng GAG per aggregate, suggesting that ACL aggregates contained significantly more GAG than BMSC aggregates; however, GAG production was not statistically significant across cell types when normalized to cell number. ACL aggregates contained 18±1 pg GAG per cell, while BMSC aggregates contained 20±4 pg GAG per cell after day 3 (Table 3.2). GAG content in ACL and BMSC aggregates on nonadhesive surfaces did not significantly change over time.





Figure 3.3. Gene expression of BMSCs from 3 different bovine donors on aggrecancoated (a-c) and nonadhesive (d-f) surfaces over 14 days, relative to cells on untreated control surfaces. (Aggrecan expression is depicted on separate axes using a different vertical scale.) Some interdonor variability was evident; however, BMSC aggregates significantly upregulated aggrecan gene expression on aggrecan-coated and nonadhesive surfaces on day 3 (n=3). \* indicates significantly different from control samples (*p*≤0.05).

| surfaces as actor million by Division assay, normalized by con manifor (n=c) |                           |              |               |  |  |
|--|---------------------------|--------------|---------------|--|--|
| Cell Type  | Mass of GAG (pg per cell) |              |               |  |  |
|  | Day 3                     | Day 7        | Day 14        |  |  |
| ACL Fibroblasts  | 18±1 pg/cell              | 21±6 pg/cell | 21±4 pg/cell  |  |  |
| BMSCs  | 20±4 pg/cell              | 24±2 pg/cell | 16±14 pg/cell |  |  |

Table 3.2. Sulfated GAG content within cell aggregates cultured on nonadhesive surfaces as determined by DMMB assay, normalized by cell number (n=3)

#### 3.3.4 Production of aggrecan in ACL fibroblast and BMSC aggregates

Immunostaining of ACL fibroblast and BMSC (both donor #3) aggregates cultured on nonadhesive surfaces verified the presence of aggrecan after 3, 7, and 14 days



(Figure 3.4b-d,f-h), compared to negative control samples stained without primary antibody for aggrecan (Figure 3.4a,e). Also, BMSCs formed noticeably better defined aggregates than ACL fibroblasts. While ACL aggregates showed some sectioning artifact, especially at early time points, in which cells dissociated from the aggregate body, BMSC aggregates appeared to possess a thin, smooth layer surrounding each aggregate and remained intact during sectioning. In immunostaining and hematoxylin and eosin staining, a distinct change in aggregate morphology was observed in BMSC aggregates from 3 to 14 days (Figures 3.4f-h, 3.5d-f). Starting at day 7 and increasing by day 14, small pocket-like spaces were present in BMSC aggregates, however, did not observe this pocket-like ECM organization, and morphology did not appear to greatly change over the course of 14 days (Figure 3.4b-d, 3.5a-c).





Figure 3.4. Presence of aggrecan within ACL fibroblast (a-d) and BMSC (e-h) aggregates as indicated by immunohistochemistry staining. ACL and BMSC aggregates stained strongly for aggrecan (b-d, f-h), compared to negative control samples stained without primary antibody for aggrecan (a,e). A noticeable change in ECM morphology is evident over 14 days in BMSC aggregates. Boxed regions are magnified in insets located in the lower left corner of each image. Scales bars in images are 100 μm, and insets are 25 μm.





Figure 3.5. Hematoxylin and eosin (H&E) staining of ACL fibroblast (a-c) and BMSC (d-f) aggregates. A noticeable change in ECM morphology is evident over 14 days in BMSC aggregates. Boxed regions are magnified in insets located in the lower left corner of each image. Scales bars in images are 100 μm, and insets are 25 μm.

#### 3.4 Discussion

The results from this study demonstrated that aggrecan-coated surfaces can be used to alter the morphology of both bovine ACL fibroblasts and BMSCs. Modification of surfaces with aggrecan molecules promoted the formation of ACL fibroblast clusters and BMSC aggregates (Figure 3.1). DMMB staining verified that the localization of aggrecan on tissue culture surfaces aligned closely with areas of cell aggregation.



Aggregation required that aggrecan molecules be adsorbed to the surface prior to cell seeding, and the presence of aggrecan suspended in the culture medium did not induce aggregate formation, suggesting that the interaction between the aggrecan and the surface prior to cell seeding reduced cell adhesion to the culture dish and thus encouraged cell clustering. However, aggregation on aggrecan-coated surfaces was not dependent on the presence of serum proteins, with aggregates forming in FBS-free media (Figure 3.1). After formation, ACL fibroblast and BMSC aggregates on aggrecan-coated surfaces contained significantly fewer cells than aggregates on nonadhesive surfaces, and aggregate cellularity did not change with time.

Aggregation of bovine ACL fibroblasts and BMSCs, even in the absence of aggrecan (as shown on nonadhesive surfaces), promoted gene expression and production of aggrecan, a key cartilaginous ECM proteoglycan (Figures 3.2-4, Table 3.2). In general, there was variability between the 3 bovine donors tested for gene expression, suggesting intrinsic differences in how these cells respond to culture conditions; however, the general trend of aggrecan upregulation was maintained regardless of cell source, which indicated that aggregation is a robust method to promote cartilaginous gene expression in these cells. Results from the DMMB assay and immunostaining verified that ACL fibroblast and BMSC aggregates cultured on nonadhesive surfaces also produced detectable amounts of sulfated GAG/aggrecan. However, direct comparison of aggrecan production between culture surfaces was not possible as cells on aggrecan-coated surfaces could not be isolated without releasing the original aggrecan coating as well, thus potentially contaminating the samples.



These data suggested that cells could potentially be pre-cultured as aggregates in order to promote the production of cartilaginous ECM within a cell population, prior to their incorporation into a portion (insertion points) of a 3D scaffold for ligament tissue engineering applications. Alternatively, biomaterial scaffolds may be designed with means to induce differing degrees of cellular aggregation along their length [174], thereby predisposing a certain population of cells to produce larger amounts of aggrecan in areas near the insertion point. RT-PCR results indicated that cells responded similarly in aggregates, regardless of the cell culture surface employed (Figures 3.2-3.3), suggesting aggregation itself was a major stimulus for the changes in gene expression observed. This leaves open the possibility of using a wide variety of pre-culture methods, including those tissue culture surfaces presented here, to encourage cells to form cartilaginous ECM for insertion point regeneration.

Similar to culture of these cell aggregates on nonadhesive surfaces, pellet culture and micromass culture are established methods to prevent the dedifferentiation of chondrocytes or to promote chondrogenic differentiation of BMSCs [6, 98, 132, 304-307]. High-density culture may mimic the cell environment of mesenchymal condensation that occurs during cartilage development prior to chondrogenesis, including cell-cell and cell-matrix interactions [24, 308]. However, to our knowledge, this is the first time aggregate culture has been investigated for chondrogenic differentiation of tendon/ligament fibroblasts. High-density culture of human tenocytes, including pellet formation by centrifugation or clustering in response to ultra-confluent monolayer culture, has been examined for maintenance of tenocyte phenotype [309-311], and one group reported accumulation of proteoglycan over 14 days with no change in collagen II

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[309]; however, further implications for chondrogenic differentiation were not explored. In contrast, this study induced aggregate formation via culture on nonadherent 2D surfaces, both with or without aggrecan treatment, to investigate expression of cartilaginous ECM by ACL fibroblasts. Micromass culture has also been applied with lactic acid treatment to promote production of cartilaginous ECM by human dermal fibroblasts [281]. These results suggest that ligament fibroblasts retain the plasticity to alter their gene expression and possibly differentiate toward a chondrogenic phenotype under the appropriate conditions.

While aggrecan was significantly upregulated in clustered cells, collagen type II, another chondrocytic ECM marker, was either significantly downregulated or remained unchanged in both ACL fibroblast and BMSC aggregates on both surfaces. This decoupled expression of aggrecan and collagen type II suggests that while aggrecan expression is upregulated in cell aggregates, the cells are not completely committed to a chondrocytic phenotype. For use as part of a tissue-engineered ligament replacement, cartilaginous ECM (aggrecan and collagen II) should be upregulated, while fibrous ECM (collagen I) expression should be either upregulated or maintained within ACL fibroblasts and BMSCs; however, this system may be useful to investigate the basic interactions of cells with aggrecan in the absence of a collagen II matrix. Also, it is important to note that gene expression of other markers such as PPAR- $\gamma$ 2 was either downregulated or unchanged over most time points, indicating that ACL fibroblasts and BMSCs are not differentiating down an alternate (adipogenic) lineage. Further work is required to determine the optimal culture conditions that would simultaneously upregulate collagen I, collagen II, and aggrecan expression for production of



fibrochondrogenic ECM. A combination of aggregate-inducing culture substrates with soluble factors or mechanical loading (not explored in this study) may be necessary to achieve an optimal cell phenotype for use in the insertion sites of future tissue-engineered ACL replacements.

The ability of BMSCs and ACL fibroblasts to alter their gene expression and ECM production may be dependent on the presence of a subpopulation of multipotent progenitor cells in both cases. BMSCs and ACL fibroblasts in this study are heterogeneous populations of cells from the bone marrow and ACL, respectively. BMSCs likely include a subpopulation of multipotent mesenchymal stem cells [77], and while cells from the ACL are mostly ligament fibroblasts, they may include a small subpopulation of ligament progenitor cells, similar to progenitor/stem cells isolated from tendon and periodontal ligament [312-313]. Overall, BMSCs displayed greater and more consistent upregulation of aggrecan expression than ACL fibroblasts (Figures 3.2-3.3), suggesting that, even though ACL fibroblasts are capable of altering their gene expression profiles, BMSCs are more consistently responsive to cell aggregation and may possess are larger population of progenitor cells than ACL fibroblasts. Ligament fibroblasts, however, have been extensively explored as a cell source for ligament tissue engineering [69, 314-315], as an injured ACL could be surgically removed and the harvested fibroblasts used as an autologous cell source to populate a ligament graft [316]. The ability of ACL fibroblasts to alter their gene expression and produce cartilaginous ECM suggests that ACL fibroblasts may also be a promising cell type in the development of a tissue-engineered bone-ligament-bone graft including fibrocartilaginous insertions.



From histological sections, distinct differences were observed between the morphology of ACL fibroblast and BMSC aggregates cultured on nonadhesive surfaces (Figure 3.4-3.5). Most noticeably, in this study, small pocket-like spaces were observed in day 14 BMSC aggregates with ECM appearing as a circular, interconnected network. The matrix organization in day 14 BMSC aggregates may be an effect of nutrient transport into these large cell aggregates. It should be noted that this ECM network has not been reported by similar studies examining micromass culture of bovine BMSCs over similar time scales [132, 307]. However, in these other studies, aggregates were formed by centrifugation resulting in much larger cell pellets, which could account for differences in observed pellet morphology. Additionally, cells in this study were cultured with 10% FBS, while other studies replaced serum with insulin-transferrin-selenium (ITS) supplement with and without transforming growth factor- $\beta$ 1 or - $\beta$ 3 (TGF- $\beta$ ). Further work is needed to fully understand the timing and events involved in differentiation and matrix production that may lead to this unique ECM structure within these aggregates.

Together the results of these studies present aggregate culture of ACL fibroblasts and BMSCs as a potential pre-culture technique to promote the regeneration of cartilaginous and fibrocartilaginous tissues.

#### **3.5** Conclusions

These studies have shown that aggrecan-coated surfaces induce the aggregation of bovine ACL fibroblasts and BMSCs, and cell clusters (on aggrecan-coated surfaces and in the absence of aggrecan) demonstrated upregulated aggrecan gene expression,



regardless of cell type. Immunostaining also confirmed the presence of aggrecan in cell aggregates over 14 days. These findings suggest that both cell types could potentially be pre-cultured as aggregates to promote production of cartilaginous ECM, specifically aggrecan proteoglycan, prior to seeding on scaffolds, for orthopaedic tissue engineering applications requiring the presence of this molecule. This is particularly attractive as it suggests that altering pre-culture conditions like degree of clustering could produce a range of phenotypes from a single cell source, such as the ACL fibroblast. As such, these findings represent a first step which may inform future approaches to producing tissue-engineered alternatives for current ACL grafting procedures through regeneration of physiological fibrocartilaginous interfaces.



#### **CHAPTER 4**

### DEVELOPMENT OF NANO- AND MICRO-SCALE CHONDROITIN SULFATE PARTICLES FOR CONTROLLED GROWTH FACTOR DELIVERY<sup>2</sup>

#### 4.1 Introduction

Chondroitin sulfate (CS) is a sulfated glycosaminoglycan (GAG) that is covalently bound to a variety of protein cores to form proteoglycans that are found throughout the human body [317]. During development, one of the key roles of GAGs is to establish morphogen gradients that pattern tissue morphogenesis by sequestering secreted growth factors at the cell membrane [318]. Negatively charged GAGs, including heparin, heparan sulfate, and CS, are capable of electrostatically interacting with positively charged growth factors, including basic fibroblast growth factor (bFGF), insulin-like growth factor (IGF), vascular endothelial growth factor (VEGF), plateletderived growth factor (PDGF), and transforming growth factor- $\beta$  (TGF- $\beta$ ), to stabilize and prevent degradation of the growth factors in solution [206, 262, 319], indicating that CS biomaterials may be a promising vehicle for delivery of cationic growth factors, particularly to direct differentiation of stem cells.

Size scale plays an important role in the release kinetics of a given delivery vehicle, including the diffusion rate of molecules out of the material and the degradation

<sup>&</sup>lt;sup>2</sup> Portions of this chapter were adapted from Lim JJ, Hammoudi TM, Bratt-Leal AM, Hamilton SK, Kepple KL, Bloodworth NC, et al. Development of nano- and microscale chondroitin sulfate particles for controlled growth factor delivery. Acta Biomater. 2011;7(3):986-95.



properties of the material [320]; therefore, control of size enables one to tailor the release profile for a variety of applications. Nanospheres and microspheres possess a high surface area-to-volume ratio, accelerating the diffusion of molecules from the particles and potentially enhancing the hydrolytic or enzymatic degradation of the carrier and further accelerating release [320-321]. The small size of nano- and microspheres, relative to cells and tissues, make them especially advantageous carriers for delivery and sustained release within dense tissues, high-density cell pellets, or tissue engineering scaffolds with numerous barriers to diffusion [322], because these small particle carriers can be incorporated or locally injected within a tissue for controlled release of specific molecules to achieve a localized or more homogenous cell response [321].

Additionally, nano-sized particles (< 500 nm diameter) are capable of being internalized by cells, permitting efficient transport across the cell membrane [321]. Generally, smaller particles are endocytosed more effectively than larger particles, though an optimal size range commonly exists around 100-200 nm [323]. Larger microscale delivery vehicles, however, may be advantageous for preventing cell endocytosis if extracellular release is desired, and their larger size scales also possess potential for greater molecule loading and prolonged release due to their larger volumes.

In this study, we explored the means to fabricate both micron-scale particles and nano-scale micelles from CS. Micelles are nanoscale, self-assembling particles composed of amphiphilic molecules. In an aqueous environment, micelles self-assemble into sphere-like structures with a hydrophilic shell surrounding a hydrophobic core, making them largely stable in the aqueous environment of the body [324]. Due to its negative charge density and hydrophilic nature, CS can be utilized as a hydrophilic segment of a



polymeric micelle. CS has been previously reported as the hydrophilic component with poly(L-lactic acid) (PLLA) and poly(lactic-co-glycolic acid) (PLGA) copolymers to form amphiphilic microspheres for use as drug and protein delivery carriers [324-326]. With its nanoscale size and high negative charge density, CS-derived micelles have promising potential for many drug and protein delivery applications.

The long term goal of this research was to examine CS interactions with stem cells, particularly as a means to control growth factor release in order to direct differentiation. Therefore, the first set of experiments in this study were designed to develop facile means to control the size of CS-based particles over a broad range of size scales (at least two orders of magnitude) and characterize the resulting materials. Subsequently, the ability of these CS-based biomaterials to interact electrostatically with positively charged growth factors, as well as the cytocompatibility of these materials with both embryonic and mesenchymal stem cells was demonstrated, confirming the potential for these materials as naturally-derived carriers for growth factor delivery for a variety of stem cell-based tissue engineering and regenerative medicine applications.

#### 4.2 Materials and Methods

#### 4.2.1 Modification of Chondroitin Sulfate

#### 4.2.1.1 Chondroitin Sulfate Modification for Micelle and Hydrogel Fabrication

To examine the effects of reaction stoichiometry on the resulting modification of CS, N-(3-dimethylaminopropyl)-N'-ethylcarbodiimide (EDC) chemistry, which forms an amide bond between a primary amine and a carboxylic acid, was used to synthesize chondroitin sulfate methacrylamide (CSMAm; Figure 4.1a). N-(3-



aminopropyl)methacrylamide (APMAm) was conjugated to chondroitin sulfate A, using EDC chemistry adapted from [327]. 2.5 mM CS (~48.7 kDa; Sigma-Aldrich, St. Louis, MO) [328] based on CS disaccharide molecular weight was reacted with APMAm (Polysciences, Warrington, PA), EDC (Sigma-Aldrich), and N-hydroxysulfosuccinimide (sulfo-NHS; Sigma-Aldrich) at various molar ratios depicted in Table 4.1. Since CS possesses 1 carboxyl moiety per disaccharide, this represented the molar ratio of reagent to each reactive carboxyl group. The reaction mixture was incubated at room temperature and pH 5.0 for 2 hours with stirring.

Sulfo-NHS stabilizes the EDC reaction and was used to control the degree of substitution of the reaction; however, to reduce the degree of modification for micelle formation, the reaction with the highest molar ratio of 4:1 was repeated in the absence of sulfo-NHS. Initially a 2:1 ratio of APMAm and EDC to CS was used and allowed to react for 2 hours, and then a second round of APMAm and EDC was added to the solution for a total 4:1 molar ratio to CS. The reaction without sulfo-NHS was allowed to proceed for 2 more hours at pH 5.0. All reaction mixtures were dialyzed in 1,000 Da molecular weight cutoff (MWCO) dialysis membrane (Spectrum Laboratories, Rancho Dominguez, CA) against distilled water (dH<sub>2</sub>O) for 3 days, and the product was then lyophilized (Labconco FreeZone 4.5, Kansas City, MO) for 4 days and stored at 4°C until use.

#### 4.2.1.2 Chondroitin Sulfate Modification for Microsphere Fabrication

For microsphere formation, CS was reacted with methacrylic anhydride (Sigma-Aldrich) to conjugate methacrylate groups to the existing hydroxyl groups in CS (Figure 4.1b). Methacrylation of CS was carried out per established protocols [328]. After complete dissolution, 60 mL methacrylic anhydride was added dropwise into 60 mL of a



25% w/v chondroitin sulfate A solution (20.6:1 molar ratio of methacrylic anhydride to reactive hydroxyl) under basic conditions. The reaction solution was stirred at 60°C for 24 hours, then precipitated and washed in cold methanol:isopropanol (10:1). The resulting chondroitin sulfate methacrylate (CSMA) precursor was subsequently dried under vacuum at room temperature. The resulting product was dialyzed (1,000 Da MWCO) against dH<sub>2</sub>O for 3 days, followed by lyophilization for 4 days.



Figure 4.1. Modification reactions of chondroitin sulfate. (a) Chondroitin sulfate A (CS) and N-(3-aminopropyl)methacrylamide (APMAm) were reacted in the presence of N-(3-dimethylaminopropyl)-N'-ethylcarbodiimide (EDC) with or without N-hydroxysulfosuccinimide (sulfo-NHS) to yield chondroitin sulfate methacrylamide (CSMAm). Protons are labeled for <sup>1</sup>H NMR results in Figure 4.2. (b) CS and methacrylic anhydride were reacted in the presence of NaOH to yield chondroitin sulfate methacrylate (CSMA).



#### 4.2.2 Characterization of CS Nano- and Microspheres

#### 4.2.2.1 Proton Nuclear Magnetic Resonance Spectroscopy

Proton nuclear magnetic resonance (<sup>1</sup>H NMR) was utilized to determine the degree of the conjugation of the APMAm groups to the CS chains. The initial components (CS and APMAm) and resulting modified CS products (CSMAm and CSMA) were solubilized in deuterated water (D<sub>2</sub>O; Sigma-Aldrich), and <sup>1</sup>H NMR was measured on a Bruker AMX-400 spectrometer (Billerica, MA) at 400 MHz. The resulting spectra were analyzed by calibrating the region from 3.61-3.88 ppm according to the known number of protons in the chondroitin sulfate backbone. This provided an internal standard which allowed monitoring of the amount of modification attached to the different modification reactions.

#### 4.2.2.2 Formation and Characterization of Bulk CS Hydrogels

Solutions of CSMAm or CSMA at 90% water content in phosphate buffered saline (PBS) were crosslinked between 2 glass slides at 1 mm thickness under 15 mW/cm<sup>2</sup>, 365 nm UV light (UVP, Upland, CA) with 0.2 wt% Irgacure 2959 photoinitiator (D2959, Ciba, Ludwigshafen, Germany) for 12.5 minutes, and 6 mm diameter discs were punched out using a cork borer. After 1 day swelling in PBS, the wet weight of crosslinked hydrogels was recorded, and following lyophilization overnight, dry weight was recorded. Swelling ratio was calculated as wet weight/dry weight (n=4).

#### 4.2.2.3 Dynamic Light Scattering and Zeta Potential Measurements of CS Micelles

Dynamic light scattering (DLS) measurements were taken to determine the average diameter of CSMAm micelles in distilled, deionized water (ddH<sub>2</sub>O), and zeta potential measurements were also recorded as a measure of surface charge and colloidal



stability in ddH<sub>2</sub>O. DLS and zeta potential measurements were taken on a NICOMP 380ZLS (Particle Sizing Systems, Santa Barbara, CA) at 23°C. A helium-neon laser at 632.8 nm wavelength was used at a detector angle of 90°. Samples of 1.0 mg/mL CSMAm were run in ddH<sub>2</sub>O, while samples of unmodified CS at the same concentration served as controls (n=4). Prior to measurement, the solutions were filtered through a 5  $\mu$ m pore size cellulose acetate filter (Sartorius, Goettingen, Germany), and the sample was briefly centrifuged (14,000 RPM) to remove any dust particles from the solution. Hydrodynamic diameter was calculated using the inverse Laplace transform of the correlation function [329].

Zeta potential measures colloid mobility of charged particles in solution under an electric field. Zeta potential measurements were taken on 0.1 mg/mL CSMAm solutions in an electric field of 4 V/cm (1.6 V, 0.4 cm electrode distance; n=3). Zeta potential was calculated according to the Smoluchowski limit [330].

#### 4.2.2.4 CS Microsphere Fabrication and Characterization

CS microspheres were fabricated using a water-in-oil, single-emulsion technique similar to a previously described protocol [331]. CSMA (55.6 mg) was dissolved in PBS (440  $\mu$ L) and combined with ammonium persulfate (30  $\mu$ L, 0.3 M) (Sigma-Aldrich) and tetramethylethylenediamine (30  $\mu$ L, 0.3 M) (Sigma-Aldrich) free radical initiators on ice. This mixture was added drop-wise into corn oil at 4°C and homogenized at 3,800 RPM for 5 minutes. The temperature was then raised to 50°C with stirring for 30 minutes to promote crosslinking. The resulting solution was centrifuged at 300 RCF, and the corn oil was removed. Microspheres were then washed with dH<sub>2</sub>O and stored in dH<sub>2</sub>O at 4°C. The diameter of the microsphere population was analyzed using a Z2 Coulter Particle



Counter (Beckman Coulter, Fullerton, CA) equipped with a 100  $\mu$ m aperture. Microspheres were suspended in Isoton II diluent (Beckman Coulter) prior to counting.

For imaging, microspheres were treated with a solution of 1,9-dimethylmethylene blue (DMMB; Sigma-Aldrich) for 30 minutes to stain CS purple, washed with ddH<sub>2</sub>O, and imaged using an inverted microscope (Nikon Eclipse TE2000-U, Melville, NY).

#### 4.2.2.5 Complexation and Release from CS Microspheres

To examine the ability of negatively charged CS microspheres to electrostatically bind positively charged TGF- $\beta$ 1, approximately 2 mg of lyophilized microspheres were suspended in 500 µL of 50 ng/mL human recombinant TGF- $\beta$ 1 (Peprotech, Inc., Rocky Hill, NJ) and 1% bovine serum albumin (BSA) in PBS. Microspheres were incubated with TGF- $\beta$ 1 overnight at 4°C with shaking (n=3).

To examine the release kinetics of positively charged TGF- $\beta$ 1 and negatively charged tumor necrosis factor- $\alpha$  (TNF- $\alpha$ ), approximately 2 mg of lyophilized CS microspheres were loaded with 5 µL of 5 µg/mL human recombinant TGF- $\beta$ 1 or TNF- $\alpha$ (R&D Systems, Minneapolis, MN) per mg of microspheres overnight at 4°C. Microspheres were then suspended in 750 µL of solution containing 1% BSA in PBS (time 0) and incubated at 37°C with gentle shaking for 5 days.

At various time points, samples were centrifuged for 90 seconds at 4,000 RPM, and the supernatant was sampled and analyzed for factor content using the corresponding enzyme-linked immunosorbent assay (ELISA; R&D Systems) kit specific to human TGF- $\beta$ 1 or TNF- $\alpha$  (n=3).



#### 4.2.3 Cytocompatibility of CS Nano- and Microspheres

#### 4.2.3.1 In Vitro Cytotoxicity of CS Micelles and Microspheres on 2D BMSC Monolayers

For CSMAm micelle cytotoxicity studies, bovine bone marrow stromal cells (BMSCs) were isolated in a modification of the procedure outlined in [213]. Briefly, bone marrow was harvested from the tibia and femur of an immature calf (Research 87, Marlborough, MA). Cells were cultured in medium composed of low-glucose DMEM (Mediatech, Manassas, VA) supplemented with 10% fetal bovine serum (FBS; Thermo Scientific Hyclone, Logan, Utah), 1% antibiotic/antimycotic, 0.1% Fungizone antimycotic (Invitrogen, Carlsbad, CA), and 1 ng/mL bFGF (Peprotech). The adherent BMSCs were expanded until confluence and were then detached with 0.05% trypsin/EDTA (Mediatech), frozen in medium containing 20% FBS and 10% dimethyl sulfoxide (DMSO; Sigma-Aldrich), and cryopreserved in liquid nitrogen until use (P1).

Prior to use, BMSCs were thawed and plated for at least 3 days to eliminate transitory effects from the thawing process. After this time, cells were trypsinized (P2) and plated at 10,000 cells/cm<sup>2</sup> in a tissue culture-treated 96-well plate. After 24 hours to allow cell attachment, culture medium was replaced with 100  $\mu$ l of medium containing CSMAm micelles or CSMA microparticles equivalent to 1, 10, 100, 320 mg per 10<sup>6</sup> cells (0.064, 0.64, 6.4, and 20.5 mg/ml, respectively). After 24 hours of exposure to the CSMAm- or CSMA-containing media, cells were stained (30 minutes) with LIVE/DEAD stain (Invitrogen). Calcein stains live cells to fluoresce green (excitation/emission: 494/517 nm), and ethidium homodimer-1 stains dead cells to fluoresce red (excitation/emission: 528/617 nm). Dead controls were exposed to 70% methanol prior to staining, while live controls were cultured in normal culture medium without CS



particles. Fluorescent images were captured using an inverted microscope, and total fluorescence of each well was measured in a SpectraMax  $M2^e$  plate reader (Molecular Devices, Toronto, Canada) at the wavelengths specified above (n=3). Relative viability was analyzed in comparison to live controls, using the following equation:

Fraction of live cells = 
$$\frac{F(517)_{sample} - F(517)_{DEAD}}{F(517)_{LIVE} - F(517)_{DEAD}}$$

where  $F(517)_{sample}$ ,  $F(517)_{LIVE}$ , and  $F(517)_{DEAD}$  are the fluorescence at 517 nm of the sample, live controls, and dead controls, respectively.

### <u>4.2.3.2 In Vitro Cytotoxicity of CS Microspheres when Incorporated within 3D ESC</u> Embryoid Bodies

Undifferentiated D3 murine embryonic stem cells (ESCs) were maintained on gelatin-coated tissue culture dishes in DMEM supplemented with 15% FBS and  $10^3$  U/mL leukemia inhibitory factor (LIF; Millipore, Billerica, MA). Embryoid bodies (EBs) were formed using a single-cell suspension by forced aggregation in AggreWell<sup>TM</sup> 400 inserts (Stem Cell Technologies, Vancouver, CA) [332]. Briefly, 1.2 x  $10^6$  cells in 0.5 mL of medium were inoculated into AggreWell<sup>TM</sup> inserts, containing approximately 1,200 wells per insert, and centrifuged at 200 RCF for 5 minutes to cluster cells in the wells. Subsequently, 200 µL of CS microsphere solution was added in a 4:1 microsphere-to-ESC ratio, and a second centrifugation was performed at 200 RCF for 5 minutes. After 24 hours of culture, aggregates were removed from the wells using a wide-bore pipette and transferred to suspension culture on a rotary orbital shaker (40 RPM) to maintain the homogeneity of the population [333].

CS microsphere incorporation within EBs was examined through histological sections of EBs after 3 days of culture. EBs were sampled and fixed in a 10% formalin



solution. The EBs were washed in PBS and resuspended in HistoGel (Thermo Richard-Allan Scientific, Kalamazoo, MI) prior to paraffin processing. Sections of 5 µm were stained using Safranin-O (Sigma-Aldrich) for GAG detection, Fast Green (Sigma-Aldrich) for cytoplasm, and Weigert's hematoxylin (Sigma-Aldrich) for cell nuclei and imaged using a brightfield microscope (Nikon Eclipse 80i).

EBs with and without CSMA microspheres were also stained with LIVE/DEAD stain (30 minutes) and imaged by laser-scanning confocal microscopy (LSM 510; Carl Zeiss, Oberkochen, Germany) to examine the cytotoxicity of CSMA microspheres on ESCs within 3D EBs.

#### **4.2.4 Statistical Analysis**

All values were reported as mean  $\pm$  standard deviation. Statistical significance was determined using Minitab Statistical Software (v15, State College, PA) with a oneway analysis of variance (ANOVA) with Tukey's multiple comparison test ( $p \le 0.05$ ) for swelling ratio and cytotoxicity analysis and with a 2-sample t-test ( $p \le 0.05$ ) for growth factor loading efficiency and cumulative release. For swelling ratio analysis, the factor was hydrogel formulation. For cytotoxicity analysis, the factor was CS concentration. For growth factor loading efficiency and cumulative release, TGF- $\beta$ 1 and TNF- $\alpha$  cumulative release were compared at time 0 and day 5, respectively.



#### 4.3 Results

#### 4.3.1 Characterization of CS nano- and microspheres

#### <u>4.3.1.1 Proton nuclear magnetic resonance spectroscopy</u>

<sup>1</sup>H NMR spectra indicate that APMAm was successfully conjugated to CS by the EDC reaction. CSMAm product synthesized in the presence of sulfo-NHS clearly showed the appearance of vinyl groups, found in the APMAm moiety (Figures 4.2b-d; peaks  $B_1$ and B<sub>2</sub>, corresponding to the labeled protons in Figure 4.1a). The modified CS product also possessed proton peaks C and E, as labeled in Figures 4.2b-c, which were present in APMAm. <sup>1</sup>H NMR spectra also suggested that the degree of modification increased with increased ratio of starting reagents per CS disaccharide equivalent. While vinyl groups were clearly visible in 4:1, 3:1, and 5:2 molar ratio reactions, vinyl peaks were barely distinguishable in the 7:10 reaction product (Figure 4.2c,d). At 1:5 molar ratio and when synthesized at 4:1 ratio without sulfo-NHS, vinyl peaks were not visible (Figure 4.2e,f). Similarly, peaks C and E did not appear in these reaction products. This indicated that despite a modification of the CS from the original reagent (as indicated by the light scattering data below), APMAm protons were not visible by <sup>1</sup>H NMR at 1:5 molar ratio and without sulfo-NHS. Successful methacrylation of CS to form CSMA via reaction with methacrylic anhydride was also verified by <sup>1</sup>H NMR.

#### 4.3.1.2 Hydrogel formation and swelling

CSMAm crosslinking and swelling results in PBS also indicated that the degree of modification increased with increasing ratio of starting reagents per CS disaccharide equivalent. While 1:5 and 7:10 ratio reactions and the reaction without sulfo-NHS resulted in products that did not form robust hydrogels, at 5:2 molar ratio and above,



CSMAm crosslinked to form bulk hydrogels (Table 4.1). As the molar ratio increased from 5:2 to 3:1 to 4:1, swelling ratio significantly decreased from  $39.8 \pm 1.8$  to  $20.4 \pm 1.2$  to  $12.8 \pm 0.6$ , respectively. Decreased swelling is indicative of increased crosslinking density [123], likely as a result of higher degree of substitution of the methacrylamide in the EDC reaction. For comparison, crosslinked CSMA materials used for microsphere fabrication possessed a swelling ratio of  $31.3 \pm 3.6$ .



Figure 4.2. <sup>1</sup>H NMR spectra of (a) chondroitin sulfate, (b) APMAm, and reaction products synthesized at (c) 4:1, (d) 7:10, and (e) 1:5 molar ratios of APMAm, EDC, and sulfo-NHS to CS disaccharide, as well as (f) 4:1 ratio without sulfo-NHS. Protons are labeled according to corresponding diagrams in Figure 4.1a, and spectra indicated successful addition of methacrylamide moieties to the chondroitin sulfate.



| CS                   | Molar Ratio to CS |      |           | Swalling Patio |
|----------------------|-------------------|------|-----------|----------------|
| (by disaccharide MW) | APMAm             | EDC  | Sulfo-NHS | Swelling Ratio |
| 2.5 mM               | 4:1               | 4:1  | None      | No Gel         |
| 2.5 mM               | 1:5               | 1:5  | 1:5       | No Gel         |
| 2.5 mM               | 7:10              | 7:10 | 7:10      | No Gel         |
| 2.5 mM               | 5:2               | 5:2  | 5:2       | 39.8 ± 1.8 *   |
| 2.5 mM               | 3:1               | 3:1  | 3:1       | 20.4 ± 1.2 *   |
| 2.5 mM               | 4:1               | 4:1  | 4:1       | 12.8 ± 0.6 *   |

Table 4.1. Molar ratios of reactants in EDC modification of CS and resulting swelling ratios of crosslinked hydrogels in PBS

\* indicates statistically different from all other samples ( $p \le 0.05$ )

#### 4.3.1.3 Size distribution of CS micelles

Highly substituted CSMAm synthesized in the presence of sulfo-NHS did not appear to form stable micelles in an aqueous environment; therefore, only the 4:1 molar ratio without sulfo-NHS was used for micelle characterization. In order to accurately characterize the size of the CSMAm micelles, DLS analysis was performed. Micelle formation was examined by DLS for independent synthesis batches (n=4) and was verified to possess similar diameters. CSMAm micelles formed with an average diameter of 324.1  $\pm$  8.5 nm in ddH<sub>2</sub>O (Figure 4.3). A lower intensity collection of smaller CSMAm micelles appeared to form at 73.2  $\pm$  4.4 nm, as well. Unmodified CS at the same concentration, did not produce detectable scattering by DLS. Zeta potential measurements indicated that the CS micelles possessed a zeta potential of -38.7  $\pm$  1.1 mV.





Figure 4.3. Size characterization of CSMAm micelles. Dynamic light scattering measurements of hydrodynamic diameter indicated that a bimodal distribution was present with average diameters of  $324.1 \pm 8.5$  nm and  $73.2 \pm 4.4$  nm. Particle sizing measurements were consistent for 4 separate samples (shown in red, blue, green, and orange).

#### 4.3.1.4 Morphology and size distribution of CS microspheres

The CS microspheres had a smooth, round morphology, and Coulter Counter analysis indicated that the microspheres exhibited a unimodal size distribution with an average diameter of  $4.3 \pm 0.93 \mu m$  (Figure 4.4a,b). Additionally, the microparticles stained positively by DMMB for sulfated GAG, appearing purple (Figure 4.4a).

#### 4.3.1.5 Complexation and release from CS microspheres

To verify that a model positively charged growth factor was able to electrostatically complex with CS, the ability of CS microspheres to bind TGF- $\beta$ 1 in solution was examined. When the CS microspheres were incubated overnight with 25 ng of TGF- $\beta$ 1 (500 µL at 50 ng/mL), 97.4 ± 1.3% of free TGF- $\beta$ 1 in solution became incorporated with the CS microspheres. Additionally, in the following release study, positively charged TGF- $\beta$ 1 exhibited no appreciable release after loading in CS microspheres. Only 1.4 ± 0.3% (0.34 ± 0.08 ng per mg microspheres) of loaded TGF- $\beta$ 1



was observed in solution at time 0, representing a low amount of unloaded or loosely affiliated growth factor and high loading efficiency, and only 0.2% (0.04 ng) additional release was detected over the following 5 days (Figure 4.4c). In contrast, negatively charged TNF- $\alpha$ , loaded identically to TGF- $\beta$ 1, demonstrated 17.7 ± 3.5% (4.43 ± 0.87 ng per mg microspheres) of loaded TNF- $\alpha$  in solution at time 0, representing a significantly lower loading efficiency than TGF- $\beta$ 1, and 43.9 ± 9.1% cumulative release (10.97 ± 2.27 ng) over the first 15 hours from CS microspheres with no further detectable release after 15 hours (Figure 4.4c). Cumulative release of TNF- $\alpha$  from 3 hours to 5 days, correcting for differences in loading, was significantly greater than TGF- $\beta$ 1 released. Overall, after 5 days, significantly more release of TNF- $\alpha$  was observed compared to TGF- $\beta$ 1.

#### 4.3.2 Cytocompatibility of CS nano- and microspheres

#### 4.3.2.1 In vitro cytotoxicity of CS micelles and microspheres on 2D BMSC monolayers

LIVE/DEAD staining of bovine BMSCs cultured in monolayer in the presence of increasing concentrations of CSMAm micelles and CSMA microspheres revealed that cells remained largely viable at the lowest concentrations of 1 and 10 mg/10<sup>6</sup> cells with  $1.21 \pm 0.14$  and  $0.77 \pm 0.05$  viability, respectively, for CSMAm micelles and  $1.09 \pm 0.11$  and  $1.12 \pm 0.29$  viability, respectively, for CSMA microspheres, compared to live controls which possessed normalized viabilities of  $1.00 \pm 0.08$  and  $1.00 \pm 0.13$  for CSMAm and CSMA experiments, respectively (Figure 4.5a). Viability of all 1 and 10 mg/10<sup>6</sup> cells samples was not statistically different from live controls. Cells appeared mostly green (live) and possessed a spread morphology, characteristic of BMSCs (Figure 4.5b,c). At 100 mg/10<sup>6</sup> cells, a significant decrease in viability was observed in the presence of CSMAm with  $0.38 \pm 0.06$  viability, compared to live controls, as BMSCs



experienced a large decrease in cell density and spreading (Figure 4.5a,d), while viability did not significantly decrease in CSMA-containing medium (0.98  $\pm$  0.16 viability; Figure 4.5a,d). At the 320 mg/10<sup>6</sup> cells concentration, very few cells remained adhered to the surface (Figure 4.5e) in both CSMAm and CSMA samples, resulting in a significant decrease in viability to 0.18  $\pm$  0.07 and -0.05  $\pm$  0.06, respectively.



Figure 4.4. Characterization of CSMA microspheres. (a) CS microspheres possessed a smooth, rounded morphology and stained positively for sulfated GAG by DMMB (purple). Scale bar = 20  $\mu$ m. (b) CS microspheres exhibited a unimodal distribution with an average diameter of 4.3 ± 0.93  $\mu$ m. (c) Positively charged TGF- $\beta$ 1 (triangles)

experienced very little release from negatively charged CS microspheres after 5 days; however, negatively charged TNF- $\alpha$  (squares) experienced large early release within 15 hours. \* indicates statistically different loading efficiency (0 hours) compared to TGF- $\beta$ 1 loaded samples. + indicates statistically different cumulative release from 3 hours to 5 days compared to TGF- $\beta$ 1 release samples ( $p \le 0.05$ ).





Figure 4.5. LIVE/DEAD cytotoxicity analysis of bovine marrow stromal cells cultured in 2D monolayer in the presence of CSMAm micelles and CSMA microparticles. (a) Fraction of live cells per well compared to live controls, as determined by fluorescence values recorded at 517 nm wavelength. LIVE/DEAD fluorescence images in the presence of (b) 1 mg, (c) 10 mg, (d) 100 mg, and (e) 320 mg CSMAm or CSMA per 10<sup>6</sup> cells for 24 hours. Live cells appeared green, while dead cells appeared red. Scale bar = 200 µm. \* indicates statistically different from live controls ( $p \le 0.05$ ).

4.3.2.2 In vitro cytotoxicity of CS microspheres when incorporated within 3D ESC

#### embryoid bodies

CS microspheres were successfully incorporated within embryoid bodies using the forced aggregation method described. The microspheres stained positively in histological sections for sulfated GAG by Safranin-O, appearing red (Figure 4.6b).



Surrounding ESCs also appeared to be morphologically normal, compared to control EBs (Figure 4.6a), with no signs of cell death or adverse cell response after 3 days, suggesting that the incorporation of CS microspheres within 3D stem cell aggregates does not negatively impact the extracellular microenvironment. Additionally, LIVE/DEAD staining of EBs observed similar degrees of viability with CSMA microspheres incorporated (Figure 4.6d), when compared to control EBs (Figure 4.6c), indicating that ESCs remained largely viable when CSMA microspheres were incorporated within 3D EBs.



Figure 4.6. Cytotoxicity of mouse embryonic stem cells with CS microspheres incorporated within 3D embryoid bodies. (a) Cells in embryoid bodies without CS microspheres. (b) CS microspheres stained positively for sulfated GAGs by Safranin-O (red stain; black arrows). Surrounding cells appeared morphologically normal, with no signs of cell death after 3 days. Scale bar = 50 μm. A boxed region of cells and microspheres is also shown at higher magnification. Scale bar = 20 μm. (c) LIVE/DEAD fluorescence images of embryoid bodies without CS microspheres and (d) with incorporated CS microspheres, showed similar degrees of viability. Live cells appeared green, while dead cells appeared red. Scale bar = 200 μm.



#### 4.4 Discussion

In the first part of this study, a model reaction using EDC (and sulfo-NHS) to conjugate a methacrylamide moiety to CS was used to determine the effects of degree of modification on the properties of the resulting GAG-based material. This approach can be applied to any polysaccharide possessing a carboxylic acid moiety, including dermatan sulfate, heparin, heparan sulfate, and hyaluronic acid [334]. The resulting CSMAm material demonstrated increasing degrees of modification with greater molar ratios of starting reagents, resulting in higher degrees of crosslinking, as verified by <sup>1</sup>H NMR data and significantly decreased swelling in bulk hydrogels (Figure 4.2 and Table 4.1).

Interestingly, at lower degrees of modification, CSMAm self-assembled into micelles in aqueous solution. When reacted at 1:5 molar ratio and without sulfo-NHS, proton peaks associated with APMAm were not visible by <sup>1</sup>H NMR (Figure 4.2e,f), though a modification of CS was apparent by the formation of nanospheres, as confirmed by DLS analysis (Figure 4.3). It is likely that the APMAm groups were not visible by <sup>1</sup>H NMR, because these hydrophobic moieties were internalized within the micelle structure that formed in the NMR solvent (water). Similar results were observed in <sup>1</sup>H NMR spectra of CS-PLLA micelles in D<sub>2</sub>O, where internal protons were not visible due to limited mobility within the micelle core [324].

In this study, EDC chemistry was used to confirm that the degree of conjugation, and resulting molecular structure, could be controlled by altering the ratios of the starting reactants during synthesis; however, in order to efficiently scale up the size of the reaction while maintaining crosslinking capabilities to form microparticles, methacrylate



groups were added to CS by reacting the GAG with methacrylic anhydride. Resulting CSMA was crosslinked in a water-in-oil, single-emulsion to form CS microspheres. While the microspheres described here had an average diameter of  $4.3 \pm 0.93 \mu m$  (Figure 4.4a,b), the size of the microparticles could be easily modified by altering the emulsion conditions, such as speed of mixing and CSMA concentration, prior to crosslinking [335]. The results of this work suggested that by altering the degree of conjugation, CS-based materials can be developed that can be presented to cells on different size scales, including nanoscale micelles and larger microspheres, as well as controlled degree of crosslinking within particles. Therefore, this set of materials provides a versatile platform to explore presentation and release of various growth factors because the basic chemistry of the delivery vehicle is not altered, while a wide range of particle sizes is achieved.

Zeta potential results indicated that CSMAm micelles possessed of zeta potential of -38.7  $\pm$  1.1 mV, confirming that the micelles were negatively charged and stable in solution and suggesting that these micelles may have application in delivery of various cationic factors. Similarly, CSMA microspheres were shown to electrostatically complex with TGF- $\beta$ 1 at physiological pH. TGF- $\beta$ 1 was capable of a higher loading efficiency than TNF- $\alpha$ 1, as indicated by the time 0 data point, despite identical loading conditions, likely due to electrostatic interactions. Considerable release of TNF- $\alpha$  was also observed from CS microspheres over the first 15 hours; however, minimal TGF- $\beta$ 1 release was seen over 5 days, indicating that TGF- $\beta$ 1 remained bound to the CS microspheres during this time (Figure 4.4c). TGF- $\beta$ 1 possesses an isoelectric point (pI) of 9.5 and thus was selected to represent a positively charged growth factor [276], while TNF- $\alpha$  has a pI of 5.3, making it negatively charged at physiological pH [336]. Such results are comparable


to other CS-based hydrogels and microspheres, which have been reported to retain positively charged lysozyme, aprotonin, and vasopressin [325, 337]. TGF- $\beta$  isoforms are important for a number of processes in mammalian development [338-339], and have been particularly implicated in chondrogenesis *in vivo* [340] and *in vitro* [341-342], making this a relevant growth factor to investigate for directing differentiation of stem cells. While 100% of loaded TNF- $\alpha$  was not released over the time course of this study, as might be expected with little affinity between the cytokine and the CS matrix, this could be explained by the potential degradation of the protein over time in solution, resulting in artificially low ELISA readings after release [343].

Cumulative release data from these experiments suggested that significantly more TNF- $\alpha$  was liberated than TGF- $\beta$ 1, thus supporting the use of these materials for controlled delivery via electrostatic interaction with growth factors. Delivery kinetics can be further tailored by customizing the size and crosslinking density of the particles by altering the synthesis and fabrication parameters [344-345]. In combination with the release of growth factors through dissociation from the CS-based carrier, these biomaterial carriers could potentially be enzymatically cleaved by chondroitinase, a naturally secreted enzyme that digests chondroitin sulfate [346], thereby providing a second method for localized delivery.

In addition to control over size and the potential for degradation, the CS-based particles presented here are particularly advantageous because both types of particles are fabricated almost entirely from only CS molecules, for nearly uniform biochemical presentation to target cells. In previous work, CS has been combined with synthetic polymers like PLLA and PLGA, as well as natural materials including chitosan, alginate,



and gelatin [324, 326, 347-349] to produce micron-scale particles. However, to the best of our knowledge, this is the first time that CS microspheres and micelles have been synthesized without significant modification with other large polymers, making this system beneficial for specifically probing GAG-stem cell interactions.

While CS plays a role in many developmental processes, CS delivery vehicles may be particularly useful to promote chondrogenesis. CS-containing proteoglycans versican and perlecan regulate mesenchymal condensation and growth factor signaling during cartilage development [14-15], and culture in CS-containing hydrogels upregulated expression and production of chondrocytic ECM by encapsulated mesenchymal stem cells [8]. Therefore, in the future, the use of CS-based materials for controlled growth factor delivery may be advantageous as a means to direct cell differentiation to promote regeneration of cartilaginous tissues.

As a next step in the development of these materials as bioactive factor delivery vehicles, the cytotoxicity of CS was investigated *in vitro* using bovine BMSCs and murine ESCs. BMSCs and ESCs were chosen for their multipotent and pluripotent potential, respectively [77, 350]. In this study, BMSC monolayers were cultured with the presence of CS micelles and microspheres in the media for 24 hours to determine cytotoxicity in 2D, and CS microspheres were incorporated into ESC embryoid bodies for 3 days to examine cytocompatibility within 3D cell spheroids. ESCs are often cultured and differentiated within embryoid bodies; however, these dense, multicellular spheroids possess numerous boundaries to diffusion [322]. CS particles could potentially be utilized to deliver growth factors throughout the EB to improve homogeneous differentiation, compared to diffusion of soluble factors from the medium [322].



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BMSCs, cultured in 2D with moderate concentrations of CSMAm micelles and CSMA microspheres of 1 and 10  $mg/10^6$  cells, remained viable with no statistical difference from live controls (Figure 4.5a-c), while CSMA microspheres also remained cytocompatible up to  $100 \text{ mg}/10^6$  cells. Similarly, ESCs surrounding CS microspheres incorporated into EBs appear to be morphologically healthy with no apparent cell death (Figure 4.6), suggesting that CS microspheres could potentially be used to deliver growth factors within dense cell aggregates. At the highest concentration tested,  $320 \text{ mg}/10^6$  cells (as well as  $100 \text{ mg}/10^6$  cells for CSMAm), BMSCs experienced statistically decreased viability after 24 hours with the presence of CSMAm micelles and CSMA microspheres in the culture media (Figure 4.5a,d-e). Cell death may have resulted at high concentrations due to endocytosis of nanoscale CSMAm micelles by cells [323], or the high negative surface charge of CS particles may have interacted with receptors on the cell membrane [351] or altered the osmotic pressure of the medium, though these mechanisms were not further explored in this study. The range of concentrations in these studies, equivalent to 0.064, 0.64, 6.4, and 20.5 mg/ml, included and exceeded previously reported values used for cytotoxicity testing of CS-based microspheres and were defined according to ASTM International Standards F1903 and F813 regarding cytotoxicity testing of particles and materials, respectively [324, 326, 352-353]. Only 1 and 10 mg  $CS/10^6$  cells samples (0.064 and 0.64 mg/ml, respectively), which experienced no significant difference in cell viability from live controls, fell within previously reported ranges for cytotoxicity testing (0-5 mg/mL) of CS-PLLA microspheres [326]. Taken together, these results indicated that CS materials did not appear to affect the morphology of stem cells when cultured at moderate concentrations.



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Overall, these studies have demonstrated the formation of novel CS-based particles over a range of size scales with significant potential for use as ECM-derived carriers for delivery of charged growth factors to promote stem cell differentiation.

#### 4.5 Conclusions

The results of these studies demonstrated that CS chains were successfully modified to form amphiphilic self-assembling CS micelles on a nanoscale (~73 nm and ~320 nm), as well as microscale fabrication of CS microspheres (~4  $\mu$ m) for a wide range of sizes, and that CS microspheres retained their ability to bind positively charged growth factors. This flexibility in particle size can potentially provide a large degree of control for release of cationic factors from these CS materials, and the use of naturally-derived polysaccharide matrix without modification by other polymers may be especially advantageous in better understanding the role of GAGs in cell differentiation. Therefore, CS nano- and microspheres with the ability to deliver growth factors via electrostatic interaction provides a controlled, yet extremely versatile platform to further explore means to direct differentiation of stem cells for a variety of applications in tissue engineering and regenerative medicine.



#### **CHAPTER 5**

## CHEMICAL DESULFATION OF CHONDROITIN SULFATE FOR CONTROLLED GROWTH FACTOR RELEASE FROM GAG-BASED HYDROGELS

#### 5.1 Introduction

The highly sulfated GAGs heparin and heparan sulfate are known to play important roles in sequestration of positively charged growth factors *in vivo*, including basic fibroblast growth factor (bFGF), insulin-like growth factor (IGF), vascular endothelial growth factor (VEGF), platelet-derived growth factor (PDGF), and transforming growth factor- $\beta$  (TGF- $\beta$ ) [263, 272, 319]. Specifically, sulfate groups contribute to the highly negative fixed charge density of sulfated GAGs and their associated proteoglycans, facilitating electrostatic interactions with basic amino acid residues in positively charged growth factors. Due to its high sulfation, heparin possesses a stronger affinity than less sulfated GAGs, including chondroitin sulfate and the nonsulfated GAG hyaluronan [268, 354-357], and selective desulfation of heparin modulates binding in a sulfation-dependent manner and with special importance of 2-Osulfation for binding [265-267, 269-271, 358-359].

While heparin binding has been investigated in detail *in vivo* and *in vitro*, growth factor interactions with chondroitin sulfate (CS) have not been as well characterized. CS is especially prominent in cartilaginous tissues, as part of aggrecan proteoglycan, and while CS is known to play an important role in maintaining osmotic pressure within cartilaginous tissues, its role in growth factor signaling is currently not well understood.



Despite its lower degree of sulfation, compared to heparin, CS has also been shown to sequester several "heparin-binding" growth factors *in vitro*, and these electrostatic interactions have been exploited for controlled delivery and sustained release of growth factors, including PDGF-BB and IGF-1 [22, 262, 264]. Additionally, oversulfation of CS has been shown to enhance binding affinity in a sulfation-dependent manner, including to the chondrogenic growth factor TGF- $\beta$ 1, suggesting that sulfation regulates CS interactions with growth factors [25, 261].

While oversulfation of CS has been investigated to determine the role of sulfates in growth factor interactions, the nonsulfated GAG hyaluronan has commonly been used as a nonsulfated control. While hyaluronan is structurally similar to CS, hyaluronan plays very different roles than CS in vivo, and hyaluronan has activity independent of CS, including roles in ECM interactions, growth factor signaling, and known cell surface receptors like CD44 and receptor for hyaluronan-mediated motility (RHAMM) [360-364]. Use of a nonsulfated chondroitin variant would facilitate proper comparison across varying degrees of sulfation, while maintaining a consistent polysaccharide backbone and structure and only altering the sulfation pattern and resulting charge density of CS-based materials. The role of a nonsulfated chondroitin in growth factor binding and signaling has yet to be investigated; therefore, to further investigate the role of sulfation in interactions with growth factors, CS was chemically desulfated via acid methanol treatment to produce chondroitin. Desulfated chondroitin was then characterized for sulfation pattern and total charge to ensure that sulfates were removed without significant modification of the remaining GAG backbone.



As a naturally-derived cartilaginous ECM molecule, CS is a promising material for cartilage differentiation and repair. Additionally, the natural ability of CS to sequester growth factors may facilitate interaction with chondrogenic growth factors, such as TGF- $\beta$ 1, IGF-1, and various bone morphogenetic proteins (BMPs), while controlled desulfation of CS materials would offer further control over growth factor retention and release to promote differentiation of mesenchymal stem cells (MSCs). To examine the role of sulfation in growth factor interactions, CS and chondroitin were methacrylated and crosslinked in order to form bulk CS and chondroitin-based hydrogels containing varying ratios of CS and chondroitin to alter the relative degree of sulfation within the constructs. Release of chondrogenic growth factor TGF- $\beta$ 1 from CS and chondroitin hydrogels over 7 days and sequestration of soluble TGF- $\beta$ 1 out of solution were determined to examine the role of sulfation on interactions with TGF- $\beta$ 1.

#### 5.2 Materials and Methods

#### 5.2.1 Desulfation of Chondroitin Sulfate

Chondroitin sulfate was desulfated using an acidic methanol treatment for up to 7 days per established protocols [365-366]. Chondroitin sulfate A (primarily chondroitin-4-sulfate, Sigma-Aldrich, St. Louis, MO) or chondroitin sulfate C (primarily chondroitin-6-sulfate, Wako Chemicals USA, Richmond, VA) was stirred at 5.0 mg/mL in methanol (VWR, Radnor, PA) containing 0.5% v/v acetyl chloride (Acros Organics, Geel, Belgium). CS was centrifuged and acidic methanol was replaced either on days 1, 2, and 3 for a 3-day reaction or on days 1, 3, and 7 for a 7-day reaction to produce a methyl ester of chondroitin (Figure 5.1). The product was then dissolved in 20 mL distilled, deionized



water (ddH<sub>2</sub>O) per gram of starting CS before precipitation in an excess of ethanol. The methyl ester of chondroitin was washed in ethanol and ethyl ether (Fisher, Waltham, MA), vacuum dried at <5 mmHg, and stored at 4 °C.

Methyl ester of chondroitin was demethylated at 25 mg/mL in 0.1 M potassium hydroxide (KOH, Fisher) for 24 hours to produce chondroitin (Figure 5.1). The chondroitin product was then neutralized in 4 mL 100 mg/mL potassium acetate (Fisher) in 10% v/v acetic acid (VWR) per gram of starting product, and precipitated in an excess of ethanol. Chondroitin was washed in ethanol and ethyl ether, vacuum dried, and stored at 4°C until use.



Figure 5.1. Desulfation reaction of chondroitin sulfate. Chondroitin sulfate was desulfated in acidic method for up to 7 days to form a methyl ester of chondroitin. The methyl ester was demethylated in potassium hydroxide for 24 hours to yield chondroitin.

#### 5.2.2 Characterization of Desulfated Chondroitin

#### 5.2.2.1 Dimethylmethylene Blue Assay

Removal of sulfate groups in chondroitin was confirmed by dimethylmethylene blue (DMMB) assay for sulfated GAGs [302]. Standard curves from 0 to 50  $\mu$ g/mL chondroitin or chondroitin sulfate were assayed according to established protocols by DMMB, and absorbance was measured at 520 nm in a plate reader (SpectraMax M2e; Molecular Devices, Sunnyvale, CA) as a measure of sulfation. The slopes of the



chondroitin standard curves were compared to the CS standard curve to determine percent desulfation of the chondroitin product.

#### 5.2.2.2 Fourier Transform Infrared Spectroscopy

Because chondroitin sulfate C after 7 days in acidic methanol showed the greatest degree of sulfation, that chondroitin product was used in all future experiments, and compared to unmodified CS-C. To further verify the removal of sulfate groups in chondroitin, CS and chondroitin materials were analyzed by Fourier transform infrared (FTIR) spectroscopy. A 3 mg/mL solution of CS and desulfated chondroitin in deuterated water (D<sub>2</sub>O; Sigma-Aldrich) were spin coated onto a silicon crystal and FTIR was measured using a Bruker Vertex 70 ATR-FTIR spectrometer (Billerica, MA) with atmospheric compensation.

#### 5.2.2.3 Strong Anion Exchange High Performance Liquid Chromatography

Samples were analyzed by strong anion exchange high performance liquid chromatography (SAX-HPLC) by the University of Georgia Complex Carbohydrate Research Center (CCRC) to determine the disaccharide composition and average charge density of the CS and chondroitin materials. A 1 mg/mL GAG solution of either CS or chondroitin in 50 mM ammonium acetate (NH<sub>4</sub>OAc) buffer, pH 8 was digested into disaccharides by 0.1 mU/mL of chondroitinase ABC (Sigma-Aldrich) at 37°C for 24 hours. The chondroitinase enzyme was inactivated by heating to 100°C for 2 minutes, and the sample was centrifuged prior to HPLC analysis.

SAX-HPLC was carried out on an Agilent 1200 system (Santa Clara, CA) using a  $4.6 \times 250$  mm Waters Spherisorb analytical column (Milford, MA) with 5 µm particle size at 25°C, using an injection volume of 10 µL and a flow rate of 1.0 mL/min. The 2.5 mM



sodium phosphate ( $Na_3PO_4$ ) solvent at pH 3.5 was transitioned from a concentration of 0.036 M NaCl up to 1.2 M NaCl over the course of 55 minutes, gradually increasing the ionic strength of the buffer and causing elution of the disaccharides based on electrostatic charge, with more negatively charged disaccharides eluting later.

Disaccharide detection was performed by post-column derivatization. A 1:1 mixture of 0.25 M NaOH and 1% 2-cyanoacetamide was added to the eluent from the column from a binary HPLC pump at 0.5 mL/min. The eluent was then heated to 120°C in a 10-m reaction coil, followed by cooling in a 50-cm cooling coil, and directed into a Shimadzu fluorescence detector (excitation: 346 nm, emission: 410 nm; Kyoto, Japan). Based on these results, a predictive model was used to calculate average charge, based on known disaccharide structures containing negatively charged sulfate and carboxylate groups.

#### 5.2.2.4 Size Exclusion High Performance Liquid Chromatography

CS and chondroitin samples were also analyzed by size exclusion high performance liquid chromatography (SEC-HPLC) by the University of Georgia CCRC to determine the average molecular weight of the CS and chondroitin chains. Solutions of CS or chondroitin were prepared at 2 mg/mL in 50 mM sodium sulfate (Na<sub>2</sub>SO<sub>4</sub>) buffer, pH 5.0, and partially depolymerized heparin fractions (4.2, 9.0, 12.0, 15.0, and 20.0 kDa) were used as molecular weight standards. Separations were carried out using a TSKGel G3000SWXL column (Tosoh Bioscience, Stuggart, Germany, 7.8 mm ID x 30 cm) and a TSKGel G2000SWXL column (7.8 mm ID x 30 cm), connected in series, on an Agilent 1200 LC instrument using refractive index detection with an injection volume of 50 µL and a flow rate of 0.5 mL/min.



A standard curve was constructed based on the molecular weights and elution volumes of the heparin standards. The data were baseline corrected, and the retention times were converted to molecular weights using the standard curve. The weight fraction  $w_i$  of each data point was calculated by dividing the detector response by the sum of detector responses over the whole peak width. The weight average molecular mass was calculated as  $M_w = \sum w_i M_i$ , and the number average molecular mass was calculated as  $M_n = 1/\sum (w_i/M_i)$ . Polydispersity index (PI) was calculated as  $PI = M_w/M_n$ .

#### 5.2.3 Synthesis of Crosslinkable Hydrogel Materials

#### 5.2.3.1 PEG-DA Synthesis

Poly(ethylene glycol)-diacrylate (PEG-DA) polymers were synthesized according to established protocols [191]. 3,400 Da ( $M_n$ ) PEG (Sigma-Aldrich) was solubilized in distilled, anhydrous methylene chloride (MeCl, Fisher) to produce a 60% (m/v) solution. Triethylamine (TEA, Sigma-Aldrich) was added to the dissolved PEG, and acryloyl chloride (AcCl, Sigma-Aldrich) was slowly added dropwise at 2:1 AcCl:PEG and 1:1 AcCl:TEA molar ratios to create a 40% w/v solution. The reaction was allowed to proceed at room temperature under nitrogen gas, and the reaction was left stirring overnight to ensure completion. 2 M potassium carbonate ( $K_2CO_3$ , Fisher) was used at a 2:1 K<sub>2</sub>CO<sub>3</sub>:AcCl molar ratio to extract TEA to the aqueous phase. The organic phase was allowed to separate from the aqueous phase, and anhydrous magnesium sulfate (MgSO<sub>4</sub>, Fisher) was used to remove any remaining aqueous phase from the solution. PEG-DA was then precipitated in ethyl ether and filtered, followed by vacuum drying at <5 mmHg. Dry polymers were stored in sealed containers at -20°C until use.



#### 5.2.3.2 OPF Polymer Synthesis

Oligo(poly(ethylene glycol) fumarate) 10K polymers were synthesized according to established protocols [367]. 10,000 Da ( $M_n$ ) PEG (Sigma-Aldrich) was azeotropically distilled in toluene (Fisher) and then combined with distilled, anhydrous MeCl to produce a 40% (v/v) solution. Fumaryl chloride (FuCl, Sigma-Aldrich) and TEA were slowly added dropwise over a period of 2 hours at 0.9 FuCl:PEG and 1:2 FuCl:TEA molar ratios. During this time, the reaction was allowed to proceed at 0°C under nitrogen gas, after which the reaction was left stirring for 72 hours under nitrogen at room temperature to ensure reaction completion. After rotovaporing (Buchi, Flawil, Switzerland) to remove excess MeCl, the resulting polymer was recrystallized twice with ethyl acetate (Fisher) and washed three times in ethyl ether, followed by vacuum drying at <5 mmHg. Dry polymers were stored in sealed containers at -20°C before use.

#### 5.2.3.3 Methacrylation of Chondroitin Sulfate and Chondroitin

CS and chondroitin were methacrylated with glycidyl methacrylate per established protocols [368]. CS or chondroitin was dissolved at 1% w/v in a 50:50 mixture of acetone (VWR) and ddH<sub>2</sub>O and allowed to stir at room temperature overnight. A 20-fold molar excess of TEA per CS or chondroitin disaccharide was added to the solution, and a 20-fold molar excess of glycidyl methacrylate (GMA, Sigma-Aldrich) per disaccharide was added dropwise to the solution. The reaction was allowed to stir at room temperature for 24 hours to produce CS-methacrylate (CS-MA) (Figure 5.2) and chondroitin-methacrylate (Ch-MA). The resulting products were dialyzed first in 50:50 acetone:water for 24 hours (1,000 Da MWCO), and then in distilled water (dH<sub>2</sub>O) for 2 days to remove unreacted reagents. The methacrylated products were lyophilized



(Labconco, Kansas City, MO) for 4 days to produce a dry product, and stored at -20°C until use.



Figure 5.2. Chondroitin sulfate methacrylation reaction. Chondroitin sulfate was methacrylated with glycidyl methacrylate in 50:50 acetone:ddH<sub>2</sub>O for 24 hours in the presence of triethylamine to form chondroitin sulfate methacrylate (CS-MA).

#### 5.2.4 Characterization of Crosslinkable Hydrogel Materials

#### 5.2.4.1 Gel Permeation Chromatography

PEG-DA ( $M_n = 3760 \pm 50$ , PI = 1.1 ± 0.02), PEG 3.4K ( $M_n = 3440 \pm 30$ , PI = 1.1 ± 0.001), OPF 10K ( $M_n = 28,100 \pm 760$ , PI = 5.2 ± 0.4), and PEG 10K ( $M_n = 12,900 \pm 210$ ; PI = 1.1 ± 0.002) were characterized via gel permeation chromatography as previously reported [174]. The molecular weight distribution of PEG-based polymers was characterized by gel permeation chromatography (Prominence; Shimadzu) equipped with a refractive index detector (Shimadzu). Polymer samples were dissolved in chloroform, filtered (0.45 µm filter, Whatman, Maidstone, UK), and injected into a column (Waters) at a flow rate of 1 mL/min (n=3).

#### 5.2.4.2 Proton Nuclear Magnetic Resonance

Proton nuclear magnetic resonance (<sup>1</sup>H NMR) was utilized to determine the degree of the conjugation of the GMA groups to the CS and chondroitin chains. The



initial components (CS, chondroitin, and GMA) and resulting modified products (CS-MA and Ch-MA) were solubilized in  $D_2O$ , and <sup>1</sup>H NMR was measured on a Bruker AMX-400 spectrometer at 400 MHz. The resulting spectra were analyzed by calibrating the region from 4.18-3.75 ppm according to the known number of protons in the chondroitin sulfate backbone, providing an internal standard to determine the degree of modification by the methacrylation reaction.

#### 5.2.4.3 Degradation of Modified Chondroitin Sulfate and Chondroitin Materials

To determine if the ability of chondroitinase enzyme to degrade CS was affected by either desulfation, methacrylation, or crosslinking, the degradation of modified CS-MA and Ch-MA was determined in solution and in crosslinked hydrogels. Soluble chondroitinase activity was assayed per established protocols [369-371]. 2 mg/mL CS-MA and chondroitin-MA solutions in buffer of 250 а mΜ tris(hydroxymethyl)aminomethane (Tris, Sigma-Aldrich), 300 mM sodium acetate (Sigma-Aldrich) with 0.05% bovine serum albumin (BSA, Sigma-Aldrich) at pH 8.0 were incubated with 0.06 U/mL chondroitinase ABC. Accumulation of the  $\Delta^{4,5}$ unsaturated disaccharide degradation product at 37°C was monitored by measuring the increase in absorbance at 232 nm UV light in a UV-transparent 96-well assay plate (Corning Incorporated, Corning, NY) by a plate reader (SpectraMax M2e; Molecular Devices) every minute for 1 hour, compared to CS-MA and Ch-MA blanks without enzyme (n=3).

To determine the ability of crosslinked CS-MA and Ch-MA hydrogels to degrade in the presence of chondroitinase enzyme, 100% CS-MA or 100% Ch-MA hydrogels at 90 wt%  $H_2O$  were crosslinked in 6 mm diameter, 1 mm deep cylindrical molds with



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0.018 M ammonium persulfate (APS, Sigma-Aldrich) and tetramethylethylenediamine (TEMED, Sigma-Aldrich) for 10 minutes at 37 °C, and swelled overnight in phosphatebuffered saline (PBS). CS-MA and Ch-MA hydrogels were then transferred into 0.125 U/mL chondroitinase ABC in 250 mM Tris, 300mM sodium acetate buffer with 0.05% BSA at pH 8.0 and incubated at 37°C. Complete degradation was determined when the bulk hydrogel was no longer visible in solution (n=3).

#### 5.2.4.4 Swelling of GAG/PEG Hydrogels

Polymer solutions containing a 1:1 ratio of OPF 10K:PEG-DA and either 10% or 50% CS-MA or Ch-MA at 90% water content in PBS were crosslinked in 6 mm diameter, 1 mm deep cylindrical molds with 0.018 M APS/TEMED thermal initiator system for 10 minutes at 37°C, and swelled overnight in PBS. A 60% PEG-DA:40% OPF mixture that possessed similar swelling properties to 50% Ch-MA formulations was used as a PEG-only swelling control for 50% Ch-MA. After 1 day swelling in PBS, the wet weight of crosslinked hydrogels was recorded, and following lyophilization overnight, dry weight was recorded. Swelling ratio was calculated as wet weight/dry weight (n=5).

As an estimate of osmotic pressure based on fixed negative charge in GAG hydrogels, Donnan swelling pressure was estimated in a 0.15 M NaCl solution at 37°C. According to Donnan equilibrium, electroneutrality, and osmotic pressure laws, Donnan osmotic pressure is calculated by the equation:

$$\pi = RT[\sqrt{(c^F)^2 + 4(c^*)^2} - 2c^*]$$

where  $\pi$  is the Donnan osmotic pressure, R is the universal gas constant (8.314 L kPa K<sup>-1</sup> mol<sup>-1</sup>), T is the absolute temperature (310 K), c<sup>F</sup> is the fixed charge density in mol of charge per L of interstitial fluid, and c<sup>\*</sup> is the salt concentration of the external electrolyte



solution (0.15 M NaCl) [372-374]. Fixed charge density was calculated based on the GAG disaccharide concentration in the hydrogels, as well as the average charge per disaccharide, determined by SAX-HPLC, as described in Section 5.2.2.3.

#### 5.2.5 Interaction of CS and Chondroitin Hydrogels with TGF-β1

#### 5.2.5.1 Release of TGF-β1 from CS and Chondroitin Hydrogels

To investigate the role of sulfate moieties in the electrostatic complexation and release of positively charged TGF- $\beta$ 1, 1:1 OPF:PEG-DA hydrogels containing 10% CS-MA, 50% CS-MA, or 50% Ch-MA (90 wt% H<sub>2</sub>O) were sterilely crosslinked in 6 mm diameter, 1 mm deep cylindrical molds with 0.018 M APS/TEMED for 10 minutes at 37°C, and swelled overnight in PBS. Hydrogels were then lyophilized for 24 hours, and loaded by reswelling with 15 µL of 12.5 µg/mL TGF- $\beta$ 1 solution (Peprotech, Rocky Hill, NJ) for 16 hours at 4°C. Loading correlated to a ~1:500,000 TGF- $\beta$ :disaccharide molar ratio in 50% GAG hydrogels and to a ~1:100,000 TGF- $\beta$ :disaccharide molar ratio in 10% CS. Hydrogels were then incubated at 37°C in 500 µL 1% BSA in PBS for 7 days. At 0, 5, 12, and 24 hours, and after 2, 3, 5, and 7 days, the supernatant was sampled and analyzed for TGF- $\beta$ 1 content by enzyme-linked immunosorbent assay (ELISA, R&D Systems, Minneapolis, MN). Cumulative release was calculated for each hydrogel over 7 days (n=4).

#### 5.2.5.2 Formation of 50% Total GAG Hydrogels with Varying Sulfation

To demonstrate that hydrogels can be fabricated with a constant GAG content and varying degrees of sulfation, CS-MA and Ch-MA were incorporated into 1:1 OPF:PEG-DA hydrogels at varying mass ratios while maintaining total GAG content constant at 50 wt%, according to Table 5.1. A 60% PEG-DA:40% OPF mixture that possessed similar



swelling properties to 50% Ch-MA formulations was used as a PEG-only swelling control for 50% Ch-MA. Polymer was dissolved in PBS for an initial water content of 90% w/w, and all macromer solutions were filter sterilized through a 0.2  $\mu$ m pore filter (Nalgene, Rochester, NY). 30  $\mu$ L of macromer solution was crosslinked with 0.018 M APS/TEMED for 10 minutes at 37°C in cylindrical molds, resulting in hydrogel disks that were 6 mm diameter and 1 mm thick. All hydrogels were swelled in PBS. After 7 days in PBS to demonstrate the retention of GAG, hydrogels were stained overnight in DMMB solution (n=2).

|                         | Mass Ratios of Total Dry Polymer |                   |     |         |  |  |  |
|-------------------------|----------------------------------|-------------------|-----|---------|--|--|--|
|                         | CS-MA                            | S-MA Ch-MA PEG-DA |     | OPF 10K |  |  |  |
| 50% CS-MA               | 50%                              | 0%                | 25% | 25%     |  |  |  |
| 10% CS-MA/<br>40% Ch-MA | 10%                              | 40%               | 25% | 25%     |  |  |  |
| 1% CS-MA/<br>49% Ch-MA  | 1%                               | 49%               | 25% | 25%     |  |  |  |
| 50% Ch-MA               | 0%                               | 50%               | 25% | 25%     |  |  |  |
| PEG Control             | 0%                               | 0%                | 60% | 40%     |  |  |  |

 Table 5.1. Hydrogel formulations with 50 wt% total GAG

### 5.2.5.3 Pull-Down (Depletion) of TGF-β1 from Solution by CS-MA and Ch-MA Hydrogels

To investigate the role of sulfation in TGF-β1 sequestration by 3D hydrogels, TGF-β1 pull-down (or depletion) from solution by PEG-based hydrogels containing varying amounts of CS-MA or desulfated Ch-MA was investigated. CS-MA and Ch-MA hydrogels were fabricated as described in Section 5.2.5.2, according to the 50% CS-MA, 10% CS-MA/40% Ch-MA, and 50% Ch-MA formulations in Table 5.1. A 60% PEG-DA:40% OPF mixture that possessed a similar swelling properties to 50% Ch-MA



formulations was used as a PEG-only swelling control. All hydrogels were swelled overnight in PBS.

After unconstrained swelling in PBS overnight, hydrogels were incubated in 1.0 mL solution of 2.0 ng/mL TGF- $\beta$ 1 in 1% BSA in PBS for 24 hours at 37°C with gentle shaking. To inhibit electrostatic binding with hydrogels, pull-down was also measured with an additional 0.5 M NaCl or 10 mg/mL soluble CS in the TGF- $\beta$ 1 solution. The TGF- $\beta$ :disaccharide molar ratio in the solutions correlated to a ~1:50,000,000 molar ratio in 50% GAG hydrogels and to a ~1:10,000,000 molar ratio in 10% CS. After 24 hours, the supernatant was collected and frozen at -20°C until analysis. TGF- $\beta$ 1 pull-down by the hydrogels was determined by assaying the remaining TGF- $\beta$ 1 in solution by ELISA (n=5).

#### **5.2.6 Statistical Analysis**

All values were reported as mean  $\pm$  standard deviation. A one- or two-factor analysis of variance (ANOVA) was used to determine statistical significance of groups, and Tukey's *post hoc* multiple comparison test with significance set at *p*≤0.05 indicated significance between individual samples. For all one-factor ANOVAs (swelling and TGF- $\beta$ 1 release), the factor was hydrogel type. For pull-down experiments, the two factors were hydrogel type and buffer composition. Statistical analysis was carried out using Minitab (v15.1, State College, PA).



#### 5.3 Results

#### 5.3.1 CS-C was desulfated after 7-day acidic methanol treatment

DMMB assay was used to measure sulfation level of the chondroitin products after acidic methanol treatment. While CS-A only experienced 54% desulfation after 3 days, CS-C experienced removal of 80% of the sulfates after the same time (Figure 5.3a). Extension of the acidic methanol treatment time from 3 to 7 days resulted in nearly complete desulfation of CS-C, as indicated by zero slope in the desulfated chondroitin standard curve by DMMB assay. As further confirmation of desulfation, FTIR spectroscopy was used to examine the bonds present in CS and chondroitin. FTIR spectroscopy verified the disappearance of sulfate peaks at 1100-1250 cm<sup>-1</sup> (Figure 5.3b, black box), while the remaining bonds in CS appeared to remain unchanged.



Figure 5.3. CS-C was desulfated by acidic methanol treatment for 7 days. (a) Dimethylmethylene blue (DMMB) assay standards for CS showed a linear trend with positive slope, indicating the presence of sulfates; however, the standard for the chondroitin product from CS-C desulfated for 7 days showed zero slope indicating the absence of sulfate groups in chondroitin. CS-A and CS-C experienced incomplete desulfation after 3 days, but CS-C showed greater susceptibility to desulfation in the same time than CS-A. (b) Fourier transform infrared (FTIR) spectroscopy showed the absence of sulfate peaks from 1100-1250 cm<sup>-1</sup> (in black box) in chondroitin materials, compared to CS.



# 5.3.2 Disaccharide composition of chondroitin indicated a reduction of negative charge density

SAX-HPLC analysis was performed to determine the disaccharide composition of CS-C and desulfated chondroitin. It was determined that CS-C disaccharides were approximately 57.8% 6-sulfated and 26.8% 4-sulfated, as well as 1.3% nonsulfated (Table 5.2). The desulfated chondroitin product, however, was 98.5% nonsulfated and only 1.5% 6-sulfated. From the elution profiles, nonsulfated disaccharides, which are the least charged, eluted earliest (9-11 min), followed by the monosulfated disaccharides (18-23 min), and the most negatively charged disulfated disaccharides eluting latest (41-47 min) (Figure 5.4a-b). The observed shift in desulfated chondroitin to a primarily nonsulfated form also indicates a reduction in negative charge density in chondroitin, compared to CS. Based on these results, a predictive model calculated CS to have an average charge of -2.3 per disaccharide, while chondroitin had an average charge of -1.02, indicating that desulfation resulted in a decrease in negative charge density of CS by over two-fold.

|             | ΔUA-<br>GalNAc | ΔUA-<br>GalNAc(6S) | ΔUA-<br>GalNAc(4S) | ΔUA(2S)-<br>GalNAc(6S) | ΔUA-<br>GalNAc(4S,6S) | ΔUA(2S)-<br>GalNAc(4S) |
|-------------|----------------|--------------------|--------------------|------------------------|-----------------------|------------------------|
| CS          | 1.3%           | 57.8%              | 26.8%              | 12.7%                  | 0.7%                  | 0.7%                   |
| Chondroitin | 98.5%          | 1.5%               | nd                 | Nd                     | nd                    | nd                     |

Table 5.2. CS and chondroitin disaccharide composition by SAX-HPLC analysis

 $\Delta UA = \Delta^{4,5}$ -unsaturated uronic acid, GalNAc = N-acetylgalactosamine, nd = not detected

SEC-HPLC analysis was also performed to determine the average molecular weight of the full CS and chondroitin chains. It was determined that CS had a weight averaged molecular mass  $(M_w)$  of 17,880 Da and a number averaged molecular mass



 $(M_n)$  of 16,300 Da (PI=1.1) (Figure 5.4c). Chondroitin, however, was notably smaller with an  $M_w$  of 6,310 Da and an  $M_n$  of 5,230 Da (PI=1.2), resulting in a later elution time than CS (Figure 5.4d). This represented a 67.9% decrease in molecular weight ( $M_n$ ) from CS, while sulfates only accounted for 17.1% of mass based on disaccharide composition.



Figure 5.4. SAX-HPLC and SEC-HPLC analysis of CS and chondroitin. (a-b) SAX-HPLC analysis of (a) CS and (b) chondroitin disaccharide composition showed a shift in negative charge density in CS and chondroitin materials. Nonsulfated disaccharides (charge: -1) eluted from 9-11 min, monosulfated disaccharides (charge: -2) eluted from 18-23 min, and disulfated disaccharides (charge: -3) eluted from 41-47 min. Based on disaccharide composition, CS possessed an average charge of -2.3 per disaccharide, while chondroitin was 98.5% nonsulfated and possessed an average charge of -1.02 per disaccharide. (c-d) SEC-HPLC analysis of (c) CS and (d) chondroitin molecular weight suggested that while CS had a M<sub>n</sub> of 16.3 kDa (PI=1.1), and chondroitin was smaller with an M<sub>n</sub> of 5.23 kDa (PI=1.2), resulting in a later elution time than CS.



#### 5.3.3 Degree of methacrylation by glycidyl methacrylate by H<sup>1</sup> NMR

H<sup>1</sup> NMR spectra indicated that GMA was successfully conjugated to CS. Vinyl peaks were visible at 5.6 and 6.0 ppm, confirming the methacrylation of CS and chondroitin by GMA to form CS-MA and Ch-MA, respectively (Figure 5.5, in black boxes). Peak integration also indicated that on average one GMA molecule was conjugated per every 4.3 disaccharides in CS-MA and every 3.7 disaccharides in Ch-MA.



Figure 5.5. <sup>1</sup>H NMR analysis of methacrylated CS and chondroitin. <sup>1</sup>H NMR confirmed methacrylation of CS and chondroitin with glycidyl methacrylate. Compared to (a) CS and (b) chondroitin, <sup>1</sup>H NMR spectra verified the presence of vinyl groups (in black boxes) in (c) CS-MA and (d) Ch-MA at 5.6 and 6.0 ppm, resulting from methacrylation of CS and chondroitin.



## 5.3.4 Chondroitinase ABC maintains enzymatic activity to degrade CS-MA and Ch-MA

Soluble methacrylated CS and chondroitin degraded in chondroitinase ABC with an increase in absorbance of 232 nm UV light over time, indicating an increase of  $\Delta^{4,5}$ unsaturated disaccharide degradation products (Figure 5.6). Complete degradation of CS-MA and Ch-MA occurred within ~30 minutes in solution. Additionally, crosslinked CS-MA and Ch-MA hydrogels completely degraded overnight in chondroitinase enzyme.



Figure 5.6. Degradation of CS-MA and Ch-MA by chondroitinase ABC. 2 mg/mL soluble CS-MA and chondroitin-MA materials were degradable in the presence of 0.06 U/mL chondroitinase ABC, as detected by absorbance of 232 nm light.

#### 5.3.5 Swelling of PEG hydrogels containing CS-MA and Ch-MA

50% CS-MA hydrogels possessed a fold swelling ratio of  $25.2 \pm 1.4$ , and 50% CS-MA swelled significantly more than 50% Ch-MA hydrogels which swelled  $16.9 \pm 2.1$  fold. 10% CS-MA with a  $22.7 \pm 2.4$  fold swelling ratio, however, did not swell significantly differently from 50% CS-MA (Figure 5.7a). 60% PEG-DA/40% OPF 10K hydrogels (17.1 ± 0.5 fold) that swell similarly to 50% Ch-MA hydrogels were fabricated



as a PEG-based swelling control. Based on estimatations of Donnan osmotic pressure in a 0.15 M NaCl electrolyte solution, 50% CS-MA hydrogels possessed an osmotic pressure of 280 kPa while 50% Ch-MA hydrogels had lower osmotic pressure of 93 kPa, representing a 3.0-fold difference in osmotic swelling pressure based on charge alone.

#### **5.3.6** TGF-β1 retention and release by CS and chondrotin

TGF- $\beta$ 1 release studies from CS-MA and Ch-MA hydrogels observed increased release of TGF- $\beta$ 1 correlating with decreased sulfation. Hydrogels containing 50% CS displayed the greatest retention of TGF- $\beta$ 1 over 7 days, with only 3.6 ± 1.0 ng TGF- $\beta$ 1 released after 7 days (Figure 5.7b). 10% CS hydrogels demonstrated significantly greater TGF- $\beta$ 1 release than 50% CS-MA with 5.6 ± 0.5 ng TGF- $\beta$ 1 released. Additionally, 50% chondroitin hydrogels exhibited the greatest release of 6.8 ± 0.2 ng TGF- $\beta$ 1.



Figure 5.7. Swelling and TGF-β1 release from CS-MA and Ch-MA hydrogels. (a) Crosslinked 50% Ch-MA hydrogels swelled significantly less than 50% CS-MA and 10% CS-MA hydrogels. 40% OPF/60% PEG-DA hydrogels were formed as PEGbased swelling controls for 50% Ch-MA. (b) 50% CS-MA hydrogels demonstrated greatest retention of TGF-β1 over 7 days, while 10% CS-MA gels exhibited greater release, and 50% chondroitin-MA gels displayed the greatest release, indicating that

TGF- $\beta$ 1 release decreased with increasing degrees of sulfation. \* indicates significantly lower fold swelling that 50% CS-MA ( $p \le 0.05$ ). # indicates significantly lower fold swelling that 10% CS-MA ( $p \le 0.05$ ). + indicates significantly greater release than 50% CS-MA after 7 days ( $p \le 0.05$ ).



#### 5.3.7 TGF-β1 pull-down by hydrogels with varying CS content

GAG-containing hydrogels were fabricated containing varying amounts of CS-MA. While total GAG was maintained at 50 wt%, the fraction of CS-MA was varied with the remainder balanced with nonsulfated Ch-MA. A pink/purple color indicated the present of sulfated GAG in a concentration-dependent manner, while a blue color indicated the absence of sulfates. Staining with DMMB indicated that CS-containing hydrogels stained positively for sulfation after 7 days, and increasing degrees of staining were apparent in hydrogels as sulfation increased up to 50% CS, while PEG only hydrogels remained blue (Figure 5.8a).

In pull-down (depletion) studies from solution, all hydrogels exhibited significant pull-down of TGF- $\beta$ 1, compared to blank wells without gels, where less TGF- $\beta$ 1 remaining in solution indicated greater pull-down; however, pull-down in 50% CS-MA hydrogels was significantly greater than less sulfated hydrogel formulations (Figure 5.8b). 50% CS-MA depleted 55.9 ± 1.5% of available TGF- $\beta$ 1 out of solution, compared to blanks. Ch-MA and PEG control hydrogels, on the other hand, experienced the least pull-down of TGF- $\beta$ 1 with 29.4 ± 2.7% and 33.8 ± 3.0% depletion, respectively. Hydrogels containing 10% CS-MA/40% Ch-MA responded in a concentration-dependent manner with 36.4 ± 3.3% depletion, which was significantly greater pull-down than 50% CS-MA and less than 50% CS-MA, indicating that greater depletion of TGF- $\beta$ 1 was observed with increasing sulfation.





Figure 5.8. TGF-61 pull-down by CS-MA and Ch-MA hydrogels with varying degrees of sulfation. (a) DMMB staining indicated increasing degrees of sulfation in hydrogels from 0% to 50% CS-MA content after 7 days. Sulfated GAGs stained purple/pink, while nonsulfated materials remained blue. All hydrogels contained 50% total GAG (with CS balanced with chondroitin). Scale bars = 1.0 mm. (b) All hydrogels demonstrated significant pull-down compared to blanks ( $p \le 0.05$ ), where less TGF-β1 remaining in solution indicated greater pull-down. 50% CS-MA hydrogel exhibited the greatest pull-down of soluble TGF-81 out of solution, while nonsulfated 50% Ch-MA and PEG control hydrogels exhibited the least depletion in 1% BSA. 10% CS-MA/40% Ch-MA hydrogels corresponded in a sulfationdependent manner with an intermediate level of depletion. Incubation in 0.5 M NaCl and 10 mg/ml soluble CS significantly decreased depletion by 50% Ch-MA, but not 50% CS-MA hydrogels. § indicates significantly greater pull-down than PEG controls ( $p \le 0.05$ ). \* indicates significantly greater pull-down than 50% Ch-MA  $(p \le 0.05)$ . # indicates significantly greater pull-down than 10% CS-MA/40% Ch-MA  $(p \le 0.05)$ . + indicates significantly less pull-down than the same hydrogel type in 1% BSA (*p*≤0.05).

In the presence of 0.5 M NaCl, pull-down of TGF- $\beta$ 1 by 50% Ch-MA and 10% CS-MA/40% Ch-MA hydrogels significantly decreased, while pull-down by 50% CS-MA did not decrease. Depletion by 50% Ch-MA hydrogels decreased to 14.8 ± 2.0%, and



depletion by 10% CS-MA/40% Ch-MA decreased to 23.0  $\pm$  2.7% in 0.5 M NaCl. Additionally, in 10 mg/mL soluble CS, a similar response was observed. In the presence of 10 mg/mL soluble CS, pull-down by 50% Ch-MA hydrogel was significantly decreased to 16.4  $\pm$  1.8% depletion; however, pull-down by crosslinked 50% CS-MA hydrogels did not decrease. TGF- $\beta$ 1 pull-down by PEG control hydrogels, on the other hand, remained unchanged in 0.5 M NaCl, compared to in 1% BSA, with 33.1  $\pm$  3.5% depletion, while soluble CS resulted in a slight decrease in depletion by PEG controls to 26.7  $\pm$  5.5%.

#### 5.4 Discussion

Together these studies demonstrated that CS could be desulfated without modification of the original CS chemical structure, CS and chondroitin materials could be crosslinked to form hydrogels, and CS-based hydrogels sequestered TGF- $\beta$ 1 in a sulfation-dependent manner. Chondroitin sulfate C was successfully desulfated by acidic methanol treatment after 7 days to yield chondroitin. DMMB assay, FTIR spectroscopy, and SAX-HPLC collectively indicated that chondroitin was ~98.5% nonsulfated after 7 days of chemical desulfation (Figures 5.3, 5.4a-b). FTIR and SAX-HPLC analysis also suggested that the remaining bonds and chemical structure of the CS disaccharides appeared to remain unmodified, and that desulfation was specific to removal of the sulfate groups. SAX-HPLC estimated that desulfation of CS resulted in a 2.3-fold reduction in negative charge density, due to removal of negatively charged sulfates (Figure 5.4a-b); however, due to the presence of carboxylates in the GAG backbone, chondroitin chains remained moderately negatively charged with approximately one



negatively charged group per disaccharide unit. These results present desulfated chondroitin materials as a highly controlled system to investigate of the role of sulfation and resulting negative charge density, without altering the chemical composition of the remaining polysaccharide GAG backbone.

Additionally, results from the DMMB assay suggested that CS-C, which is primarily sulfated at the 6-carbon, may be more susceptible to desulfation by acidic methanol treatment than CS-A, which is sulfated at the 4-carbon. After 3 days, CS-C underwent 80% desulfation, while CS-A only exhibited 54% desulfation, as measured by DMMB assay, suggesting that sulfation of the 6-carbon was more susceptible to cleavage by acidic methanol treatment. The 6-carbon of the N-acetylgalactosamine sugar of CS hangs away from the ring structure (Figure 5.1), while the 4-carbon is one of the members of the ring. This suggests that the ring structure may either limit accessibility to 4-O-sulfates for cleavage or may decrease the susceptibility of the 4-carbon to modification, slowing the rate of desulfation. Extended treatment with acid methanol after 7 days resulted in nearly complete desulfation of CS-C, suggesting that desulfation of CS was time-dependent with increasing desulfation over time (Figure 5.3a). These results suggested that a wide range of degrees of sulfation could be produced by increasing or decreasing the treatment time up to 7 days. In conjugation with various chemical methods of oversulfation [25-26], a diverse assortment of sulfated materials, with varying degrees of charge, could be developed from a single GAG structure. While desulfated chondroitin remains moderately negatively charged due to the presence of carboxylates, desulfation by acid methanol treatment methylated these carboxylate groups, producing a neutrally charged methyl ester intermediate. While this intermediate



was subsequently demethylated in KOH to form chondroitin for the purposes of these studies in which only sulfation was altered, the methyl ester of chondroitin could potentially be used as a neutrally charged alternative for further control of charge to study electrostatic interactions with GAGs.

SEC-HPLC analysis, however, suggested that desulfated chondroitin chains may be of shorter average length than the starting CS chains (Figure 5.4c-d). Acidic methanol treatment for 7 days may result in some degradation or chain shortening in chondroitin materials, possibly due to hydrolysis at low pH. To accurately characterize the true molecular weight of these materials, however, HPLC analysis would require further investigation to examine the interaction of charged GAG chains with the chromatography columns. In these experiments, the molecular weights of CS and chondroitin, which possess varying degrees of charge, were quantified relative to standards of partially depolymerized heparin, which is even more negatively charged than CS; therefore, a more thorough investigation would be required in differing buffer compositions to minimize the differential interactions of negatively charged GAG chains with the columns and to accurately quantify the molecular masses of CS and chondroitin. If further analysis determines that the average chain length of chondroitin is, in fact, substantially shorter than CS, the reaction parameters of the acid methanol treatment may be altered to minimize degradation. While shorter reaction time in 0.5% v/v acetyl chloride resulted in incomplete desulfation after 3 days, altering the concentration of reagents along with reaction time may be able to limit potential degradation of chondroitin. Reaction in either lower concentrations of acetyl chloride for a longer time or higher concentrations for a shorter time may be able to minimize potential chain



shortening in chondroitin materials. Additionally, other chemical and enzymatic procedures have been developed to desulfate CS, and these techniques could be explored as alternative chemistries to prevent degradation of chondroitin chains [375-378].

Nevertheless, while molecular weight is important in material properties and growth factor interactions, methacrylation and crosslinking of CS-MA and Ch-MA materials, as performed in these studies, were expected to mitigate potential differences in molecular mass, as GAG chains were crosslinked together and became immobilized with a highly crosslinked polymer network. Methacrylation of CS and chondroitin with GMA permitted crosslinking of GAG materials by free radical initiation, and <sup>1</sup>H NMR analysis suggested that CS-MA and Ch-MA were similarly methacrylated, approximately once every  $\sim 4$  disaccharides on average (Figure 5.5), indicating that similar crosslinking density of the two GAGs may result in comparable materials. Crosslinked CS-MA and Ch-MA hydrogels, however, did demonstrate significant differences in swelling properties, as 50% CS-MA hydrogels swelled significantly more than 50% Ch-MA gels in PBS (Figure 5.7a). These differences in swelling may be attributed in part to differences in osmotic pressure within the hydrogel networks, resulting from the disparity in negative charge density. The high negative charge density in CS-MA hydrogels is expected to attract high concentrations of positively charged molecules, while repelling negatively charged molecules, thus increasing the osmolarity of the interstitial fluid and creating a Donnan effect [379]. Donnan osmotic pressure in these materials was estimated to be approximately 3.0-fold higher in 50% CS-MA hydrogels in a 0.15 M NaCl solution than in 50% Ch-MA, based the higher negative charge of CS GAGs [45, 380]. Therefore, for use as a swelling control for release studies and future cellular



experiments, PEG-based controls composed of 60% PEG-DA/40% OPF were designed to possess similar swelling properties to 50% Ch-MA hydrogels, but in the absence of charged GAGs.

Chemical modification though desulfation of CS and subsequent methacrylation of CS and chondroitin did not appear to inhibit the ability of chondroitinase enzyme to degrade these products in solution (Figure 5.6). Additionally, crosslinked CS and Ch hydrogels completely degraded in chondroitinase enzyme, indicating that crosslinking of CS-MA and Ch-MA also did not prevent degradation of the hydrogel networks by chondroitinase. Enzymatic degradation of these biomaterials is especially important to permit interaction of the naturally-derived GAG matrix with surrounding cells. Cellsecreted chondroitinase enzyme could facilitate cell-mediated degradation of CS-MA and Ch-MA materials, and enzymatic degradation may permit localized remodeling of GAG matrix.

Desulfation of CS was found to alter the ability of CS to both retain and release the chondrogenic growth factor TGF- $\beta$ 1 *in vitro*. CS-MA and Ch-MA hydrogels were loaded with TGF- $\beta$ 1 and release was monitored over 7 days. In these studies, 50% CS-MA materials were found to largely retain TGF- $\beta$ 1, while decreasing the CS content to 10% CS-MA resulted in significantly greater release (Figure 5.7b). Additionally 50% Ch-MA, which maintained a constant GAG content compared to 50% CS-MA but with a reduction in sulfation and negative charge, showed significantly more release than 50% CS-MA, suggesting that sulfation of CS plays a fundamental role in the retention of growth factors. These differences in growth factor retention were observed in spite of differences in swelling between CS-MA and Ch-MA hydrogels. CS-MA hydrogels



swelled significantly more than Ch-MA gels (Figure 5.7a); therefore, the estimated mesh size of CS-MA materials would be expected to be higher than Ch-MA [380]. While one would expect an increase in mesh size to facilitate greater growth factor release from CS-MA hydrogels, Ch-MA materials with a smaller mesh size experienced significantly greater release of TGF- $\beta$ 1 than CS-MA. These results suggested that TGF- $\beta$ 1 was retained via electrostatic interactions with negatively charged sulfates in CS, rather than simply physical entrapment within the hydrogel network, and that sulfation can be carefully controlled within this system to alter the release kinetics of TGF- $\beta$ 1 from CS-based materials.

In particular, sulfated and desulfated CS-based materials may be promising tools to deliver TGF- $\beta$ 1 for cartilage regeneration. CS proteoglycans are expressed during early cartilage development [14-15], and sulfated GAGs have been shown to potentiate TGF- $\beta$ 1 signaling [274, 278]. TGF- $\beta$ 1 plays a critical role in promoting chondrogenesis and is positively charged at physiological pH, facilitating electrostatic interaction with sulfated GAGs [6, 24]. In response to TGF- $\beta$ 1 signaling, cartilage may also regulate subsequent CS production and sulfation patterning, suggesting that a potential feedback loop may exist in which sulfated GAGs regulate growth factor signaling [230, 279].

In addition to examining release from CS-MA and Ch-MA materials, the ability of CS-MA and Ch-MA hydrogel to "pull-down" soluble TGF- $\beta$ 1 out of solution was explored in CS-MA and Ch-MA hydrogels. Sequestration and retention appear to be more analogous to *in vivo* environments, in which proteoglycans are believed to sequester growth factors and possibly direct these signals for tissue patterning and differentiation [220, 230]. This system is also comparable to culture of GAG-based constructs in



chondrogenic media, in which hydrogels may be able to sequester TGF- $\beta$ 1 to promote signaling and differentiation. Pull-down was examined using hydrogels in which the total GAG content was maintained at 50% while the balance of CS-MA and Ch-MA was varied proportionally to alter the degree of sulfation (Figure 5.8a). Staining with DMMB demonstrated CS was retained within the hydrogel scaffolds after 1 week and increasing CS content was visible from 0% CS-MA (with 50% Ch-MA) up to 50% CS-MA (with 0% Ch-MA) by increased intensity of DMMB staining. Pull-down experiments in these gels demonstrated that 50% CS-MA hydrogels sequestered soluble TGF- $\beta$ 1, "pulling" it out of solution and sequestering TGF- $\beta$ 1 within the hydrogel network in 1% BSA (Figure 5.8). Decreasing the sulfation of the GAG matrix reduced the observed interaction with TGF- $\beta$ 1 in a concentration-dependent manner in 10% CS-MA/40% Ch-MA hydrogels and in 50% Ch-MA hydrogels, indicating that sulfation and charge play important roles in facilitating depletion of TGF- $\beta$ 1. Interestingly, increasing the ionic strength of the buffer with an additional 0.5 M NaCl significantly inhibited pull-down of TGF- $\beta$ 1 by 50% Ch-MA, while pull-down by 50% CS-MA did not appear to decrease. The moderate negative charge of chondroitin may retain a weaker electrostatic interaction with TGF- $\beta$ 1 that was inhibited in 0.5 M NaCl; however, 0.5 M NaCl was not sufficient ionic strength to shield TGF- $\beta$ 1 interactions with 50% CS-MA hydrogels, suggesting that CS-MA may possess a stronger ability to bind TGF- $\beta$ 1. Similarly, 10 mg/mL soluble CS decreased depletion in 50% Ch-MA, but was unable to competitively inhibit sequestration in 50% CS-MA gels. This suggested that TGF- $\beta$ 1 may preferentially bind crosslinked CS-MA hydrogels over soluble CS chains in solution.



In related studies, Hintze et al. demonstrated that chemically oversulfated CS and hyaluronan were able to sequester TGF- $\beta$ 1 in a sulfation-dependent manner when covalently conjugated onto 2D surfaces, as determined by ELISA and surface plasmon resonance (SPR), and incubation with 0.3 M NaCl inhibited electrostatic interactions with TGF- $\beta$ 1[25]. Similarly, Lyon et al. found that TGF- $\beta$ 1 bound heparin by affinity chromatography, and 0.5 M NaCl was sufficient to elute TGF- $\beta$ 1 from the heparinagarose columns [274]. This was consistent with a number of studies that have reported shielding of a variety of electrostatic protein interactions with sulfated GAGs in 0.5 M NaCl [381-383]; however, binding of a number of stronger protein-GAG interactions have required concentrations up to 1.5 M NaCl [384-388], suggesting that 50% CS-MA hydrogels may possess a relatively strong electrostatic interaction with TGF- $\beta$ 1 in this system and that greater ionic strength may be required to shield these interactions. An additional study by Hintze el al. showed that 10 mg/mL of soluble GAGs, including CS, was able to competitively inhibit binding of BMP-4 to surfaces conjugated with oversulfated hyaluronan by ELISA [273]. It is important to note that these other studies tested binding to GAG-conjugated 2D surfaces or affinity columns, while the work in this dissertation represents the first time that TGF- $\beta$ 1 sequestration has been examined in crosslinked GAG hydrogels. Crosslinked hydrogels may immobilize GAG chains in close proximity to alter their ability to sequester TGF- $\beta$ 1.

TGF- $\beta$ 1 is a growth factor that plays a critical role in promoting chondrogenesis and is positively charged at physiological pH, facilitating electrostatic interaction with sulfated GAGs. Heparin has been found to electrostatically interact with TGF- $\beta$ 1 and TGF- $\beta$ 2, but not the TGF- $\beta$ 3 isoform [274]. TGF- $\beta$ 1 and TGF- $\beta$ 2 are reported to have



isoelectric points (pI) of approximately 9.5 and 8.5, respectively [275-276], while TGF- $\beta$ 3 has a lower pI of 6.8, suggesting that it is actually negatively charged at physiological pH and unable to electrostatically complex with sulfated GAGs [275]. Highly sulfated GAGs were also capable of potentiating TGF- $\beta$ 1 activity in mink lung epithelial cells, suggesting that GAGs may also have a synergistic effect in TGF- $\beta$ 1 signaling. It was hypothesized that TGF- $\beta$ 1 interacts with sulfated GAGs via basic arginine and lysine residues at positions 25, 26, 31, and 37 and a histidine at 34 in TGF- $\beta$ 1, along with Arg/Lys at position 94 [272, 274]. In the TGF-B1 dimer, these two sites could be potentially be engaged by a single GAG chain approximately 60 Å apart or by two separate GAG chains. The proposed binding site was also in a similar location, at the tips of the  $\beta$ -strand loops, to where TGF- $\beta$  binds its receptors; however, little competition has been reported and in many cases sulfate GAGs appeared to potentiate the signaling effects [272]. These studies suggested that GAG-based materials may have significant application in controlling interactions with TGF-\beta1 to promote chondrogenic differentiation of MSCs.

Overall, desulfated chondroitin hydrogels provide a highly controlled system to examine the interactions between charged GAGs and growth factors, and these materials may have tremendous potential to promote chondrogenic differentiation through sequestration of growth factors.

#### 5.5 Conclusions

These studies present desulfated chondroitin as a potential tool to further investigate the role of sulfation in GAG interactions with growth factors. Chondroitin



sulfate was chemically desulfated by acidic methanol treatment, and characterization supported significant desulfation with little modification of the remaining CS chemical structure. Desulfation also resulted in a 2.3-fold decrease in negative charge density in chondroitin materials. Methacrylation of CS and chondroitin allowed the formation of CS-MA and Ch-MA hydrogels, which remained degradable by chondroitinase enzyme for potential cell-mediated degradation. 50% CS-MA hydrogels were able to sequester greater amounts of TGF-B1 than desulfated 50% Ch-MA hydrogels, as shown by retaining TGF- $\beta$ 1 from release and through TGF- $\beta$ 1 pull-down from solution, suggesting that sulfates play an important role in facilitating the electrostatic interactions between growth factors and CS. These results present desulfated chondroitin materials as a valuable tool to alter GAG sulfation in a highly controlled manner as a promising strategy to control binding and interaction with positively charged growth factors, particularly the chondrogenic growth factor TGF-\beta1. Sulfated and desulfated CS-based materials may present a promising platform to modulate growth factor interactions to control release or sequestration for chondrogenic differentiation of MSCs and development of novel tissue engineering therapies for cartilage regeneration and repair.



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# **CHAPTER 6**

# DESULFATED CHONDROITIN HYDROGELS UPREGULATED GENE EXPRESSION OF CARTILAGINOUS MARKERS BY ENCAPSULATED HUMAN MESENCHYMAL STEM CELLS IN THE PRESENCE OF TGF-β1

#### 6.1 Introduction

As a load bearing tissue, damage to articular cartilage can be both debilitating and extremely painful. Nearly 466,000 arthroscopic procedures performed on the knee in an ambulatory (outpatient) setting were diagnosed with a tear of the medial or lateral cartilage or menisci in the United States in 2006 [2]. If severe cartilage damage is allowed to persist without appropriate repair, the tissue may degenerate further and progress into development of osteoarthritis. In 2005, an estimated 27 million adults in the US had clinical osteoarthritis with indirect costs totaling approximately \$89 billion per year [3-4]. Due to cartilage's low capacity for healing, novel tissue engineering therapies seek to promote repair and regeneration of cartilaginous tissues. Mesenchymal stem cells (MSCs) are multipotent progenitor cells that are capable of differentiating into cartilaginous tissues, making them a promising cell source for tissue repair [6]; however, difficulty recapitulating the complex extracellular matrix (ECM) composition and organization of native cartilage remains a significant challenge. Recent research has suggested that cartilaginous ECM molecules may play a central role in directing differentiation of MSCs down a chondrogenic lineage.



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Chondroitin sulfate (CS) is a cartilaginous glycosaminoglycan (GAG) that appears to play an important role in chondrogenesis *in vivo*. CS-containing proteoglycans versican and perlecan regulate mesenchymal condensation during cartilage development [14, 218-219, 223-224], and several enzymes involved in CS initiation, elongation, and sulfation are required for proper skeletal development and patterning [16-18]. Recent studies have investigated the ability of CS-modified biomaterials to promote chondrogenic differentiation *in vitro* as well. Culture in CS-containing hydrogels upregulated expression and production of cartilaginous ECM by encapsulated goat and mouse MSCs [8-9], suggesting that CS may play a particularly important role in directing stem cell differentiation down a chondrogenic lineage; however, relatively few studies have specifically investigated the effect of CS on chondrogenic differentiation of human MSCs (as opposed to other mammalian species) [251-252].

In cartilaginous tissues, sulfated GAGs play essential roles in maintaining cartilage function. Specifically, the high degree of sulfation of CS carries a highly negative fixed charge density that facilitates a variety of interactions with cartilaginous ECM, signaling molecules, and interstitial fluid. In Chapter 5 of this dissertation, nonsulfated chondroitin materials were characterized as a potential system to control material interactions with growth factors, including the chondrogenic growth factor TGF- $\beta$ 1; however, the effects of sulfation of CS in chondrogenic differentiation of MSCs have not been investigated. To examine the chondrogenic response to sulfated and nonsulfated GAG materials, human MSCs were encapsulated in CS- and chondroitin-containing hydrogels and cultured in the presence of chondrogenic medium containing TGF- $\beta$ 1 over the course of 6 weeks *in vitro*. Cell viability and total DNA content were monitored over



time, and gene expression and ECM production of encapsulated MSCs were determined as measures of chondrogenic differentiation.

# 6.2 Materials and Methods

#### **6.2.1 Synthesis of Materials**

#### 6.2.1.1 Desulfation of Chondroitin Sulfate

Chondroitin sulfate was desulfated using an acidic methanol treatment for up to 7 days per established protocols [365-366]. Chondroitin sulfate C (primarily chondroitin-6-sulfate, Wako Chemicals USA, Richmond, VA) was stirred at 5.0 mg/mL in methanol (VWR, Radnor, PA) containing 0.5% v/v acetyl chloride (Acros Organics, Geel, Belgium). CS was centrifuged and acidic methanol was replaced on days 1, 3, and 7 to produce a methyl ester of chondroitin. The product was then dissolved in 20 mL distilled, deionized water (ddH<sub>2</sub>O) per gram of starting CS before precipitation in an excess of ethanol. The methyl ester of chondroitin was washed in ethanol and ethyl ether (Fisher, Waltham, MA), vacuum dried at <5 mmHg, and stored at 4°C.

Methyl ester of chondroitin was demethylated at 25 mg/mL in 0.1 M potassium hydroxide (KOH, Fisher) for 24 hours to produce chondroitin. The chondroitin product was then neutralized in 4 mL of 100 mg/mL potassium acetate (Fisher) in 10% v/v acetic acid (VWR) per gram of starting product, and precipitated in an excess of ethanol. Chondroitin was washed in ethanol and ethyl ether, vacuum dried, and stored at 4°C until use. Removal of sulfate groups in chondroitin was confirmed by dimethylmethylene blue (DMMB) assay for sulfated GAGs [302].



# 6.2.1.2 Methacrylation of Chondroitin Sulfate and Chondroitin

CS and chondroitin were methacrylated with glycidyl methacrylate per established protocols [368]. CS or chondroitin was dissolved at 1% w/v in a 50:50 mixture of acetone (VWR) and ddH<sub>2</sub>O and allowed to stir at room temperature overnight. A 20-fold molar excess of triethylamine (TEA, Sigma-Aldrich, St. Louis, MO) per CS or chondroitin disaccharide was added to the solution, and a 20-fold molar excess of glycidyl methacrylate (GMA, Sigma-Aldrich) per disaccharide was added dropwise to the solution. The reaction was allowed to stir at room temperature for 24 hours to produce CS-methacrylate (CS-MA) and chondroitin-methacrylate (Ch-MA). The resulting products were dialyzed first in 50:50 acetone:water for 24 hours (1,000 Da MWCO), and then in distilled water (dH<sub>2</sub>O) for 2 days to remove unreacted reagents. The methacrylated products were lyophilized (Labconco, Kansas City, MO) for 4 days to produce a dry product, and stored at -20°C until use. Proton nuclear magnetic resonance (<sup>1</sup>H NMR) was utilized to determine the degree of the conjugation of the GMA groups to the CS and chondroitin chains.

## 6.2.1.3 PEG-DA Synthesis

Poly(ethylene glycol)-diacrylate (PEG-DA) polymers were synthesized according to established protocols [191]. 3,400 Da ( $M_n$ ) PEG (Sigma-Aldrich) was solubilized in distilled, anhydrous methylene chloride (MeCl, Fisher) to produce a 60% (m/v) solution. TEA was added to the dissolved PEG, and acryloyl chloride (AcCl, Sigma-Aldrich) was slowly added dropwise at 2:1 AcCl:PEG and 1:1 AcCl:TEA molar ratios to create a 40% w/v solution. The reaction was allowed to proceed at room temperature under nitrogen gas, and the reaction was left stirring overnight to ensure completion. 2 M potassium



carbonate (K<sub>2</sub>CO<sub>3</sub>, Fisher) was used at a 2:1 K<sub>2</sub>CO<sub>3</sub>:AcCl molar ratio to extract TEA to the aqueous phase. The organic phase was allowed to separate from the aqueous phase, and anhydrous magnesium sulfate (MgSO<sub>4</sub>, Fisher) was used to remove any remaining aqueous phase from the solution. PEG-DA was then precipitated in ethyl ether and filtered, followed by vacuum drying at <5 mmHg. Dry polymers were stored in sealed containers at -20°C until use. PEG-DA ( $M_n = 3760 \pm 50$ , PI = 1.1 ± 0.02) and PEG 3.4K ( $M_n = 3440 \pm 30$ , PI = 1.1 ± 0.001) were characterized via gel permeation chromatography as previously reported [174].

# 6.2.1.4 OPF Polymer Synthesis

Oligo(poly(ethylene glycol) fumarate) (OPF) 10K polymers were synthesized according to established protocols [367]. 10,000 Da ( $M_n$ ) PEG (Sigma-Aldrich) was azeotropically distilled in toluene (Fisher) and then combined with distilled, anhydrous MeCl to produce a 40% (v/v) solution. Fumaryl chloride (FuCl, Sigma-Aldrich) and TEA (Sigma-Aldrich) were slowly added dropwise over a period of 2 hours at 0.9 FuCl:PEG and 1:2 FuCl:TEA molar ratios. During this time the reaction was allowed to proceed at 0°C under nitrogen gas, after which the reaction was left stirring for 72 hours under nitrogen at room temperature to ensure reaction completion. After rotovaporing (Buchi, Flawil, Switzerland) to remove excess MeCl, the resulting polymer was recrystallized twice with ethyl acetate (Fisher) and washed three times in ethyl ether (Fisher), followed by vacuum drying at <5 mmHg. Dry polymers were stored in sealed containers at -20°C before use. OPF 10K ( $M_n = 28,100 \pm 760$ , PI = 5.2 ± 0.4) and PEG 10K ( $M_n = 12,900 \pm 210$ ; PI = 1.1 ± 0.002) were characterized via gel permeation chromatography as previously reported [174].



# 6.2.2 Encapsulation of Human MSC

Human MSCs were obtained from the Texas A&M Health Science Center College of Medicine Institute for Regenerative Medicine at Scott & White (Temple, TX) at passage 1. Cells were seeded at 50 cells/cm<sup>2</sup> following recommended protocols, in growth medium containing  $\alpha$ -MEM (Mediatech, Manassas, VA) with 16.3% fetal bovine serum (FBS, Atlanta Biologicals, Lawrenceville, GA), 1% antibiotic/antimycotic (Mediatech), and 2 mM L-glutamine (Mediatech). Following expansion, cells were frozen at passage 2 in liquid nitrogen until further use. For these studies, cells from three separate donors were thawed, expanded separately, and combined prior to encapsulation at passage 3.

For cell encapsulation, human MSCs were incorporated at a final cell concentration of  $20 \times 10^6$  cells/mL into macromer solutions containing a 1:1 ratio of OPF 10K:PEG-DA and 50% CS-MA or 50% Ch-MA by dry mass. PEG controls containing 60% PEG-DA:40% OPF, which were designed to swell similarly to 50% Ch-MA as described in Chapter 5, were used as swelling controls. Dispersed MSCs were crosslinked in hydrogels using 6 mm diameter, 1 mm deep cylindrical molds with 0.018 M ammonium persulfate (APS, Sigma-Aldrich) and tetramethylethylenediamine (TEMED, Sigma-Aldrich) for 10 minutes at 37 °C. Hydrogels were cultured for 6 weeks at 37°C and 5% CO<sub>2</sub> in chondrogenic medium composed of high glucose DMEM (Mediatech) containing 1% ITS+ culture supplement (Becton, Dickenson, Franklin Lakes, NJ), 1% nonessential amino acids (NEAA, Mediatech), 50  $\mu$ g/mL ascorbate-2-phosphate (Sigma-Aldrich), and 1% antibiotic/antimycotic (Mediatech). Chondrogenic medium was also supplemented with 10 ng/mL TGF- $\beta$ 1 (Peprotech, Rocky Hill, NJ) and 100 nM



dexamethasone (Sigma-Aldrich), and culture medium was replaced every 2 days throughout the course of the study.

## 6.2.3 Analysis of MSC Response in CS-MA and Ch-MA Hydrogels

# 6.2.3.1 LIVE/DEAD Staining for Viability

Viability of human MSCs encapsulated in 50% CS-MA and 50% Ch-MA hydrogels was observed by LIVE/DEAD staining over 6 weeks. In this stain, calcein is cleaved within the cytosol of viable cells, fluorescing green (ex/em: 494/517 nm), while ethidium homodimer-1 is able to enter the ruptured cell membranes of nonviable cells, binding nuclear DNA and fluorescing red (ex/em: 528/617 nm). On days 1, 21, and 42, after rinsing samples of excess media in PBS with 100  $\mu$ g/mL CaCl<sub>2</sub> and 47  $\mu$ g/mL MgCl<sub>2</sub> (Invitrogen, Carlsbad, CA), hydrogels were stained for 60 minutes in LIVE/DEAD stain containing 1  $\mu$ M calcein and ethidium homodimer-1 (Invitrogen). Viability within all hydrogel formulations was imaged in PBS via confocal microscopy (Carl Zeiss LSM 510, Oberkochen, Germany). Images were captured every 10  $\mu$ m through the entire depth of the hydrogel from three separate regions in each sample (n=4).

# 6.2.3.2 PicoGreen DNA assay

Hydrogels were analyzed by PicoGreen assay for DNA content on days 1, 21, and 42. The PicoGreen assay uses a fluorescent dye that binds to double stranded DNA to quantify DNA content. After rinsing in PBS, samples were massed and wet mass was recorded. Hydrogels were then homogenized with pestle grinders and mixed with 500  $\mu$ L of distilled water before being frozen at -80°C until analysis. Cells were lysed through a series of freeze/thaw cycles and sonication. PicoGreen assay (Invitrogen) was used to



evaluate the total DNA content in each sample, according to established protocols [303]. Fluorescence was read at excitation 485 nm, emission 525 nm (Molecular Devices SpectraMax M2e, Sunnyvale, CA), and DNA content was determined using a standard curve of DNA. Within each hydrogel formulation, DNA content of each gel was normalized to wet mass to correct for small differences in gel size (n=4).

# 6.2.3.3 Reverse transcription polymerase chain reaction

Gene expression of encapsulated MSCs was analyzed after 1, 21, and 42 days by reverse transcription polymerase chain reaction (RT-PCR). After rinsing in PBS, RNA was extracted from samples using a QIAshredder tissue homogenizer and RNeasy kit with DNase I digestion (Qiagen, Hilden, Germany). Reverse transcription was performed using SuperScript III Reverse Transcriptase (Invitrogen) with Oligo(dT)<sub>15</sub> primers (Promega, Madison, WI) and nucleotides (Promega). Custom designed primers (Invitrogen) specific to human mRNA for collagen II, aggrecan, and SOX9 (chondrocytic markers), collagen X (hypertrophic chondrocyte marker), collagen I (fibroblastic marker), osteocalcin (osteoblastic marker), myoD (myofibroblastic marker), and peroxisome proliferator-activated receptor  $\gamma 2$  (PPAR- $\gamma 2$ ; adipocytic marker) are shown in Table 5.2. Quantitative PCR amplification for each gene target was performed on a StepOnePlus System (Applied Biosystems, Carlsbad, CA) with SYBR Green master mix (Applied Biosystems). To determine fold regulation over PEG control hydrogels on day 1, the raw fluorescence data was processed using LinRegPCR (v12.11; http://www.hartfaalcentrum.nl) [389] with glyceraldehyde-3-phosphate dehydrogenase (GAPDH) as an endogenous control (n=6).



| Target             | Primer Sequences (5'-3')    | GenBank      |
|--------------------|-----------------------------|--------------|
| Collagen II        | ACCCCAATCCAGCAAACGTT        | NM_001844    |
| (α1)               | ATCTGGACGTTGGCAGTGTTG       |              |
| Aggrecan           | ACAGCTGGGGACATTAGTGG        | NM_001135    |
|                    | GTGGAATGCAGAGGTGGTTT        |              |
| SOX9               | GCGGAGGAAGTCGGTGAAGAACGGGCA | NM_000346    |
|                    | TGTGAGCGGGTGATGGGCGGG       |              |
| Collagen X<br>(α1) | GGCCCAGCAGGAGCAAAGGG        | NM_000493    |
|                    | GTGGCCCGGTGGGTCCATTG        |              |
| Collagen I<br>(α2) | GAAAACATCCCAGCCAAGAA        | NM_000089    |
|                    | GCCAGTCTCCTCATCCATGT        |              |
| Osteocalcin        | GTGCAGAGTCCAGCAAAGGT        | NIM 100173   |
| (BGLAP)            | AGCAGAGCGACACCCTAGAC        | 11111_199173 |
| MyoD               | GTCGAGCCTAGACTGCCTGT        | NM_002478    |
|                    | GTATATCGGGTTGGGGTTCG        |              |
| PPAR-γ2            | TCCATGCTGTTATGGGTGAA        | NM_015869    |
|                    | GGGAGTGGTCTTCCATTACG        |              |
| GAPDH              | GAGTCAACGGATTTGGTCGT        | NM_002046    |
|                    | TTGATTTTGGAGGGATCTCG        |              |

**Table 6.1.** Human primer sequences for quantitative polymerase chain reaction

# 6.2.3.4 Histological staining

ECM production by encapsulated MSCs was determined by immunostaining on days 1, 21, and 42. After rinsing in PBS, samples were placed into a solution of 5% w/v sucrose (EMD, Darmstadt, Germany) in PBS under vacuum (-25 inHg). The sucrose concentration was gradually increased over the course of 2 hours to 15% sucrose. Next, gels were subjected to increasing concentrations of 20% sucrose:OCT (VWR) over the course of 4 hours until samples were infiltrated with a 1:2 volume ratio of 20% sucrose:OCT. After vacuum infiltration overnight, samples were embedded in 1:2 20% sucrose:OCT solution by gentle freezing in liquid nitrogen and stored at -80°C. Infiltrated hydrogels were cryosectioned at 20 µm thickness (Thermo Scientific, Cryostar NX70).

Sections were fixed in acetone, and OCT was rinsed in PBS. For aggrecan and collagen X staining, samples were deglycosylated with 30  $\mu$ l of 0.75 U/ml chondroitinase ABC (Sigma-Aldrich) for 1.5 hours. Samples were blocked with Image-iT FX signal enhancer (Invitrogen). For primary antibody binding, sections were incubated overnight



at 4°C in monoclonal mouse anti-human collagen I, collagen II, aggrecan (Abcam), or collagen X (Sigma). Sections were then incubated for 30 minutes with highly crossadsorbed Alexa Fluor 488-conjugated goat polyclonal anti-mouse immunoglobulin G (IgG, Molecular Probes, Carlsbad, CA) or IgM (Molecular Probes) for collagen X, and counterstained with 0.1  $\mu$ g/mL 4',6-diamidino-2-phenylindole (DAPI, Anaspec, Fremont, CA). Negative controls were stained as described, but using a monoclonal mouse IgG1 isotype control (Abcam) with no known reactivity with human antigens as the primary antibody at 10  $\mu$ g/mL. Histological sections were imaged using an epifluorescence microscope with a 20X magnification objective (Nikon Eclipse 80i, Tokyo, Japan) (n=2).

# **6.2.4 Statistical Analysis**

All values were reported as mean  $\pm$  standard deviation. For statistical analysis, PCR amplification data for each gene were first transformed using a Box-Cox transformation to obtain a normal distribution for analysis [390]. A two-factor analysis of variance (ANOVA) was used to determine statistical significance of groups, and Tukey's *post hoc* multiple comparison test with significance set at *p*≤0.05 indicated significance between individual samples. For DNA and gene expression analysis, the factors were hydrogel type and time. Statistical analysis was carried out using Minitab (v15.1, State College, PA).

## 6.3 Results

#### 6.3.1 Viability in CS-MA and Ch-MA hydrogels during 6 weeks of culture

LIVE/DEAD staining of human MSCs in 50% CS-MA and 50% Ch-MA hydrogels indicated that visible MSCs appeared mostly viable over 6 weeks of culture



(Figure 6.1a). Cells remained dispersed evenly throughout the hydrogel scaffolds with a spherical shape, and no cell aggregation or spreading was observed. Total DNA content, as a measure of cell number, suggested that cellularity decreased over time in all hydrogel formulations (Figure 6.1b). DNA content significantly decreased by approximately half from day 1 to day 21 in all hydrogels; however, no significant differences were observed in DNA content between day 21 and day 42. Additionally, DNA content was significantly greater in 50% Ch-MA hydrogels than 50% CS-MA or PEG control hydrogels on days 1 and 21, suggesting that 50% Ch-MA hydrogels contained a higher cellularity.

# 6.3.2 Upregulation of gene expression for chondrocytic markers by MSCs encapsulated in 50% Ch-MA hydrogels

Human MSCs encapsulated in nonsulfated 50% Ch-MA hydrogels significantly upregulated gene expression of the cartilaginous ECM molecules collagen II and aggrecan on days 21 and 42, over 50% CS-MA hydrogels (Figure 6.2). MSCs in 50% Ch-MA expressed 83,100  $\pm$  38,800 fold upregulation of collagen II expression, compared to only 28  $\pm$  174 in 50% CS-MA and 2,650  $\pm$  6,410 fold in PEG controls (Figure 6.2a). Similarly, aggrecan expression in Ch-MA gels was upregulated 76.0  $\pm$  37.3 fold on day 42, compared to 4.9  $\pm$  11.5 in CS-MA and 59.6  $\pm$  25.5 in PEG (Figure 6.2c). Cartilaginous transcription factor SOX9 experienced slight upregulation on day 42 in 50% Ch-MA with 2.0  $\pm$  1.5 fold regulation, while CS-MA only expressed 0.19  $\pm$  0.11 fold regulation (Figure 6.2d). Collagen X, an ECM marker of hypertrophic chondrocytes, was also significantly upregulated in Ch-MA hydrogels over CS-MA and PEG controls



on days 21 and 42 (Figure 6.2e). MSCs in 50% Ch-MA upregulated collagen X by 202  $\pm$  93 fold on day 42, compared to 11.0  $\pm$  11.0 in CS-MA and 31.8  $\pm$  14.1 in PEG controls.



Figure 6.1. Viability and cellularity of human MSCs encapsulated in 50% CS-MA and 50% Ch-MA hydrogels. (a) Human MSCs appeared largely viable in all hydrogel formulations over 42 days of culture. Cells remained dispersed with a rounded morphology. Scale bars = 100  $\mu$ m. (b) DNA content, as a measure of cellularity, decreased over time with a significant decrease observed between day 1 and day 21 in all hydrogel formulations. Additionally, greater DNA content was measured in 50% Ch-MA hydrogels on days 1 and 21, compared to 50% CS-MA and PEG controls. \* indicates significantly less DNA than day 1 ( $p \le 0.05$ ). # indicates significantly less DNA than 50% Ch-MA hydrogels at the same time point ( $p \le 0.05$ ).

Markers for other tissues including fibroblastic marker collagen I, osteoblastic marker osteocalcin, myofibroblastic marker MyoD, and adipocytic marker PPAR- $\gamma$ 2 were also examined. Collagen I did not exhibit downregulation over the course of this study (Figure 6.3a). MSCs in PEG hydrogels actually exhibited 23.3 ± 15.2 fold upregulation of collagen I by day 42, and Ch-MA gels expressed 6.2 ± 4.2 fold regulation. The ratio of collagen II:collagen I relative expression suggested that MSCs in



50% Ch-MA hydrogels exhibited nearly equal expression of collagen II and collagen I after 42 days with a ratio of  $1.03 \pm 1.20$ , indicating significant upregulation from only  $6.9 \times 10^{-5} \pm 2.6 \times 10^{-3}$  on day 1 (Figure 6.2b). Osteocalcin displayed slight upregulation in encapsulated MSCs over time, but did not exhibit significant differences across hydrogel types (Figure 6.3b). MyoD expression demonstrated high variability and no significant trends (Figure 6.3c), while PPAR- $\gamma$ 2 did not amplify within 40 PCR cycles, indicating low PPAR- $\gamma$ 2 expression at all time points.



Figure 6.2. Gene expression of chondrocytic markers by MSCs in 50% CS-MA and 50% Ch-MA hydrogels. MSCs in 50% Ch-MA hydrogels significantly upregulated gene expression of (a) collagen II, (b) ratio of collagen II:collagen I relative expression, (c) aggrecan, and (d) SOX9 after 42 days, compared to 50% CS-MA. (e) Hypertrophic chondrocyte marker collagen X was also significantly upregulated in 50% Ch-MA gels on day 21 and 42 over 50% CS-MA and PEG controls. \* indicates significantly greater than 50% CS-MA at same time point (*p*≤0.05). # indicates significantly greater than PEG controls at same time point (*p*≤0.05).





Figure 6.3. Gene expression of negative tissue markers by MSCs in 50% CS-MA and 50% Ch-MA hydrogels. While (a) collagen I, (b) osteocalcin, and (c) MyoD displayed some gene regulation in encapsulated MSCs over time, few differences were observed across hydrogel types. \* indicates significantly greater than 50% CS-MA at same time point ( $p \le 0.05$ ). + indicates significantly greater than 50% Ch-MA at same time point ( $p \le 0.05$ ).

# 6.3.3 ECM deposition in GAG-containing hydrogels

Immunostaining for ECM production demonstrated that greater ECM production overall was observed in GAG-containing hydrogels over PEG control hydrogels, which showed relatively little staining for ECM (Figure 6.4). While some accumulation of cartilaginous ECM collagen II and aggrecan was observed pericellularly over 42 days, clear differences were not apparent between Ch-MA and CS-MA formulations (Figure 6.4a-b). Staining for collagen X, however, exhibited noticeably greater staining in 50% CS-MA hydrogels than Ch-MA gels (Figure 6.4c). Deposition of collagen I appeared to persist over time but few differences were observed between CS-MA and Ch-MA hydrogels (Figure 6.4d).





Figure 6.4. Immunostaining for ECM deposition by MSCs in 50% CS-MA and 50% Ch-MA hydrogels. (a) Collagen II and (b) aggrecan were produced in small quantities by MSCs in CS-MA and Ch-MA hydrogels, compared to relatively little deposition in PEG gels. Little difference was observed between 50% CS-MA and 50% Ch-MA. (c) Greater staining for collagen X was observed in 50% CS-MA gels. (d) Collagen I was produced by MSCs, but few differences were observed across gel types. Cell nuclei = blue. Scale bars = 100 μm.



#### 6.4 Discussion

These studies demonstrated that viable MSCs remained dispersed throughout GAG-based hydrogels after 6 weeks of culture *in vitro*, despite significant decreases in DNA content over time (Figure 6.1), and that nonsulfated chondroitin materials upregulated gene expression of chondrogenic markers in human MSCs, compared to CS materials, when cultured in chondrogenic medium (Figure 6.2). MSCs encapsulated in 50% Ch-MA hydrogels exhibited significantly greater gene expression of collagen II, aggrecan, and SOX9 than MSCs in 50% CS-MA gels after 42 days of culture in chondrogenic medium containing TGF-\beta1, suggesting that nonsulfated chondroitin materials may promote greater chondrogenic differentiation over CS. As described in Chapter 5, 50% Ch-MA hydrogels swelled significantly less than 50% CS-MA; therefore, in these experiments, PEG hydrogels that swelled similarly to 50% Ch-MA hydrogels were used as swelling controls. MSCs have been shown to regulate differentiation in response to both substrate stiffness and pore size in vitro [160, 162, 391-392]; therefore, PEG swelling controls were used as approximate controls for material stiffness and porosity/mesh size in the absence of charged polysaccharides [45, 380]. In this experiment, MSCs in uncharged PEG control hydrogels expressed significant upregulation of collagen II, aggrecan, and SOX9 gene expression over 50% CS-MA hydrogels after 6 weeks; however, gene expression of chondrogenic markers remained less than that observed in MSCs in Ch-MA materials, suggesting that upregulation in 50% Ch-MA hydrogels was not solely due to differences in swelling properties.

An investigation of CS-MA and Ch-MA hydrogels, compared to PEG-based swelling controls, at earlier time points (1, 7, 14, 21 days) and lower cell density  $(10x10^6)$ 



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cells/mL) demonstrated similar trends of upregulation of gene expression by MSCs in Ch-MA hydrogels in the presence of chondrogenic medium; however, in the absence of TGF-\beta1 and dexamethasone, chondroitin materials did not exhibit significant upregulation of gene expression for cartilaginous markers (Figure A.2). These results indicated that nonsulfated chondroitin materials alone were not sufficient to upregulate gene expression of cartilaginous markers by MSCs, and that upregulation of chondrogenic markers may be dependent on interactions between the chondroitin material and chondrogenic cues from exogenously supplemented TGF- $\beta$ 1. Interestingly, CS materials did significantly upregulate collagen II gene expression, in the absence of TGF- $\beta$ 1 and dexamethasone on day 21, while chondroitin materials did not (Figure A.2), suggesting that CS materials may possess the unique ability to promote chondrogenic differentiation in the absence of exogenous soluble cues. While this presents an interesting potential utility of CS-based materials, due to inconsistency within these CS samples and the greater overall response observed in chondroitin materials in the presence of chondrogenic medium, these trends have not been examined in further detail. Together, these contrasting responses in the presence and absence of chondrogenic media suggested that MSCs encapsulated in CS and chondroitin materials differentially regulate their response to soluble chondrogenic cues.

The ability of CS and chondroitin materials to differentially "pull-down" soluble TGF- $\beta$ 1 out of solution, as discussed in Chapter 5, is analogous to culture in chondrogenic medium, as GAG scaffolds may electrostatically bind and retain TGF- $\beta$ 1 within their networks. The pull-down experiments presented in Chapter 5 demonstrated that 50% CS-MA depleted greater amounts of TGF- $\beta$ 1 from solution than 50% Ch-MA in



a sulfation-dependent manner, suggesting that CS-based materials may sequester TGF- $\beta$ 1 from chondrogenic medium, thus increasing the local concentration of growth factor in close proximity to cells or modulating signaling efficiency within GAG-based scaffolds. Although enhanced upregulation of chondrogenic markers in Ch-MA materials over CS-MA hydrogels may appear to contradict this hypothesis, the high negative charge density of CS in these materials could potentially decrease TGF-β1 activity or inhibit transport within the hydrogel network; therefore, removal of sulfate groups may promote MSC differentiation via enhanced growth factor signaling. In studies by Seto et al., similar PEG hydrogels containing 10% heparin, a highly sulfated GAG, exhibited restricted diffusion of a positively charged model protein into the hydrogel network, and encapsulated MSCs co-cultured with osteoblasts demonstrated preferential mineralization at the surface of these hydrogels, potentially suggesting sequestration and limited transport of osteoblast-secreted soluble factors into the hydrogel scaffolds [393]. Noticeable spatial differences in ECM production, as observed by immunostaining, were not apparent in the experiments presented here; however, overall ECM deposition appeared to remain relatively low throughout all hydrogel types. The highly charged CS matrix may also prevent transport of other cell-secreted signals within the hydrogel, effectively inhibiting intercellular communication, which plays important roles in supporting chondrogenic differentiation and maintaining a chondrocytic phenotype [24, 38, 394-395]. Nevertheless, it is important to note that chondroitin remains moderately negatively charged at physiological pH despite the absence of sulfates, due to the presence of carboxylates in the GAG structure, and may retain electrostatic interactions with TGF- $\beta$ 1, albeit to a lesser degree than CS. These results suggest that decreasing the



degree of sulfation in GAG-based hydrogels may be better suited for promoting chondrogenic differentiation of MSCs than highly sulfated CS, and that differential sequestration or altered presentation of TGF- $\beta$ 1 to encapsulated MSCs may contribute to the observed differences in the gene expression between CS-MA and Ch-MA materials.

TGF- $\beta$ 1 pull-down experiments, as presented in Chapter 5, also found that PEG control hydrogels demonstrated similar degrees of TGF-β1 depletion to 50% Ch-MA gels in 1% BSA solutions, even though retention was not influenced by increased ionic strength in the presence of 0.5 M NaCl. These results suggested that PEG materials may be trapping similar amounts of TGF-B1 from solution as 50% Ch-MA, although the observed interaction with neutrally charged PEG-based materials is likely by nonelectrostatic means. Nevertheless, 50% Ch-MA hydrogels enhanced gene expression of chondrogenic markers over PEG control hydrogels, despite similarities in TGF-<sup>β1</sup> adsorption and swelling properties (Figure 6.2). These results suggested that the GAGbased matrix in Ch-MA hydrogels may play a unique role in supporting chondrogenic differentiation over PEG-based materials, independent of electrostatic growth factor sequestration and swelling, and the polysaccharide network or intermediate degree of negative charge in Ch-MA materials may have other undetermined influences on chondrogenic differentiation that warrant further investigation. As opposed to TGF- $\beta$ 1 sequestration by the material scaffolds to enhance MSC response, the materials themselves may alter the chondrogenic response by MSCs in the presence of soluble chondrogenic cues. As a moderately negatively charged component of cartilaginous matrix, chondroitin may alter the stem cell microenvironment through various interactions with ECM, cell surface receptors, and signaling molecules (besides TGF- $\beta$ 1)



to promote chondrogenic gene expression in encapsulated MSCs, in ways where synthetically-derived, neutrally charged PEG materials likely would not.

Despite significant differences in gene expression between MSCs encapsulated in Ch-MA and CS-MA hydrogels, these material effects were not as apparent in immunostaining for chondrogenic ECM markers. In general, greater ECM deposition was observed pericellularly in GAG-based CS-MA and Ch-MA materials than PEG control gels, which exhibited limited staining for ECM overall (Figure 6.4); however, notable differences in collagen II and aggrecan deposition were not evident between Ch-MA and CS-MA hydrogels, despite significant differences in gene expression. It appears that MSCs may have lacked sufficient extracellular space for significant matrix production in these tightly crosslinked hydrogel scaffolds, suggesting that degradation of the hydrogel network may play a key role in facilitating ECM deposition by encapsulated MSCs. While PEG-DA and OPF materials contain hydrolytically degradable ester moieties, PEG control hydrogels were not expected to experience significant degradation over the course of the 6-week study presented here. CS-MA and Ch-MA are degradable by chondroitinase enzyme, as described in Chapter 5; however, chondrogenic human MSCs may not produce sufficient amounts of chondroitinase enzyme to effectively degrade the GAG network, thus inhibiting the deposition of cartilaginous ECM by encapsulated MSCs. A previously study by Varghese et al. reported significant cellular aggregation of goat MSCs encapsulated in CS/PEG hydrogels, suggesting that cells of other mammalian species may possess greater ability to degrade and remodel the dense CS matrix [8]. Both hydrolytic and cell-mediated enzymatic degradation have been functionally incorporated into PEG-based hydrogels in order to support scaffold degradation in a controlled manner



[169, 185-187, 396], suggesting that similar techniques to enhance material degradation may support greater ECM deposition in Ch-MA and CS-MA materials.

While MSCs significantly upregulated expression of chondrogenic markers in 50% Ch-MA hydrogels, gene expression of collagen I did not decrease and actually appeared to increase over the time course of these experiments (Figure 6.3a). Collagen type I is normally produced in high amounts by undifferentiated MSCs, but is not normally present in hyaline cartilage [105]. While expression of collagen type II was very low at day 1 relative to collagen I expression, with only a  $6.9 \times 10^{-5} \pm 2.6 \times 10^{-3}$  ratio of collagen II:collagen I relative expression, by day 42 MSCs exhibited a significantly greater collagen II:collagen I ratio of  $1.03 \pm 1.20$ , despite some large variability, suggesting that collagen II and collagen I may have been expressed in comparable amounts at the mRNA level after 6 weeks (Figure 6.2b). As key differentiation markers of osteogenic, myofibroblastic, and adipocytic phenotypes were not significantly regulated in these studies, these gene expression results are representative of a more fibrochondrocytic phenotype. Fibrocartilaginous tissues appear in the menisci, annulus fibrosis of intervertebral discs, temporomandibular joint, as well as the insertions of tendon/ligament into bone, and contain a fibrous collagen I matrix along with cartilaginous ECM, including collagen II and aggrecan [27-29].

Hypertrophy of chondrogenic MSCs remains a key challenge to *in vitro* chondrogenic differentiation [97, 103-104], and gene expression of collagen X, an ECM marker of hypertrophic chondrocytes, was significantly upregulated in 50% Ch-MA hydrogels, compared to CS-MA and PEG control gels. In contrast, immunostaining results suggested that CS-MA materials may actually facilitate greater production or



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retention of collagen X matrix than Ch-MA hydrogels, even though relatively little ECM production by encapsulated cells was observed overall. Because sulfated GAGs are known to interact with a variety of ECM molecules [334, 397], it is important to consider that CS hydrogels may also differentially retain ECM within the scaffolds, compared to chondroitin. While some collagens can electrostatically interact with GAGs [398] and collagen X is highly glycosylated *in vivo* [399], the differential effects of sulfation on ECM retention require further investigation in greater depth, specifically in degradable or larger mesh size materials that may support greater degrees of ECM production.

DNA content within hydrogel constructs appeared decrease over time with a significant decrease between days 1 and 21 in all hydrogel types. This result is consistent with long-term culture in other PEG-based hydrogels in which the small mesh size and low overall degradability of the scaffold prevent cell division, due to limited space within the dense polymer matrix. A fraction of encapsulated cells likely undergo cell death over time, resulting in a decrease in total DNA [197, 207, 393]. Additionally, undifferentiated MSCs require the presence of adhesive cues that promote integrin binding to maintain cell viability in hydrogel materials [400]; however, persistence of these adhesive cues over time may inhibit the long-term chondrogenic response of MSCs [213]. MSCs express fibronectin during pre-cartilaginous condensation in the developing mesenchyme, then downregulate expression during differentiation into chondrocytes [24, 401], suggesting that early presentation of adhesive proteins or peptides may promote MSC viability following encapsulation and that removal of these adhesive signals over time may be necessary to support chondrogenic differentiation of MSCs [137].



It is also important to note that the DNA content in 50% Ch-MA was significantly greater than in 50% CS-MA and PEG controls by approximately two-fold on days 1 and 21, based on PicoGreen assay results. Though later time points may be explained by improved viability in Ch-MA materials, the difference in DNA on day 1 indicated a difference in encapsulation density, which could potentially confound results. However, based on the large differences in gene expression in Ch-MA hydrogels, along with similar trends of upregulation observed at early time points (days 7, 14, 21) in hydrogels with more similar DNA contents (Figure A.1-2), it appears that the cellular response in Ch-MA materials was primarily influenced by material interactions with Ch-MA in the presence of soluble chondrogenic factors, rather than simply increased cell density.

Together, these results have demonstrated that nonsulfated chondroitin hydrogels promote gene expression of chondrogenic markers by MSCs in a TGF- $\beta$ 1-dependent manner, and in Chapter 5, desulfation of CS was shown to reduce sequestration of TGF- $\beta$ 1, suggesting that electrostatic interactions with GAGs may play a role regulating TGF- $\beta$ 1 signaling to promote differentiation. However, it is important to consider that sulfation and charge may also alter the extracellular microenvironment through related differences in osmotic swelling pressure or through various interactions with the ECM, cell surface receptors, or other signaling molecules to influence MSC gene expression in response to soluble chondrogenic cues, independent of biomaterial sequestration of TGF- $\beta$ 1. Therefore, to fully elucidate the role of GAG-based materials in promoting chondrogenic differentiation, further investigation would be required to determine the cellular interactions with CS and downstream signals that result in an enhanced chondrogenic response. Controlled systems of study may clarify the importance of electrostatic growth



factor interactions in promoting differentiation. TGF- $\beta$ 3 would offer a closely related molecule with similar signaling mechanism and chondrogenic response to TGF- $\beta$ 1 (pI ~9.5), but TGF- $\beta$ 3 (isoelectric point, pI ~6.8) would not be expected to electrostatically interact with sulfated GAGs at physiological pH due to its difference in charge [274-276], providing a closely related comparison to examine the role of electrostatic growth factor interactions in differentiation.

CS and nonsulfated chondroitin may be combined in varying ratios to develop materials with a range of negative charge densities, and the results presented here suggest that these materials may be valuable tools to examine the role of charge in directing differentiation of MSCs in a highly controlled manner. The combined roles of sulfation in modulating TGF- $\beta$ 1 sequestration in Chapter 5 and chondrogenic gene expression by MSCs in nonsulfated materials in the presence of soluble chondrogenic cues in these studies suggested that an intermediate degree of sulfation may be advantageous to enhance TGF- $\beta$ 1 signaling and MSC response for greater chondrogenic differentiation. Alternatively, altering the presentation of TGF- $\beta$ 1 through a dose response may examine the ability of CS-based materials to regulate signaling in response to low TGF- $\beta$ 1 concentration environments, or temporal control of TGF- $\beta$ 1 presentation by withdrawing exogenously supplemented TGF- $\beta$ 1 after a period of pre-culture may examine the persistence of these signals in CS-based hydrogels.

Sulfated and nonsulfated GAG materials provide a highly controlled system to examine the interactions between charged GAGs and MSCs to promote chondrogenic differentiation, possibly through sequestration of growth factors, and these materials



possess tremendous potential as tissue engineering constructs for controlled tissue regeneration and patterning.

# 6.5 Conclusions

These studies have demonstrated that encapsulation in CS and chondroitin hydrogels enhanced gene expression of chondrogenic markers by MSCs in 50% Ch-MA materials in the presence of TGF-\beta1, compared to 50% CS-MA and PEG-only hydrogels, suggesting that chondroitin materials may be better suited for supporting chondrogenic differentiation than CS, due to their reduced sulfation and negative charge density. However, differences in production of cartilaginous ECM were not apparent in between Ch-MA and CS-MA hydrogels, as MSCs may lack sufficient space for significant matrix deposition in these tightly crosslinked hydrogel scaffolds. Greater degradation of the hydrogel network, either through hydrolytic or enzymatic means, may be required to facilitate ECM deposition in these GAG-based materials. These results present CS and nonsulfated chondroitin materials as valuable tools to alter GAG sulfation in a highly controlled manner as a biomaterial approach to differentially regulate MSC response to soluble differentiation cues. These concepts represent a first step in the development of novel biomaterials for regulating chondrogenic cues to regenerate the complex ECM architecture of cartilaginous tissues for long-term repair.



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# **CHAPTER 7**

# **CONCLUSIONS AND RECOMMENDATIONS**

#### 7.1 Summary

Tissue engineering strategies offer exciting and innovative approaches to treat cartilage injuries for long-term regeneration and repair; however, difficulty regenerating the complex extracellular matrix (ECM) composition and organization of native cartilage remains a significant challenge. Mesenchymal stem cells (MSCs) are multipotent progenitor cells that are capable of forming cartilaginous tissues, making them a promising cell source for tissue repair [6]. In cartilage tissue, cells are expected to synthesize, maintain, and remodel the ECM to maintain function and integrity; however, the complex cues necessary to promote cartilage formation are not fully understood [5]. Recent research has begun to investigate the ability of cartilaginous ECM molecules to promote and direct MSC differentiation down a chondrogenic lineage.

Chondroitin sulfate (CS), a cartilaginous glycosaminoglycan (GAG) that is expressed both during cartilage development and in adult cartilage associated with proteoglycans, has been shown to promote production of cartilaginous ECM by MSCs [8-9]; therefore, CS appears to play a role in altering stem cell microenvironments to direct differentiation down a chondrogenic lineage. The overall goal of this dissertation was to develop versatile biomaterial platforms to control CS presentation to MSCs in order to improve understanding of the role of CS in promoting chondrogenic differentiation. To investigate chondrogenic response to a diverse set of CS materials, progenitor cells were cultured in the presence of CS proteoglycans and CS GAG chains in a variety of 2D and



3D material systems, including surfaces coated with aggrecan proteoglycan, CS-based nanospheres and microspheres, and desulfated chondroitin hydrogels. Together these studies provided valuable insight into the unique ability of CS materials to alter cell morphology and growth factor sequestration to promote chondrogenic differentiation as part of tissue engineering strategies to promote cartilage regeneration and repair.

In Chapter 3, cellular interactions with 2D aggrecan-coated surfaces were examined for morphology and production of cartilaginous ECM. Aggrecan was passively adsorbed onto tissue culture-treated surfaces, and bovine bone marrow stromal cells (BMSCs) and anterior cruciate ligament (ACL) fibroblasts were cultured on aggrecan-coated 2D surfaces, compared to tissue culture-treated control surfaces and nonadhesive surfaces that promoted cell aggregation. Culture on aggrecan surfaces promoted cell aggregation, and BMSC and ACL fibroblast aggregates significantly upregulated aggrecan gene expression and production over 14 days of culture in the absence of chondrogenic media supplements, regardless of how cell clustering was induced. These findings support the use of aggregate-inducing materials, including CS-modified surfaces, to encourage production of chondrocytic ECM.

In Chapter 4, small particle carriers were developed to study CS interactions with charged growth factors. CS-derived nanoscale micelles and microscale particles were fabricated as potential growth factor carriers to enhance stem cell differentiation, and particles were characterized for size, surface charge, cytocompatibility, as well as growth factor release. Conjugation with a hydrophobic methacrylamide group induced spherical CS micelles to self-assemble in an aqueous environment, and micelles were negatively



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charged with a bimodal distribution of ~320 and ~73 nm diameters. Larger CS microspheres, synthesized from crosslinking of methacrylated CS in a water-in-oil emulsion, possessed a rounded morphology and a diameter of ~4.3  $\mu$ m. Positively charged transforming growth factor- $\beta$ 1 (TGF- $\beta$ 1) demonstrated minimal release from CS microspheres over 5 days, while negatively charged tumor necrosis factor- $\alpha$  (TNF- $\alpha$ ) exhibited substantial burst release, suggesting that TGF- $\beta$ 1 electrostatically complexed with negatively charged CS. These studies suggested that CS-based carriers can be fabricated over a variety of nano- and microscale sizes, as potential ECM-derived carriers of positively charged growth factors to direct stem cell differentiation.

In Chapters 5 and 6, desulfated chondroitin materials were synthesized to study sulfation-dependent CS interactions in growth factor sequestration and chondrogenic differentiation. CS was chemically desulfated by acidic methanol treatment, and desulfated chondroitin materials were thoroughly characterized to ensure desulfation and decreased charge density without significant modification of the CS chemical structure. Desulfated chondroitin demonstrated ~98.5% desulfation after 7 days of chemical treatment. Crosslinked CS hydrogels consistently demonstrated greater sequestration of TGF- $\beta$ 1 as indicated by greater depletion from solution and less release after loading, than desulfated chondroitin; however, human MSCs encapsulated in nonsulfated chondroitin hydrogels experienced significantly greater gene expression of cartilaginous markers when cultured for 6 weeks in chondrogenic medium containing TGF- $\beta$ 1. These results suggested that controlled desulfation of GAG-based materials modulated growth factor interactions, and less sulfated chondroitin materials may support greater chondrogenic differentiation of human MSCs in the presence of chondrogenic medium.



Together, these findings presented in this dissertation suggested that CS plays an important role in directing chondrogenic differentiation via morphological and electrostatic interactions, and CS and chondroitin materials are promising tools to control GAG presentation within a variety of stem cell microenvironments to promote differentiation for cartilage repair.

# 7.2 Conclusions

The research presented in this dissertation advances understanding of how diverse CS-based biomaterials can control the presentation of GAGs within a variety of stem cell microenvironments to promote differentiation. Due to the abundance of CS in cartilaginous tissues, both during development and in adult tissues, highly controlled GAG presentation may have important implications in the engineering of novel strategies for cartilage repair. 2D surfaces, small scale particle carriers, and 3D bulk hydrogels were developed from CS-based materials to investigate stem cell interactions with sulfated GAGs, and the results of these studies suggested that highly negatively charged CS-based materials can 1) present morphological cues to promote cell aggregation and 2) regulate biomaterial interactions with charged growth factors to promote differentiation. The sulfation and resulting negative charge density of CS appears to play a critical role in morphological and growth factor signaling cues that may direct the chondrogenic differentiation of MSCs. Together, these results suggested that CS-based materials may possess the unique ability to deliver and modulate growth factor interactions through delivery and sequestration within stem cell aggregates.



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Adsorption of the cartilaginous CS proteoglycan aggrecan onto 2D surfaces, as described in Chapter 3, promoted cellular aggregation of bovine BMSCs and ACL fibroblasts. The presence of aggrecan appeared to resist cell attachment and spreading on tissue culture surfaces, forcing adherent cells to cluster in close contact with one another. Formation of these dense BMSC and ACL fibroblast aggregates was accompanied by upregulation of aggrecan gene expression; however, upregulation was found to be a response to morphological cues rather than a response to the aggrecan-coated surfaces. Cell aggregates on nonadhesive culture surfaces that encouraged clustering in the absence of aggrecan also promoted upregulation of aggregation may enhance ECM production through changes in cell shape, cell-cell contact, or intercellular signaling. These results demonstrated that CS-based materials could be used as a naturally-derived material to promote cell clustering and emphasized the importance of high density culture in promoting and supporting expression and production of cartilaginous ECM.

Pellet culture and micromass culture are established methods to delay the dedifferentiation of chondrocytes or to promote chondrogenic differentiation of MSCs [6, 98, 304-305]; therefore, it is not surprising that aggregation promoted a more cartilaginous phenotype in these studies. High-density culture supports close cell-cell contact and appears to mimic the cell environment of mesenchymal condensation that occurs during early cartilage development [24, 308]. Within the condensing mesenchyme, neural-cadherin (N-cadherin), neural cell adhesion molecule (N-CAM), and gap junctions form cell-cell contacts that facilitate intercellular communication and play a critical role in regulating the deposition of cartilaginous ECM [24]. As increasing



amounts of ECM are produced, the cells are pushed apart until individual chondrocytes become embedded in cartilaginous matrix [13, 27]. In this dissertation, aggrecan-coated surfaces used a native cartilage proteoglycan to control cell morphology by promoting self-assembly into dense cell aggregates, possibly due to the anti-adhesive properties of negatively charged CS [402-404]. Subsequent production of aggrecan proteoglycan by bovine BMSC and ACL fibroblast aggregates in the absence of exogenous chondrogenic factors suggested that aggrecan-mediated clustering may be especially advantageous to promote stem cell differentiation toward more cartilaginous or fibrocartilaginous phenotypes. CS-containing 3D hydrogels have also been demonstrated to support cell aggregation and subsequent production of chondrocytic ECM by encapsulated MSCs in the presence of chondrogenic medium [8]. Together, these data suggested that GAGbased materials could potentially be used as ECM-derived cell carriers to promote cell clustering and production of chondrocytic ECM in the development of tissue engineering strategies for cartilage repair.

The importance of high-density culture in chondrogenesis suggested that the development of biomaterial platforms to control presentation of CS within dense cell aggregates may be advantageous to promote chondrogenic differentiation. A significant challenge that accompanies micromass culture of stem cells involves transport properties within cell pellets. These dense, multicellular spheroids present numerous barriers to diffusion, in particular limited transport of soluble growth factors from the culture medium [322]. Insufficient transport may result in heterogeneous or disorganized differentiation within MSC pellets or embryonic stem cell (ESC) embryoid bodies; therefore, micromass culture presented a need to enhance growth factor signaling within



dense cell aggregates. CS-based materials may possess the unique ability to enhance growth factor signaling through sequestration, delivery, and release of growth factors to promote stem cell differentiation; therefore, CS-based biomaterial platforms were developed to examine electrostatic interactions between sulfated GAGs and positively charged growth factors.

Negatively charged GAGs have been shown bind and sequester a variety of charged growth factors [263, 272, 319], and these interactions can be exploited for controlled growth factor retention and release by GAG-based carriers [262, 264]. CS is a naturally occurring ECM component that is abundant in cartilaginous tissues, and TGF- $\beta$ 1 plays critical roles in promoting chondrogenesis in vivo and in vitro [340-342]; therefore, electrostatic interactions between CS and TGF- $\beta$ 1 may be particularly advantageous to promote chondrogenic differentiation of stem cells. Sulfated GAGs appear to interact with TGF- $\beta$ 1 via basic amino acid residues at the tips of its  $\beta$ -strand loops [272, 274], and highly sulfated GAGs may potentiate TGF- $\beta$ 1 activity, suggesting that GAG interactions may have a synergistic effect in TGF-β1 signaling [251, 272, 278]. CS proteoglycans are highly regulated during mesenchymal condensation during early cartilage development [14-18], and culture in CS-containing hydrogels has demonstrated upregulated expression and production of chondrocytic ECM by encapsulated mesenchymal stem cells in the presence of chondrogenic medium [8-9], suggesting that CS may play a unique role in directing growth factor signaling to promote chondrogenic differentiation of progenitor cells [14-15].

Because the high negative charge density of CS is largely attributable to the abundance of sulfate groups along its repeating GAG backbone, sulfation was expected



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to play a significant role in modulating interactions with charged growth factors; therefore, in Chapter 5, the role of sulfation in growth factor sequestration was examined through chemical desulfation of the primary CS backbone. CS was chemically desulfated by acidic methanol treatment for 7 days, and desulfation yielded a nonsulfated chondroitin product, resulting in a significant reduction in negative charge density, with little modification of the native CS chemical structure. Subsequent methacrylation of these chains allowed CS and chondroitin to be covalently crosslinked to form bulk 3D hydrogels; therefore, CS and desulfated chondroitin materials provided a highly controlled system to study the role of sulfation and resulting negative charge in electrostatic sequestration of growth factors, specifically TGF- $\beta$ 1.

Desulfation of CS was found to alter the sequestration and release of TGF- $\beta$ 1 in a sulfation-dependent manner *in vitro*. In release studies, in which PEG hydrogels containing varying amounts of CS and chondroitin were loaded with TGF- $\beta$ 1, 50% CS materials demonstrated the greatest retention and the least cumulative release of TGF- $\beta$ 1 over 7 days, while decreasing the CS content to 10% CS-MA resulted in significantly more release, suggesting that CS retained TGF- $\beta$ 1 in a concentration-dependent manner. Additionally, 50% chondroitin hydrogels, which maintained an identical total GAG content to 50% CS hydrogels but with reduced sulfation, exhibited significantly greater release of TGF- $\beta$ 1 than 50% CS-MA, and these results suggested that sulfation and corresponding negative charge were critical in facilitating electrostatic interaction with growth factors. The ability of CS to retain growth factors, sequestering them from release, proposed that CS-based biomaterials may have potential applications in sequestering growth factors for enhanced stem cell differentiation.



To examine an alternate system of sequestration, rather than release from CS materials, hydrogels containing varying ratios of CS and nonsulfated chondroitin while maintaining a 50% total GAG content were examined for their ability to deplete soluble TGF-β1 by "pulling" it out of solution and trapping it with their hydrogel networks. After incubation in a 2 ng/mL TGF-B1 solution, 50% CS hydrogels demonstrated significant depletion of soluble TGF- $\beta$ 1 out of solution, while decreasing the sulfation of the GAG matrix reduced the observed interaction with TGF- $\beta$ 1 in a sulfation-dependent manner. Nonsulfated 50% chondroitin gels exhibited significantly less pull-down of TGF- $\beta$ 1, while 10% CS/40% chondroitin hydrogels sequestered an intermediate degree of TGF- $\beta$ 1. In addition, incubation in 0.5 M NaCl or 10 mg/mL soluble CS significantly inhibited the amount of depletion by 50% Ch-MA hydrogels; however, a similar decrease in depletion was not observed in 50% CS-MA hydrogels. These results suggested that while chondroitin materials retain some ability to electrostatically sequester TGF- $\beta$ 1, likely due the presence of negatively charged carboxylates along its repeating backbone, binding with chondroitin appeared to be much weaker than with CS. The inability of soluble CS to competitively inhibit binding to 50% CS-MA also suggested that TGF-β1 may specifically possess a stronger ability to bind to crosslinked CS hydrogels over soluble CS. Together these results suggested that CS-MA hydrogels possessed a relatively strong ability to sequester soluble TGF- $\beta$ 1 out of solution, and that binding of TGF- $\beta$ 1 can be controlled in a sulfation-dependent manner through chemical desulfation of CS.

Desulfation of CS provides a well-controlled biomaterial system to alter the electrostatic interactions between CS and charged growth factors, and these materials could therefore be used to control presentation of growth factors to stem cells for



differentiation. The ability of CS and chondroitin to differentially "pull-down" soluble TGF- $\beta$ 1 out of solution provided rationale to examine the effect of differentially sulfated materials on the chondrogenic differentiation of MSCs, as CS may similarly sequester TGF- $\beta$ 1 from chondrogenic media for controlled presentation to encapsulated MSCs. To characterize the cellular response while entrapped in CS and nonsulfated chondroitin matrix, human MSCs were encapsulated in PEG hydrogels containing either 50% CS or 50% chondroitin and cultured for 6 weeks in chondrogenic media media for Controlled provided an established model for chondrogenic differentiation in PEG-based hydrogels provided an established model for chondrogenic differentiation of MSCs *in vitro* [136, 199, 206-208], and PEG-based materials, in particular, have been used in the past as systems to investigate the effects CS matrix on chondrogenic differentiation of MSCs [8-9, 252].

CS- and chondroitin-containing materials were shown to differentially modulate the chondrogenic response of encapsulated MSCs in response to chondrogenic growth factor, suggesting that sulfation may play a role in regulating chondrogenic differentiation; however, MSCs in 50% chondroitin hydrogels displayed significantly greater upregulation of chondrogenic markers than 50% CS gels after 6 weeks of culture in chondrogenic medium. Chondrogenic gene expression in chondroitin hydrogels was dependent on the presence of exogenously supplemented TGF- $\beta$ 1 in the medium, suggesting that chondroitin materials alone were not sufficient to upregulate expression and that desulfation of the CS matrix enhanced the chondrogenic response to TGF- $\beta$ 1. Despite a higher degree of TGF- $\beta$ 1 sequestration, as shown in Chapter 5, the high negative charge density of CS in these materials could possibly decrease TGF- $\beta$ 1 signaling activity or inhibit transport within the hydrogel network. The highly charged



CS matrix may also prevent transport of other cell-secreted signals within the hydrogel, effectively inhibiting intercellular communication, which plays an important role in supporting chondrogenic differentiation and maintaining a chondrocytic phenotype [24, 38, 394-395]. Removal of sulfate groups from CS, however, appeared to promote MSC differentiation, possibly through enhanced growth factor signaling. Desulfated chondroitin materials remained moderately negatively charged at physiological pH, due to the presence of carboxylates in the chondroitin structure, and chondroitin may still possess electrostatic interactions with TGF- $\beta$ 1, though binding appeared to be weaker than with CS. These results suggested that decreasing the degree of sulfation in GAGbased hydrogels may be better suited for supporting chondrogenic differentiation than highly sulfated CS, despite stronger electrostatic interaction with CS, and that CS and chondroitin materials possess potential biomaterials to as alter stem cell microenvironments to differentially regulate chondrogenic differentiation of MSCs.

Unfortunately, despite large differences in gene expression after 42 days of culture *in vitro*, relatively little ECM deposition was observed within the hydrogel constructs with no noticeable difference between CS and chondroitin hydrogels. This is likely due to the small mesh size of these highly-crosslinked hydrogel constructs, as MSCs may have lacked sufficient extracellular space for significant matrix production. Controlled degradation of the hydrogel network may play a key role in facilitating ECM deposition by encapsulated MSCs, either by hydrolytic or cell-mediated enzymatic means. Hydrolytic degradation of crosslinked hydrogel scaffolds is dependent on the number of ester moieties within the polymer backbone; however, the degradation rate of these esters can be systematically controlled by altering the hydrophilicity of the


surrounding environment. Cell-mediated degradation through the incorporation of matrix metalloproteinase (MMP)-cleavable peptide sequences can encourage controlled degradation localized to the extracellular environment, and degradation rate can be customized through these peptides for susceptibility to cleavage, as well as specificity to a single MMP type that is expressed by specific cell populations [169, 185-187, 396]. Controlled degradation of CS-MA and Ch-MA hydrogel networks may facilitate greater ECM deposition and enhanced chondrogenic differentiation by encapsulated MSCs.

Despite significant upregulation of chondrogenic markers, collagen X, an ECM marker of hypertrophic chondrocytes, was also significantly upregulated in 50% chondroitin hydrogels; however, 50% CS gels exhibited visibly greater deposition of pericellular collagen X after 42 days than 50% chondroitin. Hypertrophy of chondrogenic MSCs remains a key challenge to *in vitro* differentiation [97, 103-104]; however, just as specific signals may promote chondrogenic differentiation of MSCs, it is also likely that additional signals may be required to maintain a mature chondrocyte phenotype, while inhibiting progression toward hypertrophy. Indian hedgehog (Ihh) and parathyroid hormone-related protein (PTHrP) signaling appear to interact in vivo through paracrine signaling from the perichondrium to regulate chondrocyte maturation and hypertrophy [38]. Basic fibroblast growth factor (bFGF) and PTHrP have also been investigated for their ability to delay hypertrophy and maintain chondrocytic phenotype in vitro with some success [106-109]. Additionally, collagen type I gene expression and production did not appear to decrease in 50% chondroitin materials, representative of a more fibrocartilaginous response. Collagen type I is normally produced in high amounts by undifferentiated MSCs; however, during cartilaginous condensation in early



development, collagen I expression is turned off and replaced with production of collagen type II and aggrecan proteoglycan in the formation of articular cartilage [24]. Together these results suggested that a variety of diverse signals may be required, along with both spatial and temporal control of these signals, to properly support chondrogenic differentiation and maintenance of a chondrocytic phenotype.

The desulfation of CS provided a highly controlled biomaterial system to alter both the sulfation and charge of CS-based materials, with minimal modification of the GAG chemical structure, to examine sequestration of soluble signals for MSC differentiation toward a chondrogenic phenotype. The combined roles of sulfation in modulating TGF- $\beta$ 1 sequestration and chondrogenic gene expression by MSCs in the presence of soluble chondrogenic cues in these studies suggested that an intermediate degree of sulfation may be advantageous to enhance TGF- $\beta$ 1 signaling and MSC response for greater chondrogenic differentiation. The ability of desulfation to promote chondrogenic gene expression in these studies suggested that an optimal amount a charge may exist to facilitate growth factor interaction, while possibly permitting intercellular communication. As demonstrated in Chapters 5, CS materials can be either be partially desulfated by time-dependent acidic methanol treatment or CS and nonsulfated chondroitin can be combined in varying ratios to develop materials with a range of negative charge densities without altering the total GAG content or composition within the bulk material. Additionally these materials may possess the unique ability to potentiate signaling in low TGF-B1 concentration environments or to retain growth factor for sustained signaling after removal of soluble TGF-\u00b31 from the medium. This



demonstrates that CS and chondroitin are versatile systems for highly controlled interaction with stem cells and growth factors.

Due to the low degree of total ECM deposition observed in these tightly crosslinked hydrogel systems, ECM production may be enhanced by increasing the ratio of cells to matrix, thus allowing the encapsulated cells to further remodel the GAG network. While this model system involved encapsulation of MSCs in a large amount of crosslinked GAG matrix to examine its role in promoting a chondrogenic response, at the other extreme, lesser amounts of CS-based materials could be incorporated into a multicellular mass to allow unrestricted ECM deposition in response to cues from the incorporated CS matrix. Specifically, CS-based small particle carriers could potentially be incorporated to MSC aggregates to permit close cell-cell contact and used to deliver various growth factors throughout the cell aggregates for enhanced differentiation, compared to diffusion of soluble factors from the medium [322].

As described in Chapter 4, CS-based small particle carriers were developed for controlled delivery of charged growth factors. As a first step, CS-based nanoscale micelles and microscale particles were fabricated and characterized as small particle carriers for growth factor delivery. With low degrees of modification, methacrylamide-conjugated CS (CSMAm) self-assembled into ~73 nm and ~320 nm diameter micelles in aqueous solution. These micelle particles were found to possess a highly negative surface charge, suggesting that they may electrostatically interact with positively charged growth factors. Crosslinking of methacrylated CS (CSMA) in a water-in-oil, single-emulsion resulted in the formation of larger CS microspheres with an average diameter of  $4.3 \mu m$ .



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over 5 days, while negatively charged TNF- $\alpha$  exhibited substantial burst release within 15 hours at physiological pH. These results suggested that negatively charged CSMA microspheres electrostatically interacted with TGF- $\beta$ 1, sequestering it from release. CSMA microspheres were successfully incorporated into ESC embryoid bodies with good cytocompatibility, suggesting that CSMA microspheres may possess unique application as carriers for controlled delivery and presentation of growth factors within dense multicellular aggregates. Together with techniques for desulfation of CS to control electrostatic interactions and the ability of CS matrix to differentially modulate chondrogenic response in the presence of TGF- $\beta$ 1, this work represents a first step in the development of novel biomaterials to control presentation of CS in high-density micromass culture to promote chondrogenic differentiation of MSCs.

While traditional tissue engineering paradigms involve the culture of dispersed cells within a biomaterials scaffold and supplementation of exogenous differentiation cues through the culture medium, this work presents an alternative approach in which small scale biomaterials are entrapped throughout a cellular mass with delivery of differentiation factors from the embedded biomaterial carriers. This is a novel idea for tissue engineering of cartilaginous tissues in which high-density culture and formation of cell-cell contacts play important roles in differentiation. The ability of CS to electrostatically sequester TGF- $\beta$ 1 also may resolve transport limitations associated with traditional micromass culture. GAG-based materials have been shown to possess electrostatic interactions with a number of other chondrogenic growth factors including bone morphogenetic protein-4 (BMP-4), insulin-like growth factor-1 (IGF-1), and bFGF, some of which have demonstrated additive effects when supplemented in combination or



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sequentially with TGF- $\beta$  [25, 256, 264, 273]. These results suggest that CS-based carriers could be used as an ECM-derived carrier to deliver a variety of signals within a spatially and temporally controlled environment to promote differentiation of stem cell aggregates.

The CS-based biomaterials presented in this dissertation provide highly controlled systems to alter a variety of signals within stem cell microenvironments. CS-based materials were shown to regulate cell aggregation, presentation of sulfation and charge, growth factor interactions, and MSC response to soluble chondrogenic cues, and these materials possess tremendous potential to control growth factor signaling within stem cell microenvironments. Therefore, CS-based materials are valuable tools to investigate the role of a diverse array of soluble, physical, and morphological cues in stem cell differentiation, and additional spatial and temporal control can be engineered into this unique biomaterial system to modulate the stem cell environment to promote differentiation.

#### 7.3 Future Directions

The findings presented in this dissertation provide significant insights into the potential interactions of sulfated GAGs that alter stem cell microenvironments to promote chondrogenic differentiation, including cell aggregation and electrostatic interactions with growth factors. A variety of CS-based materials, including 2D surfaces, small scale particles, and bulk hydrogels with varying degrees of sulfation, were developed as novel tools to control CS presentation to stem cells to investigate the role of CS in chondrogenic differentiation; however, future work can expand on the insights gained



from this dissertation to examine mechanisms through which GAG-based materials control stem cell differentiation.

The results of these experiments suggested that careful control of GAG structure may present a unique opportunity to further investigate the control of charge and sulfation in supporting chondrogenic differentiation. The studies in this dissertation represent a broad examination of how sulfated CS and nonsulfated chondroitin independently influence chondrogenic differentiation; however, significant differences in gene expression in chondroitin materials suggested that an intermediate level of sulfation and negative charge may be more optimal for chondrogenic differentiation, as greater degree of TGF- $\beta$ 1 sequestration in CS materials did not translate to a stronger chondrogenic response. An intermediate charge density may facilitate moderate interaction with growth factors while still permitting transport of soluble factors and potential intercellular communication.

For a thorough investigation of charge, CS materials can either be partially desulfated by time-dependent acidic methanol treatment or CS and nonsulfated chondroitin can be combined in varying ratios to develop materials with a range of negative charge densities without altering the total GAG content or composition within the bulk material; however, the distribution of sulfation would differ in these two approaches, as the former would contain a homogenous undersulfated CS variant, while the latter would include primarily monosulfated CS chains in a mixture of completely desulfated chondroitin. Differences in these two approaches may be able to address the role of sulfation patterning and density, compared to the role of bulk charge, in CS-based materials for growth factor sequestration and chondrogenic differentiation. In addition,



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greater charge than the primarily monosulfated CS used in this dissertation can be achieved through a variety of chemical techniques to oversulfate CS [25-26], and desulfated chondroitin, which remains negatively charged at physiological pH due to the presence of carboxylates in the chondroitin structure, could be chemically modified to possess a neutral charge. A neutrally charged chondroitin material may be especially valuable to examine the effects of uncharged GAG variants, and such a material could be easily obtained in the methyl ester intermediate as a part of the acid methanol treatment, as described in Chapter 5, in which the carboxylates of chondroitin are methylated, resulting in a neutral charge. This versatility to control charge over a wide range from uncharged to oversulfated CS chains offers a highly controlled system to investigate the various roles of sulfation and charge in growth factor interactions and differentiation, without modification of the primary CS backbone.

Besides total degree of sulfation or total charge density, sulfation pattern may also play important roles in modulating GAG interactions with cells, signaling molecules, and the ECM. It has been documented that 2-O-sulfation is especially critical for growth factor interactions in heparin [266-267, 270, 358], and CS sulfation pattern has been shown to alter electrostatic interactions with  $Co(NH_3)_6^{+3}$  cations, in which monosfulated chondroitin-4-sulfate possessed greater affinity than equally sulfated chondroitin-6sulfate [405]. In Chapters 5 and 6 of this dissertation, the chondroitin sulfate used contained a mixture of 58% chondroitin-6-sulfate and 27% chondroitin-4-sulfate disaccharides, along with a 15% mixture of other nonsulfated and disulfated disaccharides, as determined in Chapter 5. This CS composition was chosen both as a result of its increased susceptibility to desulfation as described in Chapter 5, as well as



chondroitin-6-sulfate's relative prevalence in adult human articular cartilage; however, chondroin-4-sulfate is more highly expressed during growth and development of human cartilage before decreasing expression in the mature tissues [20, 231], suggesting that 4-sulfation may potentially play an increased role in differentiation and development. Differences in sulfation pattern, specifically increased chondroitin-4-sulfate composition, may exhibit important differences in growth factor interaction and stem cell differentiation that were not observed in this examination of a single CS mixture that was a majority chondroitin-6-sulfate.

It is also important to recall that CS is predominantly associated as part of proteoglycans *in vivo* in which GAGs are bound to a core protein. In cartilage specifically, versican is expressed in cartilaginous condensations during development [218-219], while aggrecan is the predominant proteoglycan in mature cartilage [31]; therefore, presentation of GAGs in conjunction with the full proteoglycan structure may further alter GAG presentation to control ECM interactions and cellular presentation. To fully understand the role of GAGs in cartilaginous tissues, a thorough investigation of cellular response in the presence of CS proteoglycans would be necessary to elucidate the roles of the protein core, GAG presentation, and interplay between GAGs and their protein cores in signaling and tissue formation.

Although these studies demonstrated that sulfation of CS influenced growth factor interactions and that differential sulfation also altered the chondrogenic response, the cellular response was not conclusively linked to growth factor signaling, as differences in charge also could potentially alter osmotic swelling pressure or other various ECM interactions that influence the extracellular microenvironment [20-21, 250]. Therefore, to



fully understand the role of CS in promoting chondrogenic differentiation, further investigation is also required to determine the cellular interactions with CS that result in an enhanced chondrogenic response. While it is difficult to completely separate the independent effects of sulfation, fixed charge, osmotic pressure, swelling, and electrostatic interactions with a variety of molecules without significant modification of these materials, due to their high interdependency, controlled systems of study may begin to start clarifying the importance of these factors in promoting differentiation.

Specifically, regarding growth factor interactions with TGF- $\beta$ 1 as presented in this dissertation, use of TGF- $\beta$ 3 would offer a closely related molecule with similar signaling mechanism and chondrogenic response to TGF- $\beta$ 1; however, TGF- $\beta$ 3 is not expected to electrostatically interact with sulfated GAGs at physiological pH [274-275]. The isoelectric point of TGF- $\beta$ 3 is ~6.8, compared to ~9.5 in TGF- $\beta$ 1, suggesting that TGF- $\beta$ 3 is slightly negatively charged at physiological pH [275-276]. Additionally, a basic amino acid at position 26 in TGF- $\beta$ 1, a central part of the hypothesized GAGbinding site, appears to be replaced with a neutral amino acid in TGF- $\beta$ 3, suggesting that it may be less likely to interact with CS [274].

To fully implicate TGF- $\beta$ 1 in the chondrogenic response observed in chondroitin hydrogels, a comprehensive investigation of the downstream receptors and signaling molecules, such as SMAD and ERK/MAPK pathways, would be required to determine the differential effects of growth factor signaling in these materials [406-408]. Subsequent receptor blocking and inhibition of these pathways could begin to answer these questions; however, signaling from chondrogenic growth factors appeared to play an important role in these interactions, as culture of MSCs in chondroitin materials in the



absence of chondrogenic medium did not observe comparable effects from the material alone.

It is, however, important to note that MSCs encapsulated in CS-MA hydrogels did significantly upregulate collagen II gene expression, and to a lesser extent aggrecan, in the absence of TGF- $\beta$ 1 and dexamethasone on day 21, while chondroitin materials did not, in an investigation of these materials at earlier time points and lower cell density (Figure A.2). Investigation of CS and chondroitin materials in low TGF- $\beta$ 1 environments warrants further examination to determine the corresponding roles of TGF- $\beta$ 1 signaling in these two materials. The effects of a TGF- $\beta$ 1 dose response on MSC expression in the presence of CS and chondroitin materials may clarify these patterns, particularly at low and intermediate TGF-B1 concentrations. Chondrogenic medium commonly uses 10 ng/mL TGF- $\beta$ 1 to promote chondrogenic response [6]; however, this concentration is quite high, in order to promote a strong signaling response and encourage chondrogenic differentiation. CS-based materials, however, may potentiate chondrogenic response by sequestering TGF- $\beta$ 1 and increasing the local concentration of growth factor in close proximity to cells or enhancing signaling efficiency within a GAG-based scaffold. CS and chondroitin materials may be uniquely capable of capturing and potentiating signals in low TGF-\u00df1 concentrations, while overwhelming TGF-\u00bf1 concentrations may negatively impact the additional signals necessary to promote chondrogenic differentiation.

As the goal of tissue engineering is to develop replacement tissues for tissue repair, it remains critically important that chondrogenic MSCs in these CS-based systems produce cartilaginous ECM in a controlled fashion. The hydrogel materials discussed in



Chapter 6, however, supported relatively little ECM production, and all matrix production was restricted to the pericellular domains. It is likely that enhanced degradation of the hydrogel network may be required to support deposition of cartilaginous ECM, and because human MSCs do not appear to produce sufficient amounts of chondroitinase enzyme to encourage substantial degradation of the GAG matrix, additional mechanisms of degradation may be required to support ECM production. Hydrolytically degradable and cell-mediated enzymatically cleavable functionality have been investigated in similarly crosslinked PEG-based system [169, 185-187, 396]; therefore, similar functionality could be incorporated in GAG-based systems to examine the role of degradation on enhanced ECM deposition.

The unique ability of CS-based materials to selectively sequester soluble signals from solution suggests that CS-based materials may also possess considerable potential in spatial patterning of scaffolds to control differential effects of signaling. CS and chondroitin materials may be used to either trap soluble signals to potentiate signaling, or alternatively to sequester cell-secreted signals and restrict paracrine signaling in a spatially-defined fashion. These principles can be applied to spatially patterned hydrogel materials through photolithographic techniques to promote differential signaling from a single cell type within a single media formulation, by differentially trapping soluble cues from the medium in the CS matrix [192, 195, 197]. Spatially controlled GAG-based scaffolds may represent novel biomaterial systems to regenerate tissue interfaces through co-culture of multiple cell types through differential signaling, as well as promoting formation of various properties within a single tissue, such as the depth-dependent zones of cartilage.



Due the importance of cell-cell contact in production of cartilaginous ECM exhibited in Chapter 3, CS-based microparticles demonstrate a versatile delivery system to examine the effects of CS within chondrogenic MSC spheroids. The use of CS microparticles within MSC micromass cultures offers an innovative system to control growth factor presentation for "inside-out" delivery, as opposed out "outside-in" signaling via diffusion from the medium, through the use of naturally-derived GAG materials that may potentiate signaling and support further interaction with surrounding cells. While microparticle carriers were developed for growth factor delivery within MSC and ESC spheroids, as described in Chapter 4, a thorough examination is required to characterize their role in promoting chondrogenic differentiation. A few studies have begun to investigate the role of adhesion peptides and TGF- $\beta$  delivery in MSC pellets to promote chondrogenic differentiation [409-410], and these experiments can be used as guiding principles in the optimization of this CS-based delivery system as an ECMderived growth factor carrier for cartilage regeneration. Additional incorporation of desulfated chondroitin materials into this microparticle carrier system may offer further control of growth factor affinity, release kinetics, and signaling activity to control differentiation, as has been suggested in Chapters 5 and 6 of this dissertation.

Aside from being a system to promote differentiation through growth factor interactions, GAG-based materials may possess significant application as an analytical tool to study stem cell signaling. GAG-based materials may be capable of "trapping" specific cell-secreted signals for characterization and potential future delivery. Microparticles may be incorporated into stem cell aggregates, such as ESC embryoid bodies, to sequester cell-secreted signals involved in the maintenance and self-renewal of



differentiation potential. Dissociation of the cell spheroids and retrieval of the microparticles for molecular analysis may be able to clarify the complex array of secreted signals that are involved in stem cell signaling. Use of GAG-based materials as an analytical tool to identify molecular players in stem cell signaling may be a novel paradigm for stem cell engineering and development of regenerative therapies for tissue repair.

Future work can expand on the various principles presented in this dissertation to develop precisely controlled temporal, spatial, and physical cues to enhance chondrogenic differentiation for tissue engineering repair, such as controlled release and presentation of numerous signaling molecules, inclusion of biomimetic peptides or adhesive cues, and application of mechanical strain. Additionally, while this dissertation chose to focus on CS and chondrogenic differentiation, due to the prevalence of CS matrix in cartilaginous tissues, numerous GAGs are found throughout the body for a variety of functions. GAGs such as heparin, heparan sulfate, dermatan sulfate, keratan sulfate, and hyaluronan may support similar roles, including interaction with a variety of signaling molecules, and these materials may possess significant application in strategies to promote tissue repair. The research presented in this thesis provided valuable insights into the contributions of GAG matrix in the development, maintenance, and repair of cartilaginous tissues, and these findings improve understanding of the role of local chemical, biomolecular, and overall physical environments in the development of therapies to promote tissue regeneration and repair.



# **APPENDIX** A

# SUPPLEMENTARY FIGURES

## A.1 Viability in CS-MA and Ch-MA hydrogels at 10x10<sup>6</sup> cells/mL

#### **A.1.1 Materials and Methods**

Materials and methods were performed as described in Chapter 6, with the lone exceptions being that MSCs were incorporated at  $10 \times 10^6$  cells/mL, hydrogels were cultured both in the presence and absence of 10 ng/mL TGF- $\beta$ 1 and 100 nM dexamethasone, and time points for analysis were performed on days 1, 7, 14, and 21.

#### A.1.1.1 Encapsulation of Human MSCs

Human mesenchymal stem cells (MSCs) were encapsulated in hydrogels containing a 1:1 ratio of OPF 10K:PEG-DA and 50% chondroitin sulfate methacrylate (CS-MA) or 50% chondroitin methacrylate (Ch-MA) by dry mass at a final cell concentration of  $10x10^6$  cells/mL. PEG controls containing 60% PEG-DA:40% OPF were used as swelling controls for 50% Ch-MA materials. Hydrogels were cultured for 3 weeks in basal medium composed of high glucose DMEM containing 1% ITS+ culture supplement, 1% nonessential amino acids, 50 µg/mL ascorbate-2-phosphate, and 1% antibiotic/antimycotic, or in chondrogenic medium supplemented with 10 ng/mL TGF- $\beta$ 1 and 100 nM dexamethasone.

#### A.1.1.2 LIVE/DEAD Staining for Viability

On days 1, 7, 14, and 21, hydrogels were stained for 60 minutes in LIVE/DEAD stain, and viability was imaged via confocal microscopy (n=4).



#### A.1.1.3 PicoGreen DNA Assay

PicoGreen assay was used to evaluate the total DNA content in each sample on days 1, 7, 14, and 21. Within each hydrogel formulation, DNA content of each gel was normalized to wet mass to correct for small differences in gel size (n=4).

#### A.1.1.4 Statistical Analysis

A two-factor analysis of variance (ANOVA) was used to determine statistical significance of groups, and Tukey's *post hoc* multiple comparison test with significance set at  $p \le 0.05$  indicated significance between individual samples. For DNA analysis, the factors were hydrogel type and time.

#### A.1.2 Results

LIVE/DEAD staining of human MSCs in 50% CS-MA and 50% Ch-MA hydrogels indicated that visible MSCs remained mostly viable over 3 weeks of culture (Figure A.1a). Cells remained dispersed evenly throughout the hydrogel scaffolds with a spherical shape, and no cell aggregation or spreading was observed. Total DNA content, as a measure of cell number, suggested that cellularity decreased over time in all hydrogel formulations (Figure A.1b). DNA content significantly decreased from day 1 to day 7 in CS-MA and Ch-MA hydrogels and by day 14 in PEG controls. DNA content was fairly consistent across hydrogel types from day 1 until day 14; however, PEG control hydrogels experienced a significant decrease in DNA content on day 21, compared to both CS-MA and Ch-MA.





Figure A.1. Viability and cellularity of human MSCs encapsulated in 50% CS-MA and 50% Ch-MA hydrogels at  $10\times10^6$  cells/mL in chondrogenic medium. (a) Human MSCs appeared largely viable in all hydrogel formulations over 21 days of culture. Cells remained dispersed with a rounded morphology. Scale bars = 100 µm. (b) DNA content, as a measure of cellularity, decreased over time with a significant decrease observed between day 1 and day 7 in CS-MA and Ch-MA hydrogels and by day 14 in PEG controls. Additionally, a significant decrease in DNA content was measured in PEG control hydrogels on day 21. \* indicates significantly less DNA than day 1 ( $p\leq0.05$ ). # indicates significantly less DNA than 50% CS-MA at the same time point ( $p\leq0.05$ ).

# A.2 Gene expression in the presence and absence of TGF- $\beta$ 1 at 10x10<sup>6</sup> cells/ml

#### **A.2.1 Materials and Methods**

Materials and methods were performed as described in Chapter 6, with the lone exceptions being that MSCs were incorporated at  $10 \times 10^6$  cells/mL, hydrogels were cultured both in the presence and absence of 10 ng/mL TGF- $\beta$ 1 and 100 nM dexamethasone, and time points for analysis were performed on days 1, 7, 14, and 21.



These differences in cell encapsulation and culture conditions are described in detail in Section A.1.1.1.

#### A.2.1.1 Reverse transcription polymerase chain reaction

Gene expression of encapsulated MSCs was analyzed after 1, 7, 14, and 21 days by reverse transcription polymerase chain reaction (RT-PCR). RNA was extracted and reverse transcribed, and quantitative PCR amplification for each gene target was performed for gene expression of human mRNA for collagen II, aggrecan, and SOX9 (chondrocytic markers), and for collagen X (hypertrophic chondrocyte marker). To determine fold regulation over 50% Ch-MA hydrogels in the absence of TGF- $\beta$ 1 on day 1, the raw fluorescence data was processed using LinRegPCR with glyceraldehyde-3phosphate dehydrogenase (GAPDH) as an endogenous control (n=6).

#### A.2.1.2 Statistical Analysis

PCR amplification data for each gene were first transformed using a Box-Cox transformation to obtain a normal distribution for analysis. A three-factor analysis of variance (ANOVA) was used to determine statistical significance of groups, and Tukey's *post hoc* multiple comparison test with significance set at  $p \le 0.05$  indicated significance between individual samples. For gene expression analysis, the factors were hydrogel type, media type, and time.

#### A.2.2 Results

In the presence of chondrogenic medium, human MSCs encapsulated in nonsulfated 50% Ch-MA hydrogels significantly upregulated gene expression of the cartilaginous ECM molecules collagen II and aggrecan on days 7, 14, and 21, over 50% CS-MA hydrogels (Figure A.2a-b). MSCs in 50% Ch-MA expressed 186  $\pm$  162 fold



upregulation of collagen II expression on day 21, compared to only  $0.022 \pm 0.015$  in 50% CS-MA and 9.9  $\pm$  19.2 fold in PEG controls (Figure A.2a). Similarly, aggrecan expression in Ch-MA gels was upregulated 27.6  $\pm$  9.5 fold on day 21 in the presence of TGF- $\beta$ 1, compared to 10.8  $\pm$  3.7 in CS-MA and 7.8  $\pm$  1.5 in PEG (Figure A.2b). Cartilaginous transcription factor SOX9 experienced slight upregulation on day 7 only in 50% Ch-MA with 1.57  $\pm$  0.29 fold regulation, while CS-MA only expressed 0.51  $\pm$  0.41 fold regulation (Figure A.2c). Collagen X, an ECM marker of hypertrophic chondrocytes, was also significantly upregulated in Ch-MA hydrogels over CS-MA controls in chondrogenic medium on days 7, 14, and 21 (Figure A.2d). MSCs in 50% Ch-MA exhibited large upregulation of collagen X of 4,350  $\pm$  1390 fold on day 21, compared to 655  $\pm$  371 in CS-MA.

In the absence of chondrogenic medium, MSCs encapsulated in 50% Ch-MA hydrogels did not upregulate any of the chondrogenic markers analyzed here. 50% CS-MA hydrogels, on the other hand, significantly upregulated collagen II gene expression compared to both Ch-MA hydrogels in basal medium and CS-MA gels in chondrogenic medium on days 14 and 21. MSCs in CS-MA hydrogels expressed  $69.9 \pm 19.2$  fold upregulation of collagen II on day 21 in basal medium, compared to  $1.28 \pm 1.66$  fold in Ch-MA. Aggrecan and collagen X were also upregulated on day 21 in 50% CS-MA in basal medium, compared to CS-MA in basal medium, but not compared to CS-MA in chondrogenic medium.





Figure A.2. Gene expression of chondrocytic markers by MSCs in 50% CS-MA and 50% Ch-MA hydrogels at 10x10<sup>6</sup> cells/mL in chondrogenic and basal medium. MSCs in 50% Ch-MA hydrogels significantly upregulated gene expression of (a) collagen II and (b) aggrecan on days 7, 14, and 21 in chondrogenic medium, and (c) SOX9 on day 7 only, compared to 50% CS-MA. 50% CS-MA hydrogels in basal medium, however, upregulated expression of (a) collagen II and (b) aggrecan after 21 days, compared to 50% Ch-MA. (d) Hypertrophic chondrocyte marker collagen X was also significantly upregulated in 50% Ch-MA gels on days 7, 14, and 21 in chondrogenic medium over 50% CS-MA, while 50% CS-MA in basal medium upregulated collagen X expression only on day 21, compared to 50% Ch-MA. \* indicates significantly greater than 50% CS-MA in same medium and at same time point ( $p \le 0.05$ ). # indicates significantly greater than PEG controls in same medium and at same time point ( $p \le 0.05$ ). + indicates significantly greater than 50% Ch-MA in same medium and at same time point ( $p \le 0.05$ ). § indicates significantly greater than the same hydrogel type in the other media type and at same time point (*p*≤0.05).



# A.3 Immunostaining for ECM in CS-MA and Ch-MA hydrogels at 10x10<sup>6</sup> cells/mL

### A.3.1 Materials and Methods

Materials and methods were performed as described in Chapter 6, with the lone exceptions being that MSCs were incorporated at  $10 \times 10^6$  cells/mL, hydrogels were cultured both in the presence and absence of 10 ng/mL TGF- $\beta$ 1 and 100 nM dexamethasone, and time points for analysis were performed on days 1, 7, 14, and 21. These differences in cell encapsulation and culture conditions are described in detail in Section A.1.1.1.

# A.3.1.1 Histological staining

ECM production by encapsulated MSCs was determined by immunostaining on days 1, 7, 14, and 21. For aggrecan staining, samples were deglycosylated with 30  $\mu$ l of 0.75 U/ml chondroitinase ABC for 1.5 hours. For primary antibody binding, sections were incubated in monoclonal mouse anti-human collagen I, collagen II, or aggrecan. Sections were then incubated with highly cross-adsorbed Alexa Fluor 488-conjugated goat polyclonal anti-mouse immunoglobulin G (IgG), and counterstained with 0.1  $\mu$ g/mL 4',6-diamidino-2-phenylindole (DAPI) (n=2). Negative controls were stained as described, but using a monoclonal mouse IgG1 isotype control with no known reactivity with human antigens as the primary antibody at 10  $\mu$ g/mL.

# A.3.2 Results

Immunostaining for ECM production demonstrated that while some accumulation of cartilaginous ECM collagen II and aggrecan was observed pericellularly over 21 days, staining was generally weak and clear differences were not apparent between CS-MA and



Ch-MA formulations (Figure A.3a-b). Deposition of collagen I appeared to persist over time, but few differences were observed between CS-MA and Ch-MA hydrogels (Figure A.3c). Isotype controls demonstrated little non-specific staining (Figure A.3d).



Figure A.3. Immunostaining for ECM deposition by MSCs in 50% CS-MA and 50% Ch-MA hydrogels at 10x10<sup>6</sup> cells/mL in chondrogenic medium. (a) Collagen II, (b) aggrecan, and (c) collagen I were produced in relatively small quantities by MSCs in CS-MA and Ch-MA hydrogels, and there were no distinguishable differences across gel types. (d) Isotype controls demonstrated little non-specific staining. Cell nuclei = blue. Scale bars = 100 μm.



# **APPENDIX B**

# LABORATORY PROTOCOLS

# **B.1 OPF Synthesis**

## Fumaryl Chloride (FuCl) Distillation (if necessary)

Warning: Fumaryl chloride has a very pungent odor. Work only in the fume hood with the sash down as far as possible. Double glove, and leave jars, glassware, used gloves in the hood overnight to air out.

1) Wash and dry:

2 x 500 ml round-bottom flasks, 1 x joint for thermometer, 1 x thermometer to fit

joint (~160°C), 1 x large condenser, 1 x glass elbow, 1 x joint for desiccators,

1 x glass stopper, 1 x glass funnel, 1 x egg-shaped FuCl stir bar.

2) Set up the distillation apparatus as shown, without the FuCl flask and collection flask:



 Replace the 500 ml collection flask with a 100 ml round bottom flask to collect the first 30 ml of distillate.



- 4) Clamp the apparatus to the scaffolding in the back of the fume hood.
- 5) Vacuum grease and clamp all connections.
- 6) Run cold, ice water UP the condenser tube using the circulation pump. This allows any bubbles to flow up and out of the condenser. Make sure that the circulation pump is not actively heating the water by turning the temperature control all the way down to -20°C.
- 7) Tie a KimWipe around the bottom of the condenser to catch external condensation.
- 8) Use the funnel to pour 150 ml (100 g) of FuCl into a 500 ml round-bottom flask.
- 9) Add the FuCl stir bar into the FuCl flask.
- 10) Vacuum grease the FuCl flask, and connect the flask to apparatus.
- 11) Place the flask in heating mantle with magnetic stirring at ~3.
- 12) Insulate the flask and neck with glass wool all the way up to the condensing tube to promote boiling and prevent condensation.
- 13) Start the transformer at 40 units.
- 14) Increase the transformer by 10 units, every 5 minutes.
- 15) Increase the transformer until the vapor temperature is 160°C (~80 units on transformer).
- 16) Dispose of the first ~30 ml of distillate by turning the neck up, removing the 1<sup>st</sup> 100 ml collection flask, and quickly replacing it with a clean 500 ml round-bottom flask.
- 17) FuCl distillate should be a light amber color.
- 18) The solution in the heated FuCl flask will become darker and more viscous.



- 19) When ~50-75 ml remains the heated FuCl flask and the solution is dark brown, turn off the transformer, remove the glass wool, turn the condenser off, and replace the heating mantle with a cork ring.
- 20) Stopper the FuCl distillate, parafilm the stopper, cover the flask with aluminum foil, and label it.
- 21) Store the distillate in the 4°C explosion-proof refrigerator.
- 22) Dilute the FuCl waste with tap water, and leave it in the hood overnight. Be careful of HCl production in the reaction with water, and pour the water in SLOWLY.
- 23) On the next day, use a spatula to break up the FuCl waste, retrieve the stir bar, and disposed of the waste in the aqueous waste container.
- 24) Glassware can be cleaned with acetone and the base bath.

# Methylene Chloride (MeCl) Distillation

Warning: Use nitrile or silver-shield gloves when handling MeCl.

1) Wash and dry:

1 x 1000 ml round-bottom flask, 1 x joint for thermometer, 1 x thermometer to fit

joint (~40°C), 1 x large condenser, 1 x glass elbow, 1 x joint for desiccators,

1 x glass stopper, 1 x glass funnel.

 Set up the distillation apparatus as shown, without the MeCl flask and collection flask:





- Replace the 500 ml collection flask with a 100 ml round bottom flask to collect the first 30 ml of distillate. Clamp the apparatus to the scaffolding in the back of the fume hood.
- 4) Vacuum grease and clamp all connections.
- 5) Run cold, ice water UP the condenser tube using the circulation pump. This allows any bubbles to flow up and out of the condenser. Make sure that the circulation pump is not actively heating the water by turning the temperature control all the way down to -20°C.
- 6) Tie a KimWipe around the bottom of the condenser to catch external condensation.
- Use funnel to add 750 ml of MeCl and calcium hydride (CaH<sub>2</sub>, if needed) into the existing MeCl + CaH<sub>2</sub> flask.
- 8) Vacuum grease the MeCl flask, and connect the flask to apparatus.
- The MeCl + CaH<sub>2</sub> flask already contains a stir bar. Place the flask in heating mantle with magnetic stirring at ~4.
- 10) Insulate the flask and neck with glass wool all the way up to the condensing tube to promote boiling and prevent condensation.



- 11) Set the transformer to 35 units.
- 12) Vapor temperature should increase to  $\sim 40^{\circ}$ C.
- 13) Dispose of the first ~30 ml of distillate by turning the neck up, removing the 1<sup>st</sup> collection flask, and quickly replacing it with a clean 1000 ml round-bottom flask.
- 14) When ~150-200 ml remains in the heated MeCl flask, turn off the transformer, remove the glass wool, turn the condenser off, and replace the heating mantle with a cork ring. You should have ~500 ml MeCl distillate in your collection flask (need 340-380 ml for remaining steps).
- 15) Vacuum grease the glass stopper, and stopper the MeCl distillate, and label it.
- 16) Store the anhydrous MeCl distillate in the hood overnight. Do not parafilm the stopper.
- 17) Vacuum grease the original  $MeCl + CaH_2$  stopper, stopper the undistilled  $MeCl + CaH_2$ , and store it in the back of the hood.
- 18) Dispose of the MeCl waste in the chlorinated organic solvents waste container.
- 19) Glassware can be dried in the hood, and then cleaned normally.

#### **Azeotropic Distillation of PEG**

Warning: Use nitrile or silver-shield gloves when handling toluene.

- 1) Wash and dry:
  - 1 x 500 ml or 1000 ml round-bottom flask, 1 x Dean stalk, 1 x condenser,
  - 1 x glass stopper, 1 x Kontes #2 glass valve with LARGE hole,
  - 1 x egg-shaped stir bar, 1 x glass funnel, 1 x 250 ml glass graduated cylinder.
- 2) Set up the distillation apparatus as shown, without the PEG/toluene flask:





- 3) Clamp the apparatus to the scaffolding in the back of the fume hood.
- 4) Vacuum grease and clamp all connections.
- 5) Run cold, ice water UP the condenser tube using the circulation pump. This allows any bubbles to flow up and out of the condenser. Make sure that the circulation pump is not actively heating the water by turning the temperature control all the way down to -20°C.
- 6) Tie a KimWipe around the bottom of the condenser to catch external condensation.



- 7) Weigh 50 g PEG 3.4K or 10K in the 500 ml round-bottom flask.
- Use the graduated cylinder and funnel to add 200 ml of toluene to the PEG in the 500 ml flask.
- Add the stir bar to the flask, vacuum grease the flask, and connect the flask to apparatus.
- 10) Place the flask in heating mantle with magnetic stirring at ~5. The PEG will dissolve with stirring and heating.
- Insulate the flask and neck with glass wool all the way up to the condensing tube to promote boiling and prevent condensation.
- 12) Set the transformer to 65 units.
- 13) When the Dean stalk fills to 20 ml, dispose of the toluene by draining the solution from the Dean stalk into a 250 ml waste beaker. The first few batches of waste may be partially cloudy, while others should be clear.
- 14) Repeat step 13 seven more times, removing 20 ml toluene at a time until ~160-180 ml toluene has been removed.
- 15) Turn off the transformer, remove the glass wool, turn the condenser off, and replace the heating mantle with a cork ring. 2-5 ml extra toluene waste may condense as you do this.
- 16) Allow the distilled PEG and toluene to cool, and then vacuum grease the glass stopper and stopper the distilled PEG. Parafilm the stopper, and label the flask.
- 17) Store the distilled PEG in the hood overnight. The PEG will solidify as it cools.
- 18) Dispose of the toluene waste in the nonchlorinated organic solvent waste container.
- 19) Glassware can be dried in the hood, and cleaned normally.



# **OPF Reaction**

Warning: Use nitrile or silver-shield gloves when handling MeCl.



# **3.4K Calculations:**

PEG MW = 3,400 Da

50 g PEG = 0.01471 mol PEG

1 PEG : 0.9 FuCl  $\rightarrow$  10% molar excess for PEG addition to ends of FuCl

(0.9 mol FuCl / mol PEG) \* (0.01471 mol PEG) = 0.01324 mol FuCl

FuCl MW = 153 g/mol



(0.01324 mol) \* (153 g/mol) = 2.025 g FuCl

FuCl density = 1.415 g/ml

(2.0235 g) / (1.415 g/ml) = 1.431 ml FuCl

1 FuCl : 2 TEA  $\rightarrow$  TEA removes Cl from ends of FuCl, 2 Cl per FuCl  $\rightarrow$  Twice as much

TEA as FuCl

(2 mol TEA / mol FuCl) \* (0.01324 mol FuCl) = 0.02648 mol TEA

TEA MW = 101.2 g/mol

(0.02648 mol) \* (101.2 g/mol) = 2.6798 g TEA

TEA density = 0.726 g/ml

(2.6798 g) / (0.726 g/ml) = 3.6912 ml TEA

# **10K Calculations:**

PEG MW = 10,000 Da

50 g PEG = 0.005 mol PEG

1 PEG : 0.9 FuCl  $\rightarrow$  10% molar excess for PEG addition to ends of FuCl

(0.9 mol FuCl / mol PEG) \* (0.005 mol PEG) = 0.0045 mol FuCl

FuCl MW = 153 g/mol

(0.01324 mol) \* (153 g/mol) = 0.6885 g FuCl

Density FuCl = 1.415 g/ml

(0.6885 g) / (1.415 g/ml) = 0.4866 ml FuCl

1 FuCl : 2 TEA  $\rightarrow$  TEA removes Cl from ends of FuCl, 2 Cl per FuCl  $\rightarrow$  Twice as much

TEA as FuCl

(2 mol TEA / mol FuCl) \* (0.0045 mol FuCl) = 0.009 mol TEA

TEA MW = 101.2 g/mol



(0.009 mol) \* (101.2 g/mol) = 0.9108 g TEA

TEA density = 0.726 g/ml

(0.9108 g) / (0.726 g/ml) = 1.2545 ml TEA

# Reaction

- 1) Wash and dry:
  - 1 x 1000 ml 3-arm round-bottom flask, 2 x 60 ml dropping funnels,
  - 1 x joint for the  $N_2$  balloon, 1 x PTFE valve for the  $N_2$  balloon,
  - 2 x Kontes #2 glass valves with LARGE holes, 2 x glass stoppers,
  - 1 x 250 ml or 1000 ml glass graduated cylinder.
- Use the funnel and graduated cylinder to add 320 ml MeCl to the distilled PEG/toluene. Dissolve PEG with stirring.
- Use the funnel to pour the PEG/MeCl into the 3-arm flask. The PEG distillation stir bar can be reused in this step.
- 4) Set up the reaction apparatus as shown, without the glass stoppers and  $N_2$  balloon:





- 5) Clamp the apparatus to the scaffolding in the back of the fume hood.
- 6) Vacuum grease and clamp all connections. Be careful not to vacuum grease over the holes in the valves, or else the FuCl and TEA will not flow through. Also the PTFE valve for the N<sub>2</sub> balloon does not require vacuum grease.
- Place the 3-arm flask in a small autoclave bin, filled with ice on a large stir plate.
  Optional: Add salt to the ice to keep the ice from melting.
- 8) For PEG 3.4K, use glass pipettes to add 30 ml MeCl to volumes of FuCl and TEA calculated above (3.4K: 1.431 ml FuCl and 3.6912 ml TEA). Add MeCl to dropping funnels first, then FuCl and TEA. MeCl removes the markings from glass pipettes, so exercise care when transferring MeCl.
- 9) For PEG 10K, use glass pipettes to add 10 ml MeCl to volumes of FuCl and TEA calculated above (10K: 0.4866 ml FuCl and 1.2545 ml TEA). Add MeCl to dropping funnels first, then FuCl and TEA. MeCl removes the markings from glass pipettes, so exercise care when transferring MeCl.
- 10) Flush system with N<sub>2</sub> gas, using the N<sub>2</sub> tank and hose and the N<sub>2</sub> filled balloon. Lift the glass stoppers from the dropping funnels slightly to purge excess air. Make sure the N<sub>2</sub> valve remains open.
- 11) Stir PEG solution on the stir plate at ~5.
- 12) Start reaction by dropping the FuCl and TEA at the same rate of 1 drop per ~3-4 seconds. A slower drop rate will result in a more efficient reaction.
- 13) Reaction will turn dark brown.
- 14) When necessary, siphon melted water from the ice bin, and replace the ice.



- 15) When dropping is complete, allow the reaction to continue on ice for the rest of the day.
- 16) When necessary, siphon melted water from the ice bin, and replace the ice.
- 17) At the end of the day, remove the ice bin, remove the dropping funnels, and replace the funnels with glass stoppers.
- 18) Allow the reaction to continue at room temperature with stirring on a cork ring for at least 2 days.
- 19) Check the N<sub>2</sub> balloon over the next 2 days, and refill the balloon if necessary (close valve when refilling the balloon).
- 20) Glassware can be dried in the hood, and cleaned with acetone and the base bath.

# **Rotovaporing of MeCl**

Warning: Use nitrile or silver-shield gloves when handling MeCl.

- 1) Wash and dry:
  - 1 x 1000 ml round-bottom flask,
  - 1 x glass funnel.
- Turn on Rotovapor by switching Vacuum Controller V-800, Rotovapor R-200, and Vacuum V-500 on.
- 3) Fill the water bath with distilled H2O, and heat to  $40^{\circ}$ C.
- 4) Use the circulating pump to flow cold, ice water through the condensing tube.
- 5) Use the funnel to pour the OPF solution into a 1000 ml round-bottom flask.
- 6) Clamp and vacuum grease the flask to the Rotovapor.
- Lower the flask into the water bath, so the OPF solution and water levels are the same.



- 8) Slowly rotate the flask in the water bath.
- 9) Turn on the vacuum on at 850 mbar ("Set"  $\rightarrow$  Up or down  $\rightarrow$  "Run").
- 10) Gradually decrease the vacuum as necessary to maintain a steady drip of condensation into the collecting flask. Vacuum can be decreased as low as 700 mbar.
- When OPF/MeCl solution is thick and "stew-like" consistency, remove OPF from Rotovapor.
- 12) Dispose of the MeCl waste in the chlorinated organic solvents waste container.
- 13) Glassware can be dried in the hood, and cleaned normally.

# Wash in Ethyl Acetate

Warning: Use nitrile or silver-shield gloves when handling ethyl acetate.

- 1) Wash and dry:
  - 2 x 2 L aspiration flasks, 1-2 x 2 L Erlenmeyer flasks, 1-2 x 2 L beakers,

2 x Buchner funnels, 1-2 x glass funnels, 1 x stir bar, 2 x spatulas.

- 2) Add ethyl acetate (EA) to the OPF solution until the flask is  $\sim 2/3$  full.
- Stir the solution while heating with the heatgun (low speed, med heat) for 15-20 min, rotating every 5 minutes.
- 4) Solution will become less viscous and salts become visible at the surface. Ethyl acetate is a solvent for the OPF, but not for the salts produced in the TEA reaction.
- Connect the vacuum and filter the solution through a Buchner funnel with #1
  Whatman filter paper (11 μm pores) into a 2 L aspiration flask.
- 6) The salts will be filtered out of the solution by the filter paper. Discard these salts.
- 7) Add EA to the OPF to a total volume of 1500-1700 ml.



- Stopper the flask with a #9 rubber stopper and parafilm the stopper and aspiration neck, and place the flask into the -20°C explosion-proof freezer for at least 2 hours or overnight if necessary.
- 9) If necessary, clean glassware for the next step.
- 10) After cooling, remove the OPF/EA from the freezer. The decreased temperature alters the solubility of the OPF, causing the OPF to precipitate out.
- 11) Connect the vacuum and filter the solution through a Buchner funnel with #1Whatman filter paper, capturing the OPF in the filter paper and pulling the EA into an aspiration flask.
- 12) While filtering, stir the solution, allowing the EA to be pulled through the filter paper.Discard the EA.
- 13) When nearly dry, transfer the OPF from the filter paper to a beaker.
- 14) Add 1 L ethyl acetate to the solution.
- 15) Stir the solution while heating with the heat gun to redissolve the OPF in the EA.Solution goes from light brown to dark brown and becomes less viscous.
- 16) Use a clean glass funnel to transfer the OPF and EA to a 2 L Erlenmeyer flask and add EA to a total volume of 1500-1700 ml.
- 17) Stopper the flask with a #10 rubber stopper, and recrystallize the OPF/EA solution in a 2 L aspiration flask at -20°C for 1.5 hours or overnight.
- 18) Dispose of the EA waste in the nonchlorinated organic solvent waste container.
- 19) After cooling, filter the solution through a clean Buchner funnel with #4 Whatman filter paper (20-25 μm pores), capturing the OPF in the filter paper and pulling the EA into an aspiration flask.



- 20) The second filtering step may proceed much more slowly than the first. If EA/OPF is solid after removing from the freezer, add more EA to the solution. Tape 2 spatulas together to scoop product from the bottom of the flask.
- 21) Dispose of the EA waste in the nonchlorinated organic solvent waste container.
- 22) Optional: Repeat steps 13-19 for a third filtering step. This may be necessary if OPF appears too dark.
- 23) Dispose of the EA waste in the nonchlorinated organic solvent waste container.
- 24) Glassware can be dried in the hood, and cleaned normally.

## Wash in Ethyl Ether

Warning: Use nitrile or silver-shield gloves when handling ethyl acetate and/or ethyl ether.

1) Wash and dry:

1 x 2 L beaker, 1 x stir bar, 1 x Buchner funnel, 2 x PTFE coated jars.

- When OPF is nearly dry and EA is mostly gone, add 1 L ethyl ether (EE) directly to the funnel to remove the EA.
- 3) Once mostly dry, transfer the OPF from the funnel and filter paper to a 2 L beaker.
- 4) Add 1 L ethyl ether (EE) to the OPF for a second wash with stirring.
- 5) Filter the solution through a Buchner funnel with #4 Whatman filter paper (20-25 μm pores), capturing the OPF in the filter paper and pulling the EE into an aspiration flask.
- 6) Optional: Add EE to the OPF for a third wash.
- As the EE filters through and the OPF dries, use a spatula to break up any clumps of OPF.


- 8) By the end, you should have a fine powder that is mostly dry.
- 9) Scoop the OPF powder evenly into 2 Teflon-coated jars with the spatula.
- 10) Leave the OPF in the hood overnight with the lid on loosely.
- 11) Dispose of the EE waste in the nonchlorinated organic solvent waste container.
- 12) Glassware can be dried in the hood, and cleaned normally.

#### Vacuum Dry OPF

1) Clamp a lyophilizer tube to the scaffolding in the back of the fume hood as shown:



- Tape aluminum foil to the top of the OPF jars and poke holes in the foil with a small gauge needle.
- 3) Add liquid nitrogen to the solvent trap.
- 4) Connect the vacuum to the solvent trap.
- 5) Close the valves to the samples (3<sup>rd</sup> and 4<sup>th</sup> from the top). Open the valve to the manometer (bottom).
- 6) The top 2 valves are open to the atmosphere and should remain closed.
- 7) Turn on manometer, and then the vacuum pump.
- 8) Once a vacuum is established, gradually open the valves to the samples. Open the valve to the first sample until a vacuum is established, then close that valve. Then



open the valve to the second sample until a vacuum is established. Now you can reopen the first sample.

- 9) Pressure should be less than 10 mbar (preferably 1-2 mbar).
- 10) Check the pressure and the liquid nitrogen level every 30-45 minutes.
- 11) Refill the liquid nitrogen if necessary.
- 12) When OPF is dry and you cannot smell any EE in the powder, open the valve to break the vacuum and turn off the pump.
- 13) Dispose of solvent from the solvent trap.
- 14) Parafilm OPF and store it at -20°C.

# **Verify Product**

- 1) A lighter brown color is preferred.
- Polymerize a 100% OPF hydrogel with thermal and photo-initiation to test crosslinking.
- 3) Run GPC on the OPF in chloroform to verify molecular weight.

# **B.2 PEG-DA Synthesis**

# **Reaction Calculations**

- 1) Begin with **24 g PEG, MW 3400**.
- 2) React with 100% excess acryloyl chloride (AcCl; 2 AcCl:1 PEG).
- 3) 24 g PEG / (3400 g/mol PEG) = 7.06 mmol PEG
- 4) 7.06 mmol PEG \* 2 end groups \* 2 (100% excess) = 28.24 mmol AcCl
- 5) 0.02824 mol AcCl \* 90.51 g/mol / (1.114 g/mL) = 2.294 mL AcCl
- 6) React with 1:1 AcCl:triethylamine (TEA)
- 7) 0.00706 mmol PEG \* 2 end groups = 0.0141 mol TEA



8) 0.0141 mol TEA \* 101.9 g/mol / (.726 g/mL) = **1.982 mL TEA** 

9) Workup with anhydrous potassium carbonate  $(K_2CO_3)$ 

10) 0.02824 mol AcCl \* 2 mol K<sub>2</sub>CO<sub>3</sub>/mol AcCl / 2 M K<sub>2</sub>CO<sub>3</sub> = 28.2 mL 2M K<sub>2</sub>CO<sub>3</sub> (aq)

11) 138.205 g/mol = 276.41 g in 1L = 27.641 g in 100 mL = 8.2923 g  $K_2CO_3$  in 30 mL ddH<sub>2</sub>O

#### Day 1 – Methylene Chloride Distillation

Distill MeCl following the instructions in the OPF synthesis protocol. Keep in mind that you only need approximately 100 mL. Distillation is necessary to remove aqueous contamination (make MeCl anhydrous) that might generate unwanted side reactions in Day 2.

#### Day 2 – Reaction

*What's going on?* PEG is being acrylated. TEA acts as a catalyst by sequestering HCl to allow the reaction to proceed to completion. MeCl is the solvent used for this reaction. **Caution**: AcCl doesn't smell as bad as FuCl, but it is worse for you (eye, throat irritant)!

- Set up 3-arm round bottom flask in the fume hood on a stir plate. Weigh and add PEG to the flask. Add stirbar.
- Attach one dropping funnel, with a glass stopper, and a PTFE valve for N<sub>2</sub> gas flow. Vacuum grease glass-glass connections for the dropping funnel (excluding glass stopper) and the PTFE valve ites

| Equipment:            | 1x balloon       |
|-----------------------|------------------|
| 1x 500 mL 3-arm round | Glass pipettes   |
| bottom flask          | 1x stir plate    |
| 1x dropping funnel    | 1x clamp         |
| 1x PTFE gas valve     | 6x green clamps  |
| 3x glass stoppers     |                  |
| 1x glass stopcock     | Chemicals:       |
| 1x PTFE stopcock      | MeCl (anhydrous) |
|                       | PEG              |
| 1x stirbar            | TEA              |
| 1x glass funnel       | AcCl             |
| 1x graduated cylinder |                  |

valve. Do not grease the PTFE valve itself.

3) Hook up N<sub>2</sub> tubing to the valve. Continually purge the whole system as you add 40 mL MeCl to the round bottom flask through the ungreased arm using a glass funnel. Stir. Gently float a glass stopper in the arm on the air being pushed out.



- 4) When dissolved, use a glass pipette to add TEA. Vacuum grease a glass stopper and gently float the dropping funnel's glass stopper as before. Let stir for 5 min.
- 5) Use a glass pipette to add 20 mL MeCl and the appropriate amount of AcCl to the dropping funnel. Be aware that MeCl (and its fumes) will remove markings from glass pipettes. Vacuum grease and stopper the funnel while you are adding and turn off N<sub>2</sub> flow.
- 6) Remove the  $N_2$  hose and attach an  $N_2$  balloon.
- Drip the AcCl/MeCl mixture into the round bottom flask (about 1 drop every 3-4 seconds).
  Drip AcCl in MeCl very slowly (about 1 drop every 4 seconds).
- 8) After dripping is complete, you can replace the funnels with stopcocks.
- 9) Let stir overnight.

#### Day 3 – Workup

*What's going on?* To remove TEA-HCl, we first react the mixture with potassium carbonate to produce KCl, which will transfer to the aqueous phase. Some TEA will remain in the organic phase for Day 3 filtration.

 Use a glass funnel to transfer the mixture from the round bottom flask to a separatory funnel with a greased stopcock.

(Don't forget to close the stopcock prior to transfer. Also, remember to wipe vacuum grease from connections before pouring.)

- 2) Add appropriate amount of  $2M K_2CO_3$  to the separatory funnel.
- Stopper funnel, hold vertically, and give it a quick shake or two. Immediately open the stopper to release CO<sub>2</sub>. Repeat a few times.
- 4) Hold the separatory funnel horizontally, but with the tip tilted higher. One hand should hold the glass stopper, the other holding the stopcock knob up. Rotate vigorously, and open the



Equipment: 1x 250 mL separatory funnel 1x glass stopper 1x glass funnel 1x glass stopcock 1x ring clamp Chemicals:

K<sub>2</sub>CO<sub>3</sub>, anhydrous

stopcock periodically to release  $CO_2$ . Repeat until all gas is released. Solution should have the consistency of a milky-white emulsion.

5) Leave overnight. Place a beaker underneath to capture any leaked product.

#### **Day 4 – Filtration and Drying**

*What's going on?* We isolate the mixture from KCl in the aqueous phase, add MgSO<sub>4</sub> to remove any additional aqueous solution, and precipitate PEG-DA in ethyl ether. TEA should remain in solution.

- 1) Drain the two organic phases into a 250 mL beaker on a stir plate with stir bar.
- 2) While stirring, add MgSO<sub>4</sub> until the mixture goes from a lumpy consistency to a well dispersed mixture of powder and organic solvent. It should appear as opaque milk if it doesn't look like milk, add more MgSO<sub>4</sub>. Add ~20-40 mL MeCl to keep the PEG-DA in solution (may help to have a smaller bottle or beaker with MeCl). The goal here is to add as little MeCl as possible to keep the solution saturated with

PEG-DA. But, if you add too much, no sweat.

- Prepare a Buchner funnel with aspiration flask and filter paper and pre-wet the filter paper with MeCl.
- Pour the mixture into the filter and a clear liquid should be collected (containing PEG-DA). If the liquid is cloudy, filtration should be performed again. The vacuum will also begin to evaporate MeCl. Thus, you can elect to evaporate MeCl if necessary.
- *Equipment:* 2x 250 mL beakers 1x 1L aspiration beaker 1x 2-3L beaker 2x stir bars 2x Buchner funnels Filter paper (fine pores, 42) *Chemicals:* MgSO<sub>4</sub>, MeCl

Ethyl ether

 Prepare a 2L beaker with 1.7L ethyl ether and a stir bar. Pour in the PEG-DA solution and wait 10 min to precipitate PEG-DA.

6) Prepare another Buchner funnel with two filter paper sheets, and pre-wet with ethyl ether.



- Filter to separate PEG-DA. If the filtrate is not clear, re-filter. Pound into bits. See an older grad student to measure your performance.
- Dry under vacuum until no ether can be smelled (at least 5 hours). At least some of this vacuum drying must be performed immediately after filtration.
- 9) Store at -20°C. Expected recovery is approximately 75%.

Note: If necessary, you may need to dialyze the resulting product to remove impurities.

Use a 1000 MWCO dialysis membrane at 0.2 g/mL and lyophilize after dialysis.

# **B.3** Desulfation of Chondroitin Sulfate with Acidic Methanol to Form Chondroitin

Adapted from: Schubert M. Chondroitin From Chondroitin Sulfate. Methods Carbohyd

Chem. 1965;5:109-10.

# Materials:

Chondroitin-6-sulfate – Wako 032-14612, 25 g

Acetyl chloride – Fisher AC21947-2500, 250 ml

Can scale quantities/volumes accordingly.

# Desulfation of Chondroitin Sulfate to Form Methyl Ester of Chondroitin

- 1) Slowly add 5 ml acetyl chloride in 1000 mL methanol (very acidic, pH 0.0)
- 2) Let the solution stir for a few hours
- 3) Add 5.0 g chondroitin sulfate in 1000 ml of acidic methanol
- 4) Stir for 1 day at room temperature
- 5) Centrifuge (4000 RPM for 5 min) and discard clear solution
- 6) Repeat on days 3 and 7
- 7) Dissolve white residue in 100 mL water
- 8) Precipitate in 600 mL 95% ethanol



- 9) Separate by centrifugation
- 10) Wash with ethanol and separate by centrifugation at least twice
- 11) Wash with ether and separate by centrifugation at least twice
- 12) Vacuum dry
- 13) Yields ~2.8 g methyl ester of chondroitin
- 14) Store product at 4°C

#### **Demethylation to Form Chondroitin**

- 1) 2.5 g methyl ester of chondroitin in 100 mL 0.1M aqueous KOH
- 2) Keep 1 day at room temperature on shaker plate
- 3) Add to 1 mL glacial acetic acid, 1 g potassium acetate in 10 ml water
- 4) Precipitate product with 400 ml ethanol
- 5) Separate by centrifugation (4000 RPM for 5 min)
- 6) Wash with ethanol and separate by centrifugation at least twice
- 7) Wash with ether and separate by centrifugation at least twice
- 8) Vacuum dry
- 9) Yields ~2.3 g chondroitin
- 10) Store product at 4°C

#### B.4 Chondroitin Sulfate Methacrylamide Synthesis with EDC/NHS

Purpose: To add methacrylamide group to chondroitin sulfate chains.

#### Materials:

Small glass vial and small stir bar

Chondroitin sulfate A, 1 mg (Sigma)



N-(3-dimethylpropyl)-N-ethylcarbodiimide hydrochloride (EDC)

N-3-aminopropyl methacrylamide

NaCl

Dialysis tubing

Dialysis clips

# Methacrylation

 Add 1.14 mg chondroitin sulfate (~0.0025 mmol by -mer MW, 457.4 Da) per mL of distilled water.

Add 0.96 mg (0.005 mmol) of EDC and 0.89 mg (0.005 mmol) of N-3-aminopropyl methacrylamide for every 1 mL of solution.
 Optional: Add 1.09 mg (0.005 mmol) of Sulfo-NHS.

- 3) Incubate the mixture for 2 hrs at pH 5.0 with stirring at room temperature.
- Add another 0.96 mg (0.005 mmol) of EDC and 0.89 mg (0.005 mmol) of N-3aminopropyl methacrylamide for every 1 mL of solution.
- 5) Incubate the mixture for 2 more hrs at pH 5.0.

# Dialysis

- 1) Dialyze (1,000 MW cutoff) solution against 10 mM NaCl solution for 1 day.
- Cut the dialysis tubing into strips a little shorter than the beaker that will be used for dialysis.
- 3) Since the dialysis tubing is stored in sodium azide it needs to be rinsed. To do this put the dialysis tubing in a beaker with about 1L of dH<sub>2</sub>O and mix. Refresh the dH<sub>2</sub>O every 15 minutes (3x) in order to thoroughly rinse the tubing.
- 4) After the tubing is thoroughly rinsed remove it from the beaker and close one end of



the dialysis tubing with a weighted closure. After the reaction in step 5 is complete, use a transfer pipette to add the mixture to the tubing until the tubing is about half full. Close the open end of the tubing with a non-weighted closure.

- 5) Use teflon tape to hold the tubing high in the water as it stirs.
- 6) Dialyze (1,000 MW cutoff) solution against distilled water for another 2 days.
- If starting on the morning of the first day change the water the morning and night of the second day.
- 8) Lyophilize product for 4 days.

# **B.5** Chondroitin Sulfate Methacrylate Synthesis with Methacrylic Anhydride Purpose:

To conjugate methacrylate groups onto a chondroitin sulfate polymer that enable photopolymerization into a cohesive CS-MA hydrogel that maintains degradative properties. The goal of this experiment is to create CS-MA with various degrees of methacrylate (MA) substitution by reacting chondroitin sulfate with methacrylic anhydride (MAA) in the presence of sodium hydroxide (NaOH).

#### **References:**

- Bryant SJ et al. Synthesis and characterization of photopolymerized multifunctional hydrogels: water-soluble poly(vinyl alcohol) and chondroitin sulfate macromers for chondrocyte encapsulation. *Macromolecules*. (2004). 37:6726-6733.
- Wang L-F et al. Synthesis and characterization of chondroitin sulfate-methacrylate hydrogels. *Carbohydrate Polymers*. (2003). 52:389-396.



- Huang S-J et al. Controlled immobilization of chondroitin sulfate in polyacrylic acid networks. *Journal of Biomaterials Science. Polymer Edition*. (2007). 18(1):17-34.
- Tsai M-F et al. Characterization of hydrogels prepared from copolymerization of the different degrees of methacrylate-grafted chondroitin sulfate macromers and acrylic acid. *Journal of Biomedical Materials Research. Part A.* (2008). 84A:727-739.

**Reaction Chemistry:** 



The synthesis of CS-MA is shown above. Any of 3 hydroxyl groups in each disaccharide unit of CS may under nucleophilic acyl substitution via a carbonyl group in methacrylic anhydride to form an ester bond, producing a methacrylic acid by-product. The reaction is driven forward by the addition of excess strong base (NaOH) to neutralize the methacrylic acid (pKa = 4.58) that is formed (Le-Châtelier's principle). Since the solubility of highly charged CS is limited in organic solvents, this methacrylation scheme is carried out in an aqueous environment MAA. To compensate for the tendency of MAA to react with water and form methacrylic acid, an excess of MAA is used in all cases. **Reagents:** 

chondroitin sulfate A (Chondroitin 4-sulfate, C 9819, Sigma)



www.manaraa.com

- Storage temperature 2-8°C
- Approximate MW 48,700 Da and PI 1.49 per Bryant et al, MW of

disaccharide unit = 458.37 Da

- ddH<sub>2</sub>O
- methacrylic anhydride (Product # 276685-500mL, Aldrich)
  - Stored liquid at RT, acids cabinet
  - Use respiratory and contact precautions, wear eye protection
  - Must be added dropwise to aqueous solutions
  - Assay 94%; MW 154.16; density 1.042 g/mL at 25 °C; FP 84°C, BP 87°C
- 5N NaOH
  - Stored solid pellets at RT, bases cabinet
- Methanol
  - Stored 4°C, explosion-proof refrigerator
- D<sub>2</sub>O

# **Equipment:**

- 100mL beaker and stir bar for dissolving CS
- 500mL, 3-neck round-bottom flask for reaction, thermometer fitted into glass stopper for flask, 125mL dropping funnels (2), glass valves and clips (2), glass stoppers and clips (3)
- Heating mantle, magnetic stir plate and stir bar
- Pipette pump and glass serological pipettes
- Silicone vacuum grease
- Cork ring



- 2L Erlenmeyer flask
- Large Buchner funnel and 125mm diameter #42 Whatman quantitative cellulose filter membranes, 1L filtration flask
- High vacuum system, lyophilizer tube (1), liquid nitrogen, scintillation vials (1), aluminum foil, syringe needle

# **Procedure:**

# Day 1

#### Preparation

- Clean stir bars and any glassware. For glassware, begin with soap and water, followed by dH<sub>2</sub>O, and then acetone. For plastics, use ethanol in lieu of acetone. Dry in oven or at room temp.
- 2) Make solution (>150mL) of 5N NaOH (NaOH MW: 40.00).
- 3) Prepare scintillation vials that will contain solution for pH readings.
- 4) Bring CS to room temperature (~20min).
- Place small heating mantle on stir plate inside the fume hood and connect power to nearby transformer. Plug in transformer and stir plate into hood outlet. *Conjugation*
- 6) Dissolve 15g chondroitin sulfate in 60mL ddH<sub>2</sub>O (yields a volume of ~100mL) using a beaker. This will take ~30-45 min at moderate stir velocity. Transfer to 3-neck round bottom flask and place on heating mantle on stir plate inside fume hood, setting at speed 7. Begin stirring. The flask now contains a 25% w/v solution of chondroitin sulfate, which has  $19.062 \times 10^{-3}$  moles of hydroxyl groups that are available for conjugation.



- 7) Cap the flask with a rubber stopper fitted with a thermometer that is submerged in the reaction solution (left-most neck of the flask). Switch on the transformer at setting 0 and raise the temperature to 60°C (~ transformer setting of 30).
- 8) Assemble dropping funnels. Place an *extremely thin layer* of silicone vacuum grease on either side of the hole on the glass valve, being careful to avoid getting any in the hole. Place the valve in its corresponding slot on the dropping funnel and fix in place with a metal clip on the other side. Mount a dropping funnel on the middle neck of the flask after coating the male end with a very thin layer of vacuum grease. For now, cap the right-most neck with a glass stopper that has also been coated with a very thin layer of vacuum grease.
- 9) Using a manual pipette pump and glass serological pipette, slowly add specified volume of methacrylic anhydride via the top end of the dropping funnel, making sure the valve is in the closed position first. Cap the dropping funnel with a glass stopper (coated with vacuum grease) until intended time of addition.
- 10) Once the temperature of the reaction vessel has stabilized, add methacrylic anhydride in specified amount in the table below to the flask in a dropwise fashion via the dropping funnel, making sure that the rate of dropping is slow and consistent over time until all MAA has been added. To enable a slow addition, partially open the valve.
- 11) Replace the MAA dropping funnel with one containing 5N NaOH. [Note: This is a highly corrosive base! Wear silver nitrate gloves and face shield!] Add 5N NaOH dropwise (consistent rate) into the flask to reach a pH ~10. Measure and record the pH of the reaction vessel by drawing 10mL of solution after mixing from the



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| Volume<br>Methacrylic<br>Anhydride<br>(mL) | Moles<br>Methacrylic<br>Anhydride | MAA:CS-<br>Hydroxyl<br>Ratio | Volume 5N<br>Sodium<br>Hydroxide<br>Added (mL) | Moles<br>Sodium<br>Hydroxide<br>Added | NaOH:MAA<br>Ratio |
|--|-----------------------------------|------------------------------|--|---------------------------------------|-------------------|
| 60   | 405.6 x 10 <sup>-3</sup>          | 20.6:1                       |  | x 10 <sup>-3</sup>                    | :1                |

rightmost neck of the flask. Once the appropriate pH is reached, cap the dropping funnel and all open necks with glass stoppers until next measurement/adjustment.

12) Stir reaction vigorously at 60°C for 24h under fume hood. Extract 10mL of reaction volume to pH at start of reaction, and again after 2, 4, 8 and 24h. Maintain pH at ~10 as necessary.

#### Day 2

#### Preparation

- 13) Prior to setting up equipment, make sure *all* glassware and related equipment is washed and dried. For glassware, begin with soap and water, followed by dH<sub>2</sub>O, and then acetone. For plastics, use ethanol in lieu of acetone. Dry in oven or at room temp. Weigh scintillation vials that will be used to later store CSMA (will be used to calculate an approximate percent yield).
- 14) Set up filtration apparatus: a) Place filter membrane in Buchner funnel and mount in neck of 1L filtration flask; b) Attach plastic tubing to connection in neck of flask and secure to vacuum nozzle in hood; c) secure to mounted clamp in back of hood to stabilize flask.

# Precipitation, Filtration, Washing, and Drying

15) Turn off transformer, cap all necks of flask with glass stoppers, and (using thermal gloves) place reaction flask on cork ring to allow reaction to cool to room temperature. Carefully pour reaction through a single neck (wipe vacuum grease off



with a Kimwipe first) into a 2L Erlenmeyer flask. Add ~500mL of methanol to flask to precipitate CS-MA and remaining CS. Following complete precipitation, thoroughly mix the solution and pour into filtration funnel in increments. Wash the resulting filtrate 5x with 50mL methanol (10mL methanol:1g CS reacted). Transfer filtrate to scintillation vial, cover with aluminum foil secured with tape, and poke several small holes in foil with small gage needle.

- 16) Place vial inside glass lyophilizer tube and connect tube to a vacuum pump via a liquid N<sub>2</sub> condenser trap.
- 17) Establish vacuum and dry for ~12h or until no methanol is detected by wafting.[Note: Vacuum must be monitored every 30min while running and cannot be left unattended longer than 1h, thus cannot be left on overnight. Doing so presents a significant explosion hazard.]
- 18) Store vials at 4°C.

#### *Characterization by* <sup>1</sup>*H NMR*

- 19) Rinse NMR glass tubes with acetone and dry in oven.
- 20) Prepare 10mg/mL sample of CS-MA in D<sub>2</sub>O in microcentrifuge tube and transfer to NMR tubes (~1mL). Cap and number tubes with sample identifier. In addition, prepare a D<sub>2</sub>O blank, a CS control, and a MAA control in NMR tubes.
- 21) Perform <sup>1</sup>H NMR and analyze data for the following peaks:
  - a) Residual MAA or MA acid: Integrate peak corresponding to methyl groups on MAA is observed at 1.6ppm



- b) Confirmation of MA substitution: Integrate peaks corresponding to methylene portion of conjugated methacrylate moiety at 5.65 and 6.10ppm. Peak corresponding to methyl portion of conjugated methacrylate moiety at 1.89ppm.
- c) Determination of degree of substitution: Expand region from 1.6 to 2.1ppm, then deconvolute and integrate peaks at 1.89 (see above) and 1.99ppm (methyl group of native CS). Calculate ratio of peak intensity at 1.89ppm to that at 1.99ppm to calculate degree of MA substitution on CS. [Since there are 3 hydroxyl groups that could be substituted by methacrylate, maximum DS is 3. According to Bryant et al, a heteronuclear shift correlation through multiple bond connectivities (HMBC) experiment (analyzes relationship between protons and carbons in structure) did not reveal a preferred substitution site.]

#### B.6 Chondroitin Sulfate Methacrylate Synthesis with Glycidyl Methacrylate

1) Dissolve 1 g of chondroitin sulfate (CS) per 50 mL dH<sub>2</sub>O

CS disaccharide: 457.4 Da (Chondroitin disaccharide: 378.35 Da)

1 g = 2.19 mmol = 1 equivalent

- Add 50 mL acetone, resulting in a 1% w/v solution of CS in 50:50 mixture of acetone:water
- 3) Stir overnight at room temperature
- 4) Add 20 molar equivalents of triethylamine (TEA)

TEA: 101.19 Da, 0.726 g/mL

20 equivalents = 43.7 mmol = 4.42 g = 6.09 mL

5) Slowly drip in 20 molar equivalents of glycidyl methacrylate (GMA)



GMA: 142.15 Da, 1.042 g/mL

20 equivalents = 43.7 mmol = 6.22 g = 5.97 mL

- 6) Stir for 24 hr at room temperature
- 7) Rotovap to concentrate if necessary
- Dialyze in 50:50 acetone:water for 1 day using 1,000 Da MWCO regenerated cellulose membrane
- 9) Dialyze in  $dH_2O$  for 2 days
- 10) Lyophilize product for 4 days
- 11) Store product at -20°C

#### **B.7** CSMA Microparticle Formation

Remove CSMA from -20°C and allow to sit at room temperature for 30 minutes. Keep ALL ingredients on ice.

- 1) Place 60 mL of corn oil in the 4°C refrigerator.
- 2) Make APS solution in 1.5 mL tube: 34.2 mg APS in 500  $\mu$ L PBS
- 3) Make TEMED solution in 1.5 mL tube: 22.5  $\mu$ L in 500  $\mu$ L PBS
- Weigh out 55.6 mg Chondroitin Sulfate Methacrylate (CSMA) in a scintillation vial and add 440uL of PBS.
- 5) Homogenize corn oil at 3900 rpm for 5 minutes.
- Add 30 uL of APS and 30 uL of TEMED to the CSMA solution, mix, and add solution dropwise to corn oil. Make sure everything is on ice to prevent crosslinking. Homogenize for 5 minutes.



# \*DO NOT MIX APS AND TEMED TOGETHER BEFORE ADDING TO CSMA SOLUTION-ALWAYS ADD SEPERATELY

- 7) Place corn oil mixture on a hot plate (set to 100°C) with the largest stir bar that will fit in the container (to keep particles moving so as not to crosslink with one another) and a thermometer (to monitor temperature-I usually let it go up to 50-60°C). Allow to crosslink for 30 minutes.
- Separate mixture into two 50 mL conical tubes and centrifuge at 1500 rpm for 5 minutes at 4°C.
- Remove oil from microparticles and resuspend pellet in ddH2O and transfer to 1.5 mL tube (if possible). Wash 3 times to remove oil. Store at 4°C.

# **B.8** Passive Adsorption of Aggrecan to TCPS Sufrfaces

# Materials:

Aggrecan (Sigma)

Sterile D-PBS (w/o Ca<sup>2+</sup> or Mg<sup>2+</sup>)

24-well tissue culture-treated polystyrene plates

# **Protocol:**

Aggrecan pre-treatment

- 1) Sterile filter a 0.05 mg/mL aggrecan solution in PBS.
- 2) Warm up the aggrecan solution to 37°C.
- Dispense 100 μL of aggrecan solution into each well of a 24-well tissue culturetreated polystyrene plate. This gives each well 5 μg of aggrecan.



4) Leave plates, with the lids askew, in the sterile hood overnight (with the blower running, UV lights off) to allow the D-PBS to evaporate and aggrecan to adsorb.

#### **B.9** Bovine BMSC Harvest

**Purpose:** Isolation of bMSCs from a bovine tibia and femur bone.

#### Materials:

Saw + saw blade

Tweezers, Spatulas

MSC media - Low glucose DMEM, FBS, Antibiotic/Antimicotic, bFGF

PBS+ solution – Sterile PBS, Antibiotic/Antimicotic

PBS+ bottle

Acetic Acid

Trypan blue

Isopropyl alcohol (for Mr. Frostys)

\*\* Contact Kim x51547 in IBB to make sure there's enough room for our bag in the meat

freezer\*\*

# **Protocol:**

- Make sure to autoclave all appropriate tools (saw blade, tweezers, spatulas, PBS+ bottle).
- 2) Make up MSC media.

500 ml Low glucose DMEM

56.18 ml FBS

5.62 ml Antibiotic/Antimycotic



562 ul Fungizone

 $56.2 \ \mu l \ bFGF$ 

3) Make up PBS+

500 ml Sterile PBS

5 ml Antibiotic/Antimicotic

- 4) Prepare the hood area by laying down sterile drapes and taping them down.
- 5) Cut away all tissue from the leg. This can be done non-sterilely outside of the hood.
- Bring the bone into the tissue culture hood. Using a bone saw cut the bone into two roughly equal pieces. Squirt the cut with PBS+.
- 7) Remove bone marrow using tweezers and spatulas and transfer into a 50 ml tube.
- 8) Squirt PBS+ solution into the tube.
- Pippette up and down in a 50 ml pipette until the solution starts to flow smoothy. Tapping the pipette on the bottom of the tube as pipetting will help to break up clumps faster.
- 10) Repeat using a 10 ml pipette.
- 11) Run the solution through needles of 16 G, 18 G, 20 G. If the needle becomes clogged try pulling back on the plunger and then adding pressure again. If there is still an obstruction remove the needle and replace.
- 12) Repeat previous step.
- 13) Spin down at 1200 rpm for 15 min.
- 14) Aspirate supernatant, not blood.
- 15) Pool remainder and spin down if needed.



- 16) Add PBS+ to bring the solution up to a desired level. Take a 1 ml sample for counting and spin down.
- 17) Make a 4% acetic acid soln. by mixing 80 uL acetic acid in 2 mL sterile PBS.
- 18) Add 100 uL 4% acetic acid solution to 100 uL cell suspension to lyse RBCs
- 19) Add 175 µl media, 100 µl trypan blue, and 25 µl of the cell solution/acetic acid to a tube for counting. This creates a 1:12 dilution. The cells can be spun down in the centrifuge while the counting process in taking place.
- 20) Inject approximately 10-15  $\mu$ l of the cell solution into the groove of the hemocytometer.
- 21) Look at the hemocytometer under the microscope and count the cells in the 4 sections at the corners of the grid.
- 22) Resuspend in DMEM for a concentration of 40e6 cells/25 ml of media (1.6e6 cells/ml).

Calculations

- (Total number of cells in the 4 grids) x (2500) = Concentration of solution in cells/ml. Multiply this number by 12 to get the concentration of the full solution.
- Multiply the concentration by the volume in the tube to get the total number of cells.
- 23) Add 40e6 cells in 25 ml of media to a 150x25 mm dish and leave in incubator for 30 min. Neutrophils and fast adhering cells will stick to the plate. MSCs won't.
- 24) Add 12.5 ml of fluid from the plate and 12.5 ml of media to a T-150 flask. Tilt the flask to make sure the fluid is reaches all corners of the flask and cells get distributed somewhat evenly.



- 25) Incubate overnight x2 (for two days).
- 26) Aspirate media from the T-150s.
- 27) Wash in 10 ml PBS.
- 28) Aspirate the PBS.
- 29) Repeat the previous two steps.
- 30) Add 20-25 ml of media to the flasks.
- 31) Grow till confluency (~10 days).
- 32) Freeze down cells (use Mr. Frosty!). Note: You will probably have less than the 20e6 cells/plate that were first plated.

Dispose of leg as appropriate. Double bag the remains in biohazard bags and take it to

IBB 0230 (basement) before 4 pm and ring the doorbell for assistance.

#### **B.10** Bovine Tendon/Ligament Fibroblast Harvest

**Purpose:** Tendon and ligament will be digested to remove matrix (i.e. collagen) and isolate the fibroblasts for use in experiments. Approximately 10g of tendon and ligament tissue can be obtained with each harvest, resulting in around 100e6 cells.

# Materials Needed:

Tissue:

Mammal legs (intact joint capsule)

Solutions:

Sterile PBS w/o  $\mbox{Ca}^{++}$  and  $\mbox{Mg}^{++}$ 

Antibiotic/antimycotic (100X)

PSN (100X)



Kanamycin (100X)

Gentamicin (1000X)

Fungizone (1000X)

70% ETOH or IPA solution

#### DMEM

Collagenase type II, Gibco, 17101-015

#### Supplies:

#4 blade handles

#22 blades

Sterilized razor blades

Autoclaved Forceps

Autoclaved Nalgene squirt bottle

Hemostats

# For cell isolation:

Autoclaved cell strainer (with 74 µm mesh)

 $60\ ml$  syringes and  $0.2\ \mu m$  filters

T-75 flasks

Shaker plate

\*\* Contact Kim x51547 in IBB to confirm room for our bag in the meat freezer\*\*

# **Solution Preparation**

Harvest

- 1. PBS+
  - a. Use undiluted 1x bottle of Dulbecco's PBS w/o  $Ca^{++}$  and  $Mg^{++}$ .



- b. Add 10ml/L antibiotic/antimycotic (stock @ 100x).
- 2. DMEM+ a.k.a. Arnold (overnight antiobiotics for tissue samples)
  - a. Use sterile high glucose DMEM. Add 10 ml/L PSN (stock @ 100x), 10 ml/L kanamycin (stock @ 100x), 1 ml/L gentamicin (stock @ 1000x), and 1 ml/L fungizone (stock @ 1000x).

#### Cell isolation

- 1. 0.4% collagenase digest solution
  - a. In DMEM+ (see above) add 0.4% collagenase. Make enough for 10 ml digest solution per 1 gram tissue + extra volume.
  - b. CALCULATION: [total mass of tissue]\*10 ml/g tissue \* 0.004 = \_\_\_\_ g of collagenase.
  - c. Sterile filter collagenase solution using 60 ml syringe and 0.2  $\mu$ m filter.

#### **Procedure for Tissue Isolation**

- 1) Prepare sterile PBS and place in a sterile 500ml Nalgene squirt bottle.
- 2) Cut the majority of the meat away from around the joint capsule, but do not penetrate the capsule. Make sure to cut enough meat away from around the capsule in order for someone to hold the bones comfortably.
- 3) Transfer the leg to the hood.
- 4) Wash the capsule surface with EtOH and then PBS+.
- 5) Articulate the joint to identify the femoral heads and patella.
- 6) Using a #4 scalpel with a #22 blade, take light slices on the outside of the joint capsule along the imaginary line immediately below the femoral condyles. These slices will expose the lateral and medial collateral ligaments.



- Penetrate the joint capsule on the anterior side exposing the patellar ligament (inferior to the patella) and the fat pads. Squirt the inside of the capsule with PBS+.
- Use a second scalpel and blade to transect the collateral ligaments and the patellar ligament.
- Transect the anterior and posterior cruciate ligaments (between the two femoral condyles) without damaging the menisci or the cartilage on the condyles.
- 10) At this point the femur and the tibia should be easy to separate, if not, go back and see which ligaments have not been fully transected.
- 11) Cut through the meat on the posterior side of the leg to fully separate the tibia and femur.
- 12) Detach the tendons and ligaments from bone.
- 13) Transfer ligaments and tendon to Petri dishes and squirt with PBS+.
- 14) Dispose of leg as appropriate. Double bag the remains in biohazard bags and take it toIBB 0230 (basement) <u>before 4 pm</u> and ring the doorbell for assistance.

#### **Procedure for Cell Isolation**

- 1) With a new scalpel blade, attempt to remove any excess soft tissue and fibrous tissue.
- 2) Dice the tissue into roughly  $1-3 \text{ mm}^3$  cubes.
- Record weight of empty Petri dish. Aspirate excess PBS and weigh tissue in Petri dish to get tissue only weight.
- 4) Place tissue into T-75 flasks so that there is about 2-3 grams of tissue per flask.
- 5) Add 10 ml of 0.4% collagenase digest solution per 1 gram of tissue to each T-flask. T-flasks should be placed on their sides and secured firmly to the shaker plate. The whole assembly is then placed into the incubator for 12-48 hours. Agitate at 3-5 Hz.



#### After digestion (~48 hr)

- 6) When fully digested, filter cell suspension by pipetting suspension into a 60 ml syringe attached to a metal cell strainer with 74µm mesh. Strain cell solution into a 50 ml conical. Dilute the suspension 1:1 with warm PBS to decrease the viscosity of the collagenase solution.
- 7) Centrifuge all samples at 1300 rpm for 10 minutes. For ligament and tendon the solution above the pellet may appear murky due to cells that did not completely separate out during the centrifugation. If this occurs leave at least 5 ml of solution above the pellet when aspirating. Then, dilute and repeat centrifugation to get the maximum number of cells to pellet down.
- Aspirate off collagenase solution and resuspend cells in 10 ml sterile PBS w/o IONS or DMEM. Recombine and repeat if necessary.
- 9) Count cells to determine how much media to later add for the desired cell density.
- 10) Spin the cells at 1300 rpm for 7 min. Aspirate off the PBS.
- 11) Add an appropriate amount of media with 10% FBS and 10% DMSO. Use the auto to pipette up and down to mix cells thoroughly throughout the solution.
- 12) Add 1 ml of cell solution to each tube you will be freezing down.
- Place the cell vials in the appropriate cooling container and place them in a -80°C freezer.
- 14) After enough time has passed put the now cold vials into the liquid nitrogen for longterm storage.



#### **B.11 Cell Encapsulation**

Purpose: Gels will be created with cells incorporated in them.

# Materials:

Media (prepared day before)

#### Sterile PBS

15 and 50 mL conicals as needed

3 and 10 mL syringes, with one 18G needle for each.

1 mL syringe

Sterile filters

12-well plates

Sterile containers for your gels (Teflon molds, or other molds)

Spatulas, sterilized

# The day before your encapsulation

- 1) Combine the polymers desired in a scintillation vial. You can freeze this vial down until the day of encapsulation. Note: You will lose about 200-300  $\mu$ L due to sterile filtration, so adjust accordingly by making a gel that requires 300  $\mu$ L more than the PBS you would have added to your polymers.
- 2) Sterilize Teflon molds and spatulas as required. Always a good idea to have extra.
- 3) Make sure you have enough sterile filters and other items for your encapsulation.

# Day of encapsulation

#### Polymer side

4) Warm up your polymer powders, trypsin, and your media.



- If doing a thermal encapsulation, create 0.3 M stock solutions of APS and TEMED in scintillation vials as you would normally.
  - a. APS: 684.6 mg in 10 mL PBS
  - b. TEMED: 0.348 g (~350 µL) in 10 mL PBS
- 6) Add sterile PBS to each vial according to your calculations and vortex.
- 7) Let the solutions sit at 37 °C for at least 30 min in order to remove bubbles.
- Filter sterilize APS and TEMED solutions (if you're using them) into 15 mL tubes.
  Make sure to cover the tubes with aluminum foil to prevent photoinitiation.
  - a. Open the packaging of a syringe filter, but leave the filter in its container.
  - b. Attach a needle to a 10 mL syringe, and draw up the contents of a scintillation vial.
  - c. Invert the syringe and remove the needle.
  - d. Grip the sides of the filter container, and screw the syringe onto the filter.
  - e. Filter into an open 15 mL tube.
- 9) Prepare the UV lamp and warm it up if you're doing a photo encapsulation.
- 10) Place sterile, autoclaved molds into petri dishes (one mold per dish).
- 11) Filter sterilize your prepolymer mixtures. Tip: For a gel less than 1 mL, use a 3 mL syringe, then after pulling up the gel, pull up additional air. The air helps with sterile filtering.
- 12) Transfer the correct amount of gel to a separate 15 mL tube.
- 13) Place your sterile spatulas.
- 14) Prepare and label your 12-well plates.



**Important note:** At minimum, steps 4-11 **must** be completed before the actual encapsulation itself.

#### Cell side

- 15) Rinse your flasks twice with PBS, and trypsinize your flasks as normal.
- 16) While you count your cells, spin down the cells in the centrifuge. **Caution:** Before you spin down the cells in this step, make sure steps 4-11 are done.
- 17) Aspirate the media off the cells, then resuspend the pellet in sterile media, typically at a concentration of  $50 \times 10^6$  cells/mL.
  - Amount of media to resuspend in  $\mu L = \#$  of cells/desired concentration x 1000
  - e.g., with 10,000,000 cells total, then 10e6 / 50e6 x 1000 = resuspend my cell pellet in 200 μL of media.

#### Crosslinking/encapsulation

- 18) Add cell solution according to your calculations and quickly mix with the pipette (do **not** vortex). If the cell solution has a concentration of  $50 \times 10^6$  cells/mL and you used the standard gel calculator, this will give a final solution of  $10 \times 10^6$  cells/mL.
- 19) If doing a thermal encapsulation, add APS according to your calculations and mix with pipette. Then, add TEMED according to your calculations and mix with pipette.
- 20) If doing a photo encapsulation, add sterile D2959 to your solution and mix with pipette.
- Deposit the appropriate amount of the resulting solution in your molds. The standard is 30 μL of gel in a 1.0-mm Teflon mold well.
- 22) Cover petri dish and place it in the incubator for 10 min for thermal polymerization;leave under lamp for photo polymerization.



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- 23) Aliquot 2 mL media into each well of a 12-well plate, depending on how many gels you have.
- 24) Remove from the heat and carefully move the gels into the 12-well plate using a sterile spatula. Put the 12-well plate into the incubator.
- 25) Repeat steps 18-24 as necessary with other polymer solutions.
- 26) Recommended: Change media one hour after encapsulation.

#### **B.12** Dimethylmethylene Blue (DMMB) Assay

#### Materials:

Chondroitin sulfate A or B (B = Dermatan sulphate) Sigma Aldrich C-4384

9-dimethylmethylene blue chloride (DMMB) Sigma Aldrich 341088

Na<sub>2</sub>HPO<sub>4</sub> Fisher Sci BP332-500

Di-sodium-EDTA Fisher Sci S311-100

NaCl Fisher Sci S640-500

Glycine Fisher Sci G46-500

Cysteine-HCl Sigma Aldrich C7477

0.1 M stock HCl solution

Regular flat-bottom 96-well plates (clear)

# **Reagent Preparation:**

<u>PBE buffer, pH 6.5:</u> Dissolve 7.1 g Na<sub>2</sub>HPO<sub>4</sub> and 1.86 g Na<sub>2</sub>EDTA in 495 mL of ddH<sub>2</sub>O. Calibrate pH meter, then read and adjust pH to 6.5 +/- 0.1 with concentrated HCl (~3 mL 12 M stock HCl and ~8 mL 1 M HCl), using a glass pipette. Adjust volume to 500 mL and filter sterilize (buffer can be stored for 1-2 months under refrigeration).



- <u>GAG standard:</u> Dissolve 17.5 mg of cysteine-HCl in 10 mL of PBE buffer. Make a stock GAG soln of 20 mg/mL in PBE/cysteine, aliquot, and store at -20C. For the GAG standard, a working solution of 200 ug/mL is needed (100 X dilution of stock – 10 uL stock in 990 uL PBE)
- <u>DMMB solution</u>: Add 95 mL of 0.1 M HCl solution (792 uL of 12 M HCl in 94.2 mL  $ddH_2O$ ) to 905 mL  $ddH_2O$  and dissolve 3.04 g glycine and 2.37 g NaCl. Verify the pH = 3 (correct if necessary) and, while stirring, add 16 mg DMMB to the buffer. When stored in the dark at RT, the solution is stable for 3 months may have to filter before use.

NaCl solution, 2.3 M: Dissolve 13.4 g of NaCl in 100 mL ddH2O.

# GAG Assay:

1) Pipette standards into wells of a clear bottom 96-well assay plate.

Standard Curve: Should be linear from 0-2 µg/well.

| μg GAG/well | μL GAG stock (200 μg/mL) | μL PBE |
|-------------|--------------------------|--------|
| 2           | 10 µL                    | 15 μL  |
| 1.5         | 7.5                      | 17.5   |
| 1.25        | 6.25                     | 18.75  |
| 1           | 5                        | 20     |
| 0.75        | 3.75                     | 21.25  |
| 0.5         | 2.5                      | 22.5   |
| 0.25        | 1.25                     | 23.75  |
| 0           | 0                        | 25     |

2) Add 25  $\mu$ L of sample to empty wells.

Samples: Often need 2-3X dilution to get in linear range.

- 3) Add 5  $\mu$ L of 2.3 M NaCl to each well.
- 4) Add 200 µL of DMMB solution to each well (can use multichannel pipette).
- 5) Read absorbance at 520 nm.



#### **B.13 PicoGreen Assay**

**Purpose:** The Pico Green assay quantifies the amount of DNA present which is an indirect measure of the number of cells.

#### Materials:

12 well plates

Sterile PBS

Spatulas, sterilized

Tweezers, sterilized

96 well plates

#### For Gels

- 1) Remove media.
- 2) Add 2ml iPBS.
- 3) Remove iPBS.
- 4) Add 3-4ml iPBS
- 5) Place in incubator for at least 30 min. Leave in incubator until gels are no longer the color of the media and have become clear.
- 6) Transfer the gels between pieces of weigh paper to remove excess water and weigh the gels. Then put the gels in 1.7 ml eppendorf tubes.
- Homogenize the gels with a pellet grinder in the tubes. Be careful not to lose any of the gel.
- 8) When the gel is well ground add  $ddH_2O$  to the tube depending on the size of the gel (500 µl for 30 µl gels from 1.0 mm Teflon molds). Some of the  $ddH_2O$  can be used to



wash the pellet grinder tip into the tube and to wash the side of the tube. This will maximize the amount of gel that is available for the bioassay.

- 9) Let sit at room temperature for 30 min.
- 10) Store at -80 °C if needed.

#### For Plated Cells in a well plate

- 1) Aspirate liquid from the wells of the plate
- 2) Add 2 ml PBS to rinse.
- 3) Repeat steps 1-2.
- 4) Add 1 ml distilled, deionized water (ddH<sub>2</sub>O).
- 5) Let sit at room temperature for 30 min.
- 6) Store at -20 °C if needed.

#### Lysing the cells

- 1) Freeze the cells at -80 °C for a minimum of 1 hour.
- 2) Thaw at room temperature for 30 minutes.
- 3) Sonicate for 30 min. to lyse the cells.

*Note:* The sonicator will heat up after each usage. Ice should be added to the water in order to cool the sonicator down. If this is not done the DNA in the sample could break down.

*Safety note:* The sonicator is meant to lyse the cells. Do not put fingers into the water bath while the sonicator is on or cells in your fingers will be lysed.

- 4) Repeat steps 1-3 two additional times.
- 5) If the samples are being stored before the assay is done freeze them at  $-80^{\circ}$ C.



# Making the plates

# Standards

- Make sure to thaw out samples and PicoGreen reagents for about 20 minutes before doing the assay.
- Make up buffer solution and PicoGreen according to the PicoGreen calculator. Vortex both solutions to make sure they are mixed thoroughly.

*Note*: When creating the buffer solution make sure to take the volume in the DNA standards into account. Also, cover the PicoGreen solution in aluminum foil to protect it from light.

3) Make up DNA standards according to the table below.

| Conc. (ug/ml) | DNA (ul)        | <b>Buffer Solution</b> |
|---------------|-----------------|------------------------|
| 5             | 25              | 475                    |
| 3             | 15              | 485                    |
| 1             | 5               | 495                    |
| 0.5           | 50 of 5 ug/ml   | 450                    |
| 0.3           | 50 of 3 ug/ml   | 450                    |
| 0.1           | 50 of 1 ug/ml   | 450                    |
| 0.05          | 50 of 0.5 ug/ml | 450                    |
| 0.03          | 50 of 0.3 ug/ml | 450                    |
| 0.01          | 50 of 0.1 ug/ml | 450                    |
| 0             | 0               | 1000                   |

4) Add 43  $\mu$ l of your sample solutions into the appropriate wells of a 96 well plate.

Make sure to change tips between samples.

- 5) Add 107  $\mu$ l of the buffer solution into the wells.
- 6) Add 150 of the PicoGreen Solution to the wells.
- 7) Read in a plate reader: ex 485, em 528.



# **B.14** Confocal Microscopy

#### Notes

• You must be trained to use the confocal microscopes. Email

steve.woodard@ibb.gatech.edu for the next training session.

- There are three scopes that you can use:
  - LSM 510 UV, in room 1328.
  - LSM 510 NLO, in room 1326. This is "the multi-photon" confocal microscope.
  - LSM 510 VIS, in room 1326. This is currently the preferred scope.
- Make sure to reserve a spot on the confocal microscope up to one week in advance at http://my.ilabsolutions.com/account/login/. You will need to make an account there if you haven't already.
- These instructions are for the LSM 510 VIS.

#### **Materials Needed**

- Bring a box containing gloves and your samples.
- When working with gels, bring additional equipment:
  - Attofluor cell chamber and round glass coverslips
  - PBS squirt bottle (Complete PBS with  $Ca^{2+}$ ,  $Mg^{2+}$ )
  - Spatulas and tweezers as necessary, plus Kimwipes for any spills.

#### **Setting Up**

- Turn on the two switches to the left of the microscope turn these on if needed, plus the computer.
- 2) At the login screen, login with your Georgia Tech AD username and password.
- 3) Click on ZEN 2008, in the middle of the screen, and click Start System.



- Adjust the workspace zoom using the slider on the upper right side of the screen. Use an existing preset or make your own, clicking the "open folder" or "save" icons to save or apply changes.
- Turn the first two lasers on (488 must go to "standby" first, 543 is simply on/off). No need for 633.
- Below the laser controls, open the Temenoff Lab setting. This is specially configured for Live/Dead imaging.
- 7) At the top of the second column, note the current objective being used.
  - a) On the LSM 510 VIS, there are 10x, 20x and 40x air objectives. 10x is the most commonly used. 20x and 40x objectives give better magnification, but have less light exposure (and thus produce less bright images as a result).
  - b) Two objectives, 40x and 63x objectives are oil-based.
    - i) Select the objective using the software and place one drop of oil on the objective.
    - ii) Mount your sample and raise the objective until it just touches and oil spreads out.
    - iii) When you are done, wipe off the oil with lens paper, **not kimwipes**.
- Note other settings for averaging images (usually 1, with more for 20x/40x objectives), the resolution of your images (preferred is 1024x1024), and make sure scan speed is set to "Max" (9).
- Note your physical controls. To the right of the microscope is an X-Y controller. Below your monitor is a touchscreen with a coarse and fine control for controlling Z direction.

# **Preparing Samples**

- 1) Prepare a sample for viewing.
  - a) If you've got a plate, just place the plate on top and you are all set.


- b) To prepare for viewing a gel, place a glass cover slip on the bottom piece of the cell chamber. Screw both pieces together and squirt some PBSi into the chamber to keep the gel hydrated. Then load your sample into the circular grooves on the microscope platform.
- 2) Control your imaging using the Channels area in the  $2^{nd}$  column.
  - a) Pinhole size refers to how much light is let through. Higher pinholes let through more light but will increase the amount of light captured in the Z-direction (that is, above and below your image). A preferred pinhole size is 1 AU (Airy unit, showing the lowest Z distance that will be captured). You can increase this if necessary.
  - b) Detector Gain amplifies light received by the detector. Higher gain, brighter images, but also more noise. You should settle for a max of around 800 on both channels; sometimes you may even go below 700.
  - c) Digital Offset eliminates light below a certain threshold. Use this to eliminate stray bright pixels.
- 3) Click Fast on the upper left hand of the screen to begin continuously scanning.
- 4) How to make images clear
  - a) Something important is that imaging is **highly subjective** and depends on what you are looking for. Microscopy experts don't like saturating images, but our lab usually tends towards that. Use your experience and be consistent in how you image, and things'll turn out fine.
  - b) Set the offset and gain on each channel to near maximum. Lots of noise should fill the screen.



- c) To the left of the image, choose Split to separate the red and green signals.
- d) Scan to where you can find cells. This can be done by moving up and down (using the focus knob on the side) or by xy translation. Look at where the laser hits the gel to determine where you are. For Z location, check the Z Stack section (lower right, 2<sup>nd</sup> column) for your height.
- e) Below the image, there are colors indicating what is being shown. Click the colors below the "Merged" button to change the display to Range Indicator mode. Red mean oversaturation, blue means undersaturation, and black/gray is somewhere in between.
- f) For each channel, you can use the Detector Gain and Offset controls to change how things look. You want just a tiny amount of red, and any spots NOT of interest should be totally blue. Eliminate as much noise as possible without losing areas of interest.
- g) Exit Range Indicator mode and see how your images look. You may want to retweak as necessary.
- h) Your calibration should be good for the rest of your wells/gels.

### **Taking Pictures**

- 1) Taking single pictures
  - a) Use Fast to assist as you scan for areas of interest. When you find one, click Single in the upper left, then save the current image.
  - b) Images should be saved in D:\[your name] in .lsm format.
- 2) Taking Z stacks



- a) The Z Stack section is below Channels. Check the box to the right of the title to activate it (without it, the Start button in the upper left will be grayed out).
- b) Use Fast to select your bottom image (if imaging whole gel, start with a completely black image) and click Set First.
- c) Use Fast to select your top image and click Set Last, then click Stop.
- d) Make sure Keep Interval is selected. This will keep the z slice size constant, even if you later decide to change the # of slices or the upper and lower limits in your stack. Typically the slice size in our lab is set to 10 µm.
- e) Click Start. A scan will automatically begin. To view the images as they are being generated, click Gallery to the left of the image.
- f) Don't forget to save your Z stack (also in .lsm format).

### **Saving Data**

Note: Do not expect your data to be backed up. It usually is, but there's no guarantee. We make it easy for you to save your data to the Temenoff lab drive.

- 1) Click Start > Run:  $\zeta cb.edu\temenoff-lab$
- 2) When it asks for a username and password, give it your BME credentials.
- The Temenoff lab server should now be mounted. Copy your images over as appropriate.

## **Cleaning Up**

- 1) Lasers
  - a) Shut down the Argon laser (or put it on Standby if you know someone else will use it).
  - b) Shut down the HeNe laser (or leave it on).



- If you need to shut down the system, give the lasers a few minutes to finish cooling (you should hear a noise stop in 2-3 minutes). Shut down the computer first, then flip both switches (the ones to the left of the scope) off.
- 3) If working with gels, rinse out the cell chamber and throw away the glass cover slip in biohazard glass trash. Wash the cell chamber with soap and distilled water and leave the individual pieces on the rack in the cell culture room to dry. Do not use ethanol; it dries out the rubber O-ring.

#### **Working with Pictures**

- The free LSM 510 Image Browser (downloadable from Zeiss) can be used to view images. Pretty much the only reason you'd really need to use it, other than viewing pictures and exporting them to various formats, is the Projection feature.
  - a) Select your Z stack and open it.
  - b) Select Projection. Rotate around the Y axis, and choose the most projection images (64/panorama) to generate to ensure a smooth 3D projection.
  - c) The created projection can be slowed and manipulated, as well as stored along with the database.
  - d) This can also be exported to a movie file.
- You can also use the image browser to create your own databases with custom images. Use the Copy and Paste functions in the toolbar.
- You can also use the Zeiss ZEN LE software (essentially, ZEN without the microscope controls).
- 4) You can also use ImageJ, especially if you're not on Windows.



#### **Confocal Microscope Settings**

You can use the FITC/CY3 as an initial setting.

Below the imaging buttons, check Z-Stack.

Turn on the Argon and HeNe543 lasers. (Argon must go to standby first, and after a

while you can set it to On. This is very important - don't forget to turn it off of Standby.)

Imaging Setup

| Mode         | Switch track every |
|--------------|--------------------|
| Channel Mode | Frame              |

First channel name: Fluorescein 488-517 (FITC)

Second channel name: Rhodamine 543-600

Select the fluorescein channel.

Light path: To the left of laser, select HFT 488/543.

Above HFT 488/543, select NFT 545.

To the right of NFT 545, select BP 505-530, and check Ch2. Select green for Ch2.

Above NFT 545, select Plate.

Next on the path, select LP 560. Check Ch1 and select Red.

Select the rhodamine channel and perform the exact same steps.

Acquisition mode:

| Objective | Scan mode | Frame size  | Line step | Speed     | Averaging |
|-----------|-----------|-------------|-----------|-----------|-----------|
| 10x       | Frame     | Click X*Y   | 1         | Click Max | 1         |
|           |           | 1024 x 1024 |           | 9         |           |



Averaging isn't important unless you want to fiddle around with how it averages images.

Useful for poor visibility.

Channels: Fluorescein

| Laser power | Pinhole | Gain              | Digital offset  | Digital gain   |
|-------------|---------|-------------------|-----------------|----------------|
| 3.0         | 1 AU    | 1100-1200 set;    | > -0.10 set;    | 1.00 operating |
|             |         | 700-750 operating | -0.30 operating |                |

#### Channels: Rhodamine

| Laser power | Pinhole | Gain           | Digital offset  | Digital gain   |
|-------------|---------|----------------|-----------------|----------------|
| 35.0        | 1 AU    | 1100-1200 set; | > -0.10 set;    | 1.00 operating |
|             |         | 800 operating  | -0.30 operating |                |

#### Z Stack: Rhodamine

| Interval   | Keep     |
|------------|----------|
| 10 microns | Interval |

Use "Set First" and "Set Last" to specify the beginning and end of a stack.

Once all of this has been set up, click the Save icon near the Configuration name near the

upper left corner, and save it under your own custom preferences.

### **B.15** Reverse Transcription Polymerase Chain Reaction (RT-PCR)

#### **RNA Extraction**

### Extract RNA using Qiagen RNeasy Mini Kit

### (Consult the RNeasy kit manual for more specific protocols depending on your

#### cell/tissue source)

<u>Materials</u>

Molecular BioProducts RNase Away Spray (VWR 17810-491; 475 ml)

Aerosol Filter Pipette Tips for Rainin LTS, 20 µl (VWR 83009-688; pack of 960)



Aerosol Filter Pipette Tips for Rainin LTS, 200 µl (VWR 82003-196; pack of 960) Aerosol Filter Pipette Tips for Rainin LTS, 1000 µl (VWR 82003-198; pack of 576)

QIAshredder (Qiagen 79654 or 79656; 50 or 250 runs)

RNeasy Mini Kit (Qiagen 74104 or 74106; 50 or 250 runs)

Optional Alternative: RNeasy Protect Mini Kit (Qiagen 74124 or 74126; 50 or 250 runs)

EMD 2-Mercaptoethanol (VWR EM-6010; 100 ml)

RNAse-free DNase Set (Qiagen 79254; 50 runs)

### <u>Protocol</u>

For plated cells

- I) Trypsinize cells and centrifuge (10 min, 1000 rpm).
- II) Aspirate supernatant, resuspend pellet in media, and centrifuge (10 min, 1000 rpm).
- III) Aspirate supernatant, rinse pellet with PBS, and centrifuge again (10 min, 1000 rpm), aspirate PBS.

## For gels

- I) Soak gels in PBS for ~ 1h to remove media.
- II) Transfer the gel to an RNase free, DNase free microcentrifuge tube.
- III) Break the gel into small pieces using a pellet grinder.
- Lyse cell pellet or cells in gel in 350 μl Buffer RLT with β-mercaptoethanol (add 10 μl BME per 1 ml Buffer RLT).
- 2) Put solution in purple QIAshredder column and centrifuge (2 min, 14000 rpm).
- 3) Discard filter and add **350 µl 70% ethanol** to eluted substance.



- 4) Transfer suspension to **pink RNeasy column** and centrifuge (15 sec, 14000 rpm).
- 5) Discard eluted substance and put filter back on. (Do NOT mix Buffer RLT or RW1 with bleach when discarding. Contact with acids forms highly reactive guanidine salts and liberates very toxic gas)
- 6) Add **350 µl Buffer RW1** to column and centrifuge (**15 sec, 14000 rpm**).
- 7) Discard eluted substance and put filter back on.
- Add 10 μl DNase I (lyophilized DNase I is resuspended in 550 μl RNase-free water) to 70 μl Buffer RDD for each sample.
- Add 80 µl of DNase/RDD solution directly onto RNeasy membrane and incubate at room temperature for 15 min.
- 10) Add 350 µl Buffer RW1 to column and centrifuge (15 sec, 14000 rpm).
- 11) Discard eluted substance and put filter back on.
- 12) Add 500 µl Buffer RPE (add 44 ml of 96-100% ethanol to starting 11 ml of Buffer

RPE concentrate before first time use) to column and centrifuge (15 sec, 14000 rpm).

- 13) Discard eluted substance and put filter back on.
- 14) Add **500 µl Buffer RPE** to column and centrifuge (**2 min, 14000 rpm**).
- 15) Discard eluted substance and transfer column to new 2 ml collection tube.
- 16) Centrifuge (1 min, 14000 rpm).
- 17) Discard 2 ml collection tube, and transfer column to new 1.5 ml collection tube with cap.
- 18) Add **30 µl RNase-free water** and centrifuge (1 min, 14000 rpm).
- 19) The water elutes the RNA into the collection tube cap the tube and store RNA at  $80^{\circ}$ C.



### Quantify and check purity of RNA

### <u>Materials</u>

MP Biomedicals RNase, DNase-free water (VWR IC821739; 500 ml)

### <u>Protocol</u>

- Take absorbance readings of 2 µl of undiluted RNA at 260 nm and 280 nm light in NanoDrop.
- 2) Quantity of RNA can be calculated using the following equations:
  - Corrected  $A_{260}$  = average sample  $A_{260}$  average blank  $A_{260}$
  - Concentation in µg/ml = (corrected A<sub>260</sub>) \* (44 µg/ml) \* (dilution factor) (using above protocol, dilution factor = 50)
  - Total mass in µg = (µg/ml concentration value) \* (µl volume) / 1000
     (volume of RNA extraction sample; using above extraction protocol, volume = 30-50 µl)
  - Volume in μl needed for 1 μg RNA = (1 μg RNA) \* 1000 / (μg/ml concentration value)

(1 ng to 5 µg RNA can be used for Reverse Transcription)

- Volume of water in µl needed = 10 µl total volume RNA volume determined above
- 3) Purity of RNA can be calculated using the following equation:
  - Purity =  $A_{260} / A_{280}$
  - This value should be between 1.5 and 1.9.



### **Reverse Transcription**

### **Reverse Transcription with SuperScript III Reverse Transcriptase**

### <u>Materials</u>

Oligo(dT)<sub>15</sub> Primer (Promega C1101; 20 µg)

PCR Nucleotide Mix, 10 mM (Promega C1141; 200 µl)

SuperScript III Reverse Transcriptase (Invitrogen 18080-093 or 18080-044; 2000 or

10000 units)

Invitrogen RNaseOUT RNase Inhibitor, 40 units/ml (Invitrogen 10777-019; 5000

units)

### <u>Protocol</u>

- 1) Add the following components to a nuclease-free PCR tube:
  - 1 ng to 5 µg total RNA: 10 µl
  - Oligo(dT)<sub>15</sub> (500 µg/ml): 1 µl
  - dNTP Mix (10 mM each): 1 µl
  - RNase, DNase-free water: to 12 µl final volume

 Heat mixture to 65°C for 5 min and chill on ice for at least 1 min. Collect the contents of the tube by brief centrifugation and add:

- 5X First-Strand Buffer (250 mM Tris-HCl, 375 mM KCl, 15 mM MgCl<sub>2</sub>): 4 µl
- 0.1 M DTT: 1 µl
- RNaseOUT (40 units/µl): 1 µl
- 3) Mix contents of the tube gently. Incubate at 42°C for 2 min.



- Add 1 μl (200 units) of SuperScript III RT and mix by pipetting gently up and down.
- 5) Incubate at 50°C for 30-60 min.
- 6) Inactivate the reaction by heating at  $70^{\circ}$ C for 15 min.
- 7) Store cDNA at  $-20^{\circ}$ C.
- 8) Amplification of PCR targets (>1 kb) may require the removal of RNA complementary to the cDNA. To remove RNA complementary to the cDNA, add 1 μl (2 units) of *E. coli* RNase H and incubate at 37°C for 20 min.

### **Primer Preparation**

### Reconstitute primers (100 µM)

#### <u>Materials</u>

Custom Primers/Oligonucleotides, desalted (Invitrogen; 25 nmol)

### <u>Protocol</u>

- 1) Find the total nmoles from the information sheet that came with the primer.
- The volume of DNase-free water needed to create a 100 µM stock can be calculated using the following equation:
  - Volume of DNase-free water in  $\mu$ l = (nmoles of primer) \*1000 / (100  $\mu$ M)

### <u>Make 10 µM aliquots</u>

### <u>Protocol</u>

- 1) Briefly spin the primers and add the needed amount of water for a 10-fold dilution.
- Make a 10-fold dilution (1 part primer: 9 parts DNase-free water) to obtain a 10 μM aliquot.



### **Polymerase Chain Reaction**

## Run PCR

### <u>Materials</u>

SYBR Green Master Mix (Applied Biosystems 4309155; 5 ml)

Applied Biosystems StepOnePlus Real-Time PCR System and Materials

- MicroAmp Fast 96-Well Reaction Plates, 0.1 ml (Applied Biosystems 4346906; 20 plates)
- MicroAmp Optical Adhesive Films (Applied Biosystems 4360954; 25 films)
   OR
- MicroAmp Fast 8-Tube Strips, 0.1 ml (Applied Biosystems 4358293; 125 strips)
- MicroAmp Optical 8-Cap Strips (Applied Biosystems 4323032; 300 strips)
- MicroAmp 96-Well Trays for Veriflex Blocks (Applied Biosystems 4379983; 10 trays)

## <u>Protocol</u>

- 1) Thaw the following components on ice:
  - SYBR Green mix
  - 10 µM forward primer of interest
  - 10 µM reverse primer of interest
  - Sample cDNA
- 2) To make Master Mix, count the number of wells needed for each primer, add 2, and multiply this number by the following to obtain enough Master Mix for all wells of the primer:



- 12.5 µl SYBR Green mix
- 10.5 µl DNase-free water
- 0.5 µl 10 µM forward primer
- 0.5 µl 10 µM reverse primer
- 3) Load PCR wells:
  - Load **24 µl of Master Mix** into each well for that primer (target sequence of interest).
  - Add **1** µl of sample cDNA into the well with Master Mix and mix by pipetting up and down (change pipette tips between each well).
- 4) After all wells are loaded, cover with optical tape or caps, and put entire plate on ice until ready to run PCR.
- 5) Load plate into StepOnePlus system:
  - Load plate into machine (A1 in upper-left, H12 in lower-right).
  - Open "StepOne Software v2.0" program.
  - Click "Advanced Setup" button.
  - Under "Experiment Properties," enter an Experiment Name, select "StepOnePlus Instrument (96 Wells)" for the instrument, select "Quantitation – Comparative  $C_T$  $(\Delta\Delta C_T)$ " for the experiment, select "SYBR Green Reagents" for the reagent (melt curve is optional), and select "Standard" for the ramp speed.
  - Under "Plate Setup," add your targets and samples under "Define Targets and Samples," changing the Reporter to "SYBR" and the Quencher to "None." Under "Assign Targets and Samples," assign the appropriate targets and samples to each



well, selecting your control sample as the reference sample, GAPDH as the endogenous control, and "ROX" as the "dye to use as a passive reference."

- Under "Run Method," change the reaction volume to 25 μl and set the method to the following:
  - Hold at 95°C for 10 min (100% ramp).
  - Cycle 40 times at 95°C for 15 sec and 60°C for 1 min, collecting data at 60°C.
  - Optional melt curve: 95°C for 15 sec, 60°C for 1 min, and +0.3°C ramp,
     ending at 95°C for 15 sec and collecting data during the +0.3°C ramp.
- Save the file.

(Templates can be saved to reduce setup time.)

- Select "Start Run."
- After the run, click "Analyze" (settings can be modified under "Analysis Settings") and resave the file.
- Export results to Excel by clicking "Export..." and export "Results" as "One File" with ".xls" file type.

## **B.16 Quantitative PCR Analysis**

## Adapted from:

Ruijter JM, Ramakers C, Hoogaars WMH, Karlen Y, Bakker O, van den Hoff MJB, and Moorman AFM. Amplification efficiency: linking baseline and bias in the analysis of quantitative PCR data. Nucleic Acids Res. 2009 Apr;37(6):e45.



#### Introduction

Let's review the basics of PCR amplification and detection. We start with a few basic equations:

$$N_c = N_0 E^c \qquad E = \frac{N_{c+1}}{N_c} \qquad N_0 = \frac{N_t}{E^{C_t}}$$

where N = concentration of amplicon, E is the efficiency of the PCR reaction (where 1 means no amplification, and 2 is 100% amplification), and C is a cycle threshold of interest. If you want to compare a gene's amplification to its target gene,

$$\frac{N_{0,\text{collagen I}}}{N_{0,\text{GAPDH}}} = \frac{\frac{N_{t,\text{collagen I}}}{E_{\text{collagen I}}^{C_{t,\text{collagen I}}}}}{\frac{N_{t,\text{GAPDH}}}{E_{\text{GAPDH}}^{C_{t,\text{GAPDH}}}}} \approx \frac{E_{\text{GAPDH}}^{C_{t,\text{GAPDH}}}}{E_{\text{collagen I}}^{C_{t,\text{collagen I}}}}$$

assuming the concentrations of the two genes are the same (given, say, a fluorescence threshold). If you then make the next assumption that the efficiencies are the same (say, 100%), then

$$E_{\text{common}}^{C_{t,\text{GAPDH}}-C_{t,\text{collagen I}}} = E_{\text{common}}^{-(C_{t,\text{collagen I}}-C_{t,\text{GAPDH}})} \approx 2^{-\Delta C_{t}}$$

To compare to day 1, we write

$$\frac{N_{0,\text{collagen I, day 7}}}{N_{0,\text{collagen I, day 1}}} = \frac{E_{\text{common}}^{-(C_{t,\text{collagen I, day 7}}-C_{t,\text{GAPDH, day 7}})}}{E_{\text{common}}^{-(C_{t,\text{collagen I, day 1}}-C_{t,\text{GAPDH, day 1}})}} \approx \frac{2^{-\Delta C_{t,\text{day 7}}}}{2^{-\Delta C_{t,\text{day 1}}}} = 2^{-(\Delta C_{t,\text{day 7}}-\Delta C_{t,\text{day 1}})}$$
$$= 2^{-\Delta \Delta C_{t}}$$

These calculations are contingent on a few assumptions:

- Baseline fluorescence is something that can be eliminated.
- Efficiency is common (and perhaps, also, that it's 100%) among different genes, primer sets, etc.



• Cycle to cycle efficiency does not vary throughout amplification.

#### **Typical PCR analysis errors**

If there *is* baseline fluorescence in your sample, you will introduce error into your estimation of *efficiency*:

$$N_c^* = \text{baseline} + N_0 E^c$$
  $E^* = \frac{\text{baseline} + N_{c+1}}{\text{baseline} + N_c}$ 

and if you also fail to account for differences in amplification efficiency, you also introduce error:

$$\operatorname{error} = \frac{E_{\text{GAPDH}}^{C_{t,\text{GAPDH}}}}{E_{\text{collagen I}}^{C_{t,\text{collagen I}}}} \div \frac{E_{\text{common}}^{C_{t,\text{GAPDH}}}}{E_{\text{common}}^{C_{t,\text{collagen I}}}} = \left(\frac{E_{\text{GAPDH}}}{E_{\text{common}}}\right)^{C_{t,\text{GAPDH}}} \left(\frac{E_{\text{collagen I}}}{E_{\text{common}}}\right)^{-C_{t,\text{collagen I}}}$$
$$= \left(\frac{E_{\text{gene}}}{E_{\text{common}}}\right)^{C_{t,\text{GAPDH}}+C_{t,\text{collagen I}}}$$

which for even a 5% difference in efficiency can be  $1.05^{20+20} = 1.05^{40} = 7.03$  times different.

#### What LinRegPCR does

- Estimates individual well baseline by reconstructing the log-linear portion of the amplification curve:
  - > Discards any samples where low amplification is detected.
  - > Calculates the second derivative maximum (SDM) to find the plateau.
  - > Subtracts baseline, then fits both the upper and bottom portions of log-linear region.
  - > When the lower and upper sections' slopes match, the baseline has been found.
- ✤ Calculates individual PCR efficiencies for every well
  - > Calculates mean efficiency and starting amount  $(N_0)$  for a given amplicon.
  - > Mean efficiencies are adequate for calculating fold regulation values



- Differences are more a result of statistical error than true differences in efficiency.
- What about running multiple plates for the same gene?
- Individual efficiencies can vary significantly

#### How to use LinRegPCR

- 1) Process your data in StepOne first.
  - a) Do a quick quality control check on your samples and remove samples that didn't amplify well, that have a high standard deviation, etc.
  - b) **Disable** the auto baseline feature in StepOne for each target.
  - c) Export the data to Excel check *Sample Setup* and *Amplification Data*.
- 2) Open the spreadsheet and then open LinRegPCR.

Note: LinRegPCR will not work correctly unless Excel is opened beforehand.

- 3) In LinRegPCR:
  - a) Read in your data.
    - i) Check the drop-down boxes to make sure Amplification Data is selected.
    - ii) Select Step-One Plus (ABI) and DNA binding dye (SYBR Green).
    - iii) Select columns A through D, rows 8 through 3848 (that is, a 40-cycle run. A 50-cycle run would go to row 4808.)
    - iv) Make sure ss cDNA is checked, and select No for "Data are baseline-corrected."
  - b) Click the red button labeled **determine baselines**.
  - c) Check each sample to examine efficiencies and tweak if necessary.
  - d) If necessary, use the Amplicon Groups tab to set amplicon groups. The boxes there can help you auto-select amplicon groups quickly for each gene.
  - e) Set the log(fluorescence) value on the left.
    - i) Note: Make sure to keep this consistent between all plates with the same gene.



- ii) Note: In StepOne, the threshold fluorescence is usually less than 1, which is why the log(fluorescence) value in LinRegPCR is negative.
- f) Save this data to Excel.
- 4) Back in Excel:
  - a) Use the sample setup tab to help re-label your samples and rearrange your data to your liking. You may want to pick up some Excel-fu to summarize and put all of your wells together, do any extra quality control, etc.
    - Note: Remember to use geometric means when averaging data. (Taking the geometric mean is the same as taking the arithmetic mean of the cycle thresholds.)
  - b) To calculate fold regulation, use the starting concentration of your gene, or  $N_0$ . The value of  $N_0$  has already been calculated by LinRegPCR using the mean efficiency for the amplicon group.
    - i) If many of your wells did not use the mean efficiency to be calculated, it may be a good idea to calculate a new  $N_0$  based on each individual well's PCR efficiency.
    - ii) You can do this using the equation  $N_0^* = \frac{\text{fluorescence threshold value}}{\text{Eff}^{Cq}}$ .
  - c) Calculate a manual  $\Delta\Delta$  method; that is, divide a sample's gene's N<sub>0</sub> by the sample's GAPDH N<sub>0</sub> value, then divide by the geometric mean of your day 1 sample.
  - d) Perform statistical analysis on the divided-by-GAPDH values.

#### **B.17** Immunostaining

#### **Rules to remember:**

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- Do not let your sample dry out. This will encourage non-specific binding (and staining) from anything floating around in the air, etc. If you need to stop, make sure samples are wet or otherwise sitting in PBS.
- Store samples at -80°C prior to staining to maintain maximum antigenicity.

 Rinsing with PBS can be done using a slide rack. Place slides into the stain rack, then gently lower into the stain holder with PBS. After two minutes, raise and lower the rack gently, then remove the rack completely. Do this twice, discarding PBS between washes. You'll need ~1000-1200 mL PBS for 12 slides.

### Day 1

### For Fixed Paraffin Sections

- 1) Deparaffinize
  - a) Deparaffinize slides in xylene and alcohol (Program 4 in autostainer).
- 2) Antigen retrieval

For heat-induced:

- a) Citrate Buffer (10 mM, pH 6.0)
  - i) Citric acid (anhydrous) 1.92 g
  - ii) Distilled water 1000 ml
  - iii) Mix to dissolve. Adjust pH to 6.0 with 1 N NaOH, and then add 0.5 ml Tween 20 and mix well.
  - iv) Store this solution at room temperature for 3 months or at 4°C for longer storage.
- b) Place slides in 10 mM citrate buffer.
- c) Perform heat-induced antigen retrieval in pressure cooker (high setting) in citrate buffer for 15 min.
- d) Allow slides to cool to room temperature before rinsing.
- e) Rinse twice in PBS.



- a) TE-CaCl<sub>2</sub> Buffer (50 mM Tris Base, 1 mM EDTA, 5 mM CaCl<sub>2</sub>, 0.5% Triton X-
  - 100, pH 8.0)
  - i) Tris Base 6.10 g
  - ii) EDTA 0.37 g
  - iii)  $CaCl_2 0.56 g$
  - iv) Triton X-100 5 ml
  - v) Distilled Water 1000 ml
  - vi) Mix to dissolve. Adjust to pH 8.0 using concentrated HCl (10 N).

vii) Store this buffer at room temperature.

- b) Proteinase K Stock Solution (20X, 400 µg/ml)
  - i) Proteinase K (30 units/mg) 8 mg
  - ii) TE/CaCl<sub>2</sub> Buffer 10 ml
  - iii) Glycerol 10 ml
  - iv) Add proteinase K to TE-CaCl<sub>2</sub> buffer until dissolved. Then add glycerol and mix well.
  - v) Aliquot and store at -20°C for 2-3 years.
- c) Proteinase K Working Solution (1X, 20 µg/ml)
  - i) Proteinase K Stock Solution 1 ml
  - ii) TE-CaCl<sub>2</sub> Buffer -19 ml
  - iii) Mix well.
  - iv) This solution is stable for 6 months at  $4^{\circ}$ C.
- d) Circle samples with PAP pen to create hydrophobic barrier around your samples.



- e) Cover sections with Proteinase K working solution, and incubate for 10-20 min at 37°C in humidified chamber.
- f) Allow sections to cool at room temperature for 10 min.
- g) Rinse twice in PBS.

### For Unfixed Frozen Sections

- 1) Fixation
  - a) Warm up slides at room temperature for 30-60 minutes.
  - b) Lightly fix in ice cold acetone for 10 minutes. Place a stain holder containing acetone into an ice bucket, then lower the slide rack inside. Change acetone between slide racks if you're doing more than 12 slides at a time.
  - c) Let slides air dry for 30 minutes.
- 2) Wash off OCT
  - a) Rinse twice in PBS.
  - b) At this point, samples should be completely free of OCT and other trash. Circle each sample with a PAP pen to isolate future liquids to the sample.

## Deglycosylation with Chondroitinase Digestion for Aggrecan Staining

- 1) Chondroitinase Digestion
  - a) Activate chondroitinase ABC in buffer of 50mM Tris, 60 mM sodium acetate,
    0.02% BSA, pH 8.0. Dilute 1 U/ml chondroitinase 3:4 with 4X activating buffer
    (eg. 0.75 ml chondroitinase, plus 0.25 ml buffer).
  - b) Apply 30 µl of 0.75 U/ml of chondroitinase ABC to each sample, and incubate in a humid chamber for 1.5 hr at 37°C, then PBS rinse.



#### Staining for Immunohistochemistry

- 1) Preparation (while slides dry)
  - a) 1% BSA solution (make 1 g BSA / 100 mL PBS; store at 4°C)
  - b) Serum from species of secondary Ab (goat, horse, or rabbit)
  - c) Make a humidified chamber using a 150 mm petri dish with a wet paper towel in the bottom, plus 4-5 small petri dishes with water inside. Each chamber can hold 4 or 5 slides.
- 2) Peroxidase Blocking (For frozen sections only)
  - a) Use a slide rack/stain holder to immerse slides in 0.3% peroxide/methanol solution (1 mL 30% H<sub>2</sub>O<sub>2</sub> in 100 mL methanol; store in foil < 1 week at 4°C) for 10 minutes at room temperature. You can reuse the peroxide/methanol solution between racks.</li>
- 3) Serum Blocking
  - a) Normal Serum Block Solution (60  $\mu$ L/sample, mix in 10 mL and store at 4°C):
    - i) 1% BSA solution in PBS (stabilizer) 9.7 mL
    - ii) 2% serum (blocking) 200  $\mu$ L
    - iii) 0.1% Triton X-100 (penetration enhancer) 100  $\mu$ L
  - b) Incubate sections for 20 min in blocking solution serum should be same species as secondary antibody. Tap the slide on a kimwipe to remove the solution. Don't do a PBS wash afterwards.

#### Staining for Immunofluorescence

1) Preparation



- a) Make a humidified chamber using a 150 mm petri dish with a wet paper towel in the bottom, plus 4-5 small petri dishes with water inside. Each chamber can hold 4 or 5 slides.
- 2) Image-iT Signal Enhancer Blocking
  - a) Cover section in Image-iT Signal Enhancer (for AlexaFluor488-conjugated IgG)
  - b) Incubate sections for 30 min in blocking solution, then PBS wash.

#### **Primary Antibody**

- 1) Primary Antibody
  - a) Prepare the working dilution of the primary antibody in 1% BSA. If using an antibody for the first time, you'll need to run a series of dilutions to determine the optimal working concentration). With our antibodies, 1:10, 1:100, 1:500, 1:1000 is a good start.
  - b) Use a 10  $\mu$ L pipette to carefully apply 8  $\mu$ L primary antibody to each sample.
  - c) Incubate sections in a humid chamber at 4°C overnight. Parafilm each chamber to prevent evaporation.

#### Day 2

#### For Immunohistochemistry

- 1) Peroxidase Blocking (For paraffin sections only)
  - a) Use a slide rack/stain holder to immerse slides in 0.3% peroxide/methanol solution (1 mL 30% H<sub>2</sub>O<sub>2</sub> in 100 mL methanol; store in foil < 1 week at 4°C) for 10 minutes at room temperature. You can reuse the peroxide/methanol solution between racks.</li>
- 2) Secondary Antibody



- a) Remove slides from the humidified chamber and PBS wash.
- b) Incubate sections in 1:200 goat biotin conjugated secondary antibody in PBS for 30 minutes at room temperature, then PBS wash.
- 3) Signal Amplification
  - a) Solution: Vectastain ABC Elite Standard kit
    - i) 5 mL PBS + 2 drops of solution A + 2 drops of solution B, added into the provided bottle.
    - ii) Allow solution to sit for 30 min before use.
  - b) Drop solution onto slides. Incubate for 30 min, then PBS wash.
- 4) Chromagen/Substrate Stain
  - a) <u>Immediately after rinsing</u>, make the peroxidase substrate solution using 1 mL substrate buffer and 20 µL chromogen. There should be no delay between making the solution and adding it to your slides.
  - b) Allow color to develop for 5-10 minutes (you will need to experiment with this time).
  - c) Immediately rinse with PBS twice. All DAB waste should be added into a separate waste container and brought to the histology room for disposal.

#### For Immunofluorescence

- 1) Secondary Antibody
  - a) Remove slides from the humidified chamber and PBS wash.
  - b) Incubate sections in 1:200 goat anti-mouse Alexa Fluor 488-conjugated secondary antibody in PBS for 30 minutes at room temperature, then PBS wash.
- 2) Nuclear counterstain



a) Incubate sections in 0.1  $\mu$ g/ml DAPI working solution for 5 min at room temperature, then PBS wash.

#### **Coverslip**

- 1) Cover slip
  - a) The slide should be wet from the PBS wash you just did. Place a drop of
     FluoroGel with Tris buffer on each of your samples within a PAP pen circle.
     FluoroGel should mix with the water already on the slide.
  - b) Place additional FluoroGel (can use two drops; may use more if desired) outside of the PAP pen circles.
  - c) Tilt the bottom edge of the cover slip against the bottom edge of the slide. Tilt the slide as necessary so the cover slip catches a consistent edge of liquid along the bottom, then gently lower the top edge so the liquid rides over the PAP pen circles and catches the entire surface of the slide.
  - d) Allow slides to dry overnight. You can view them on a microscope sooner than that (a few hours after coverslipping), but sometimes liquid will leak from a slide as it dries. Clean liquid that has leaked onto the outside with wet kimwipes.
  - e) Seal the edges of the slides with clear nail polish to prevent bubbles from forming under the coverslip.



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